

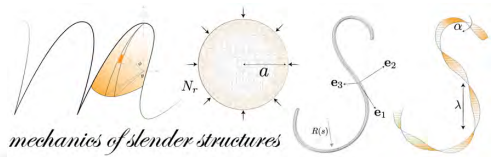
Swelling of Elastic Materials

Fluids Deforming Solids

Douglas P. Holmes

Mechanical Engineering
Boston University





Fluids & Elasticity

Flow through porous medium

- Darcy's Law

Elastic deformation of medium

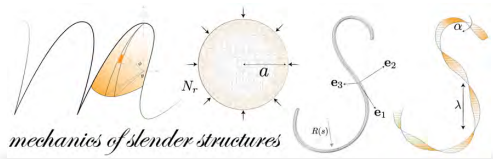
- Biot's Poroelasticity

Swelling

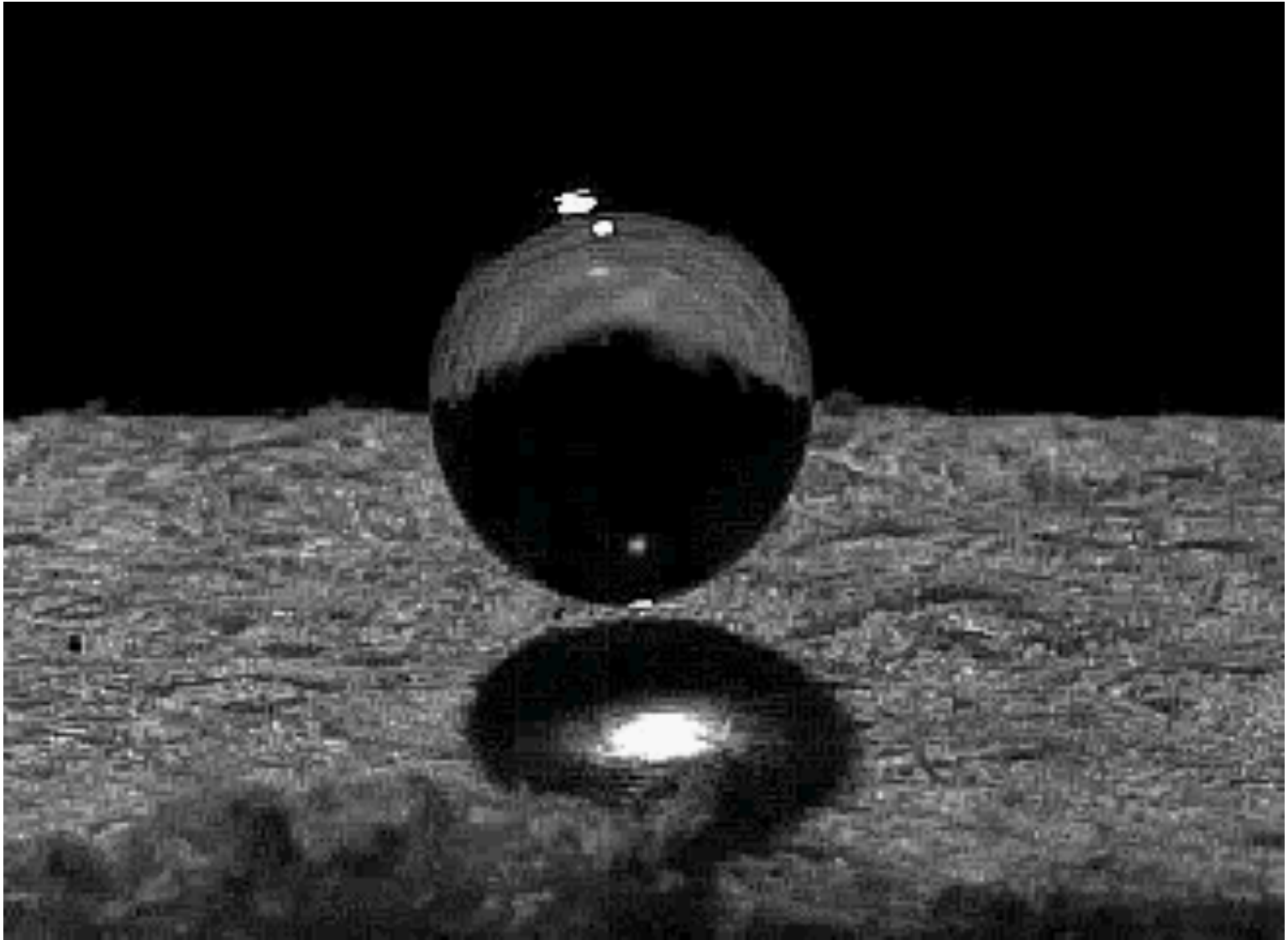
- Polymers
-

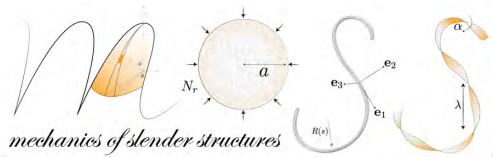
Fluids Deforming Solids

- Surface Tension – Elastocapillarity
- Swelling & Growth
- Maxwell Stresses

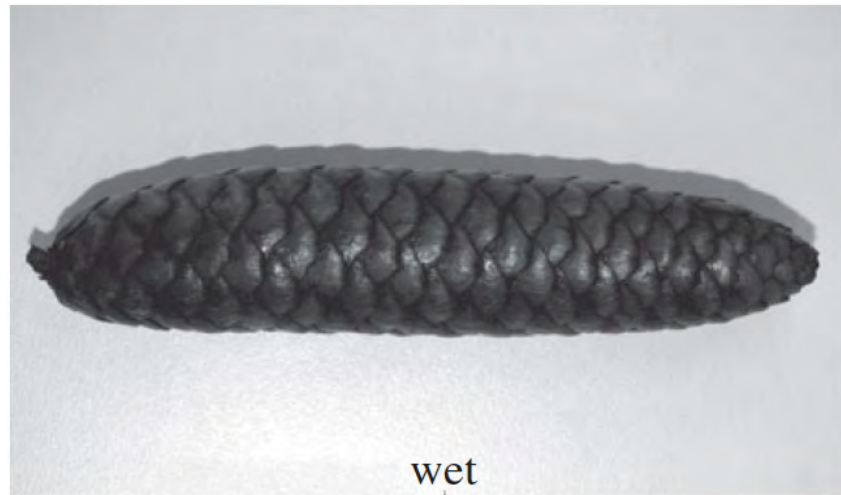


Swelling a Sponge





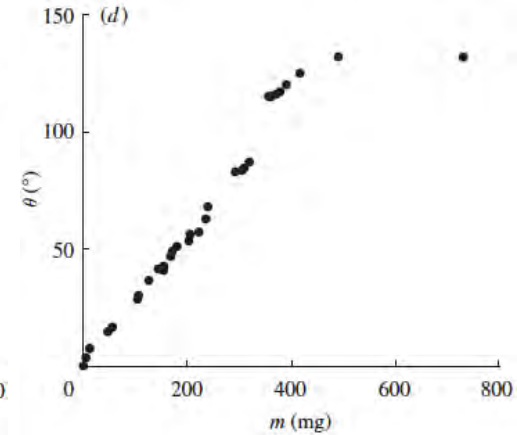
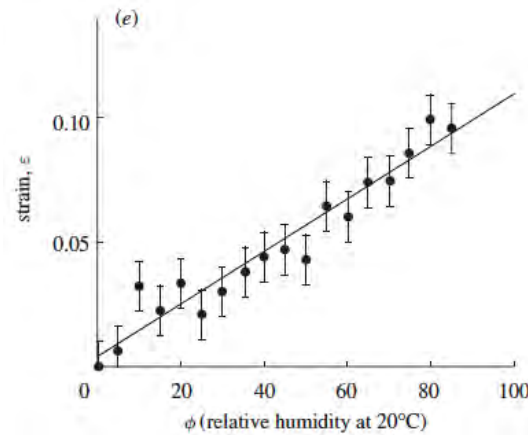
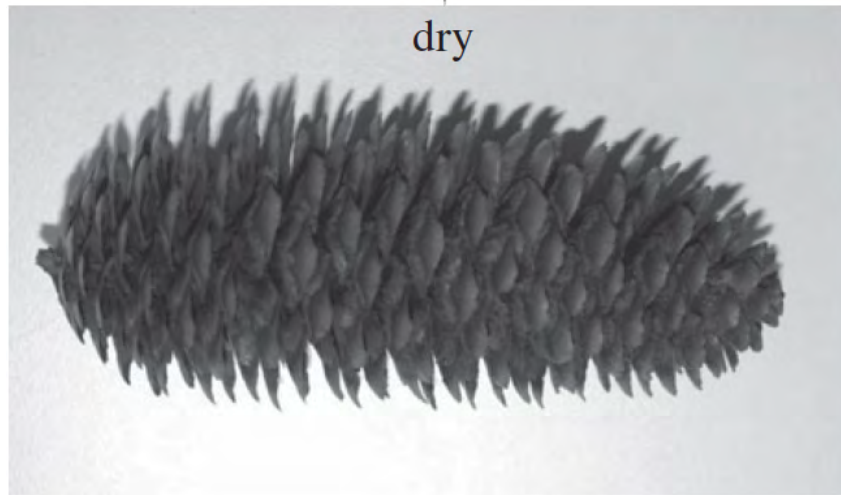
Pine Cones



wet



dry

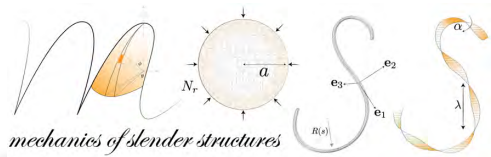


Tree-bound pine cones:

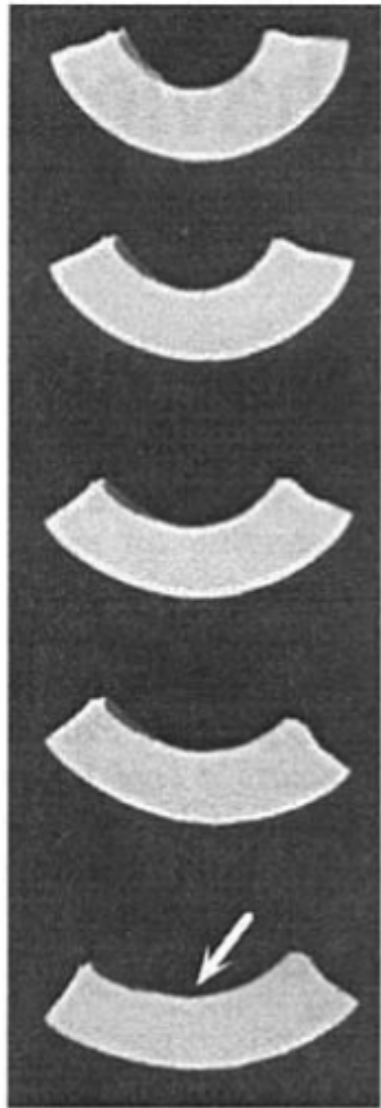
Hydrated & **closed**, protecting seeds

Fallen pine cones:

Dried out & **opened**, releasing seeds



Articular Cartilage



0.015M NaCl

Shape change caused by ion concentration.

0.05M NaCl

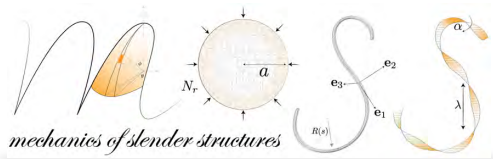
0.15 M NaCl

Residual strain at physiological conditions: **3-15%**

0.5 M NaCl

Tensile prestress in cartilage protective against frequent compresses forces.

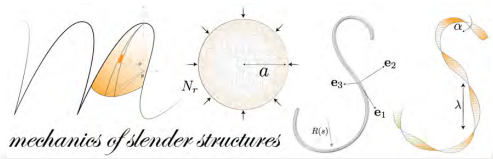
2M NaCl



Lichens in the Rain

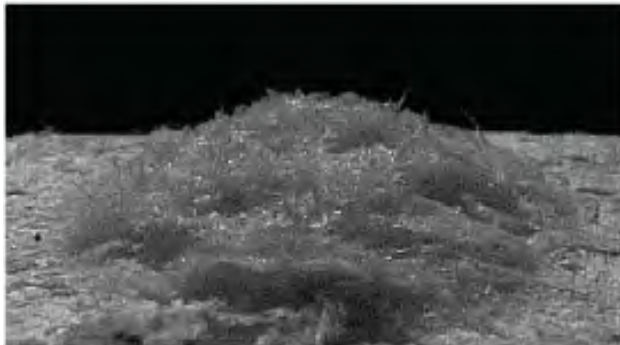
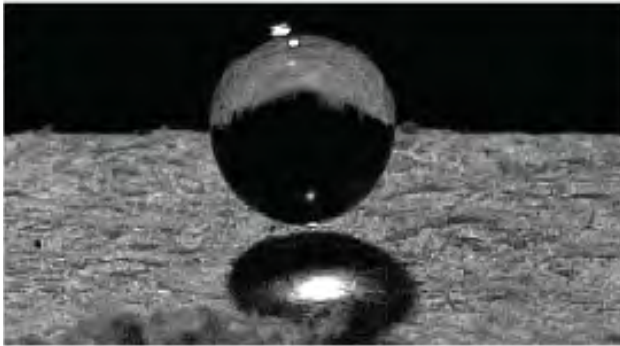


Francis cheefilms, "Lichens time lapse", <https://www.youtube.com/watch?v=FWfPMOKnW2M>, (2015)



Swelling & Growth

Materials Science

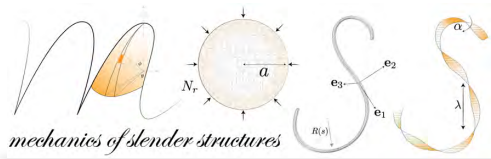


Swelling of a sponge.

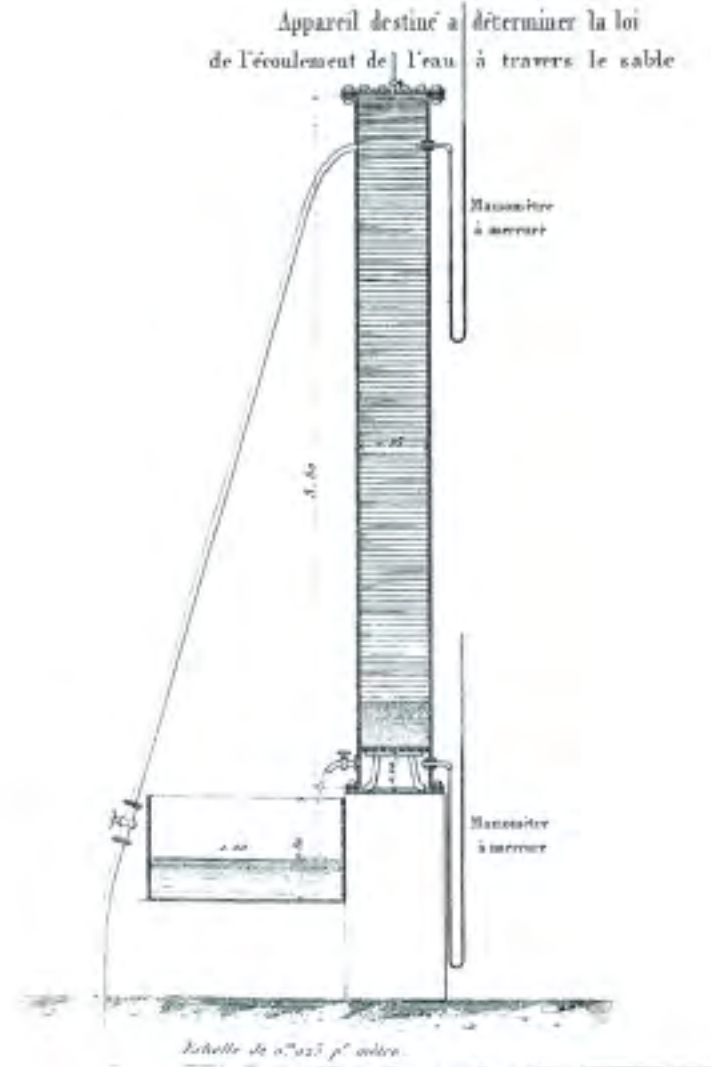
Mechanics

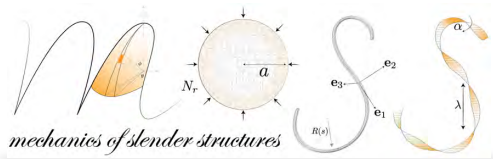


An almond leaf which was attacked by *Taphrina Deformans*.

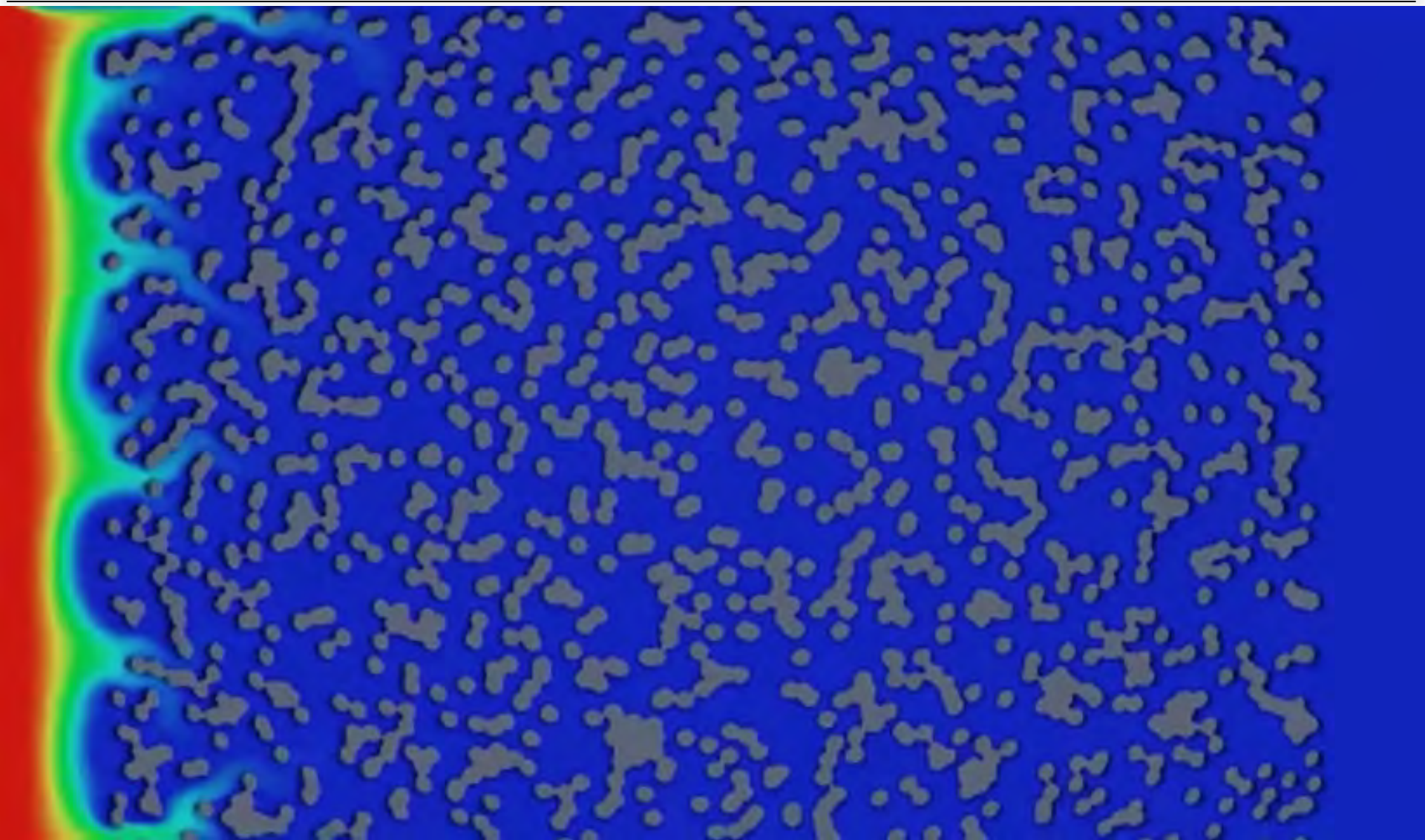


Flow in Porous Media

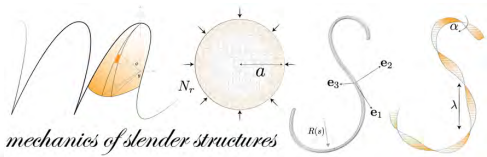




Flow in Porous Media



I. Moumine, <https://www.youtube.com/watch?v=qTvHXRT9qt4>, 2015.



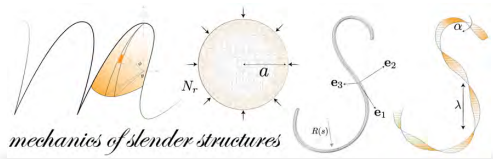
Fluid Dynamics

Navier-Stokes Equations (momentum conservation)

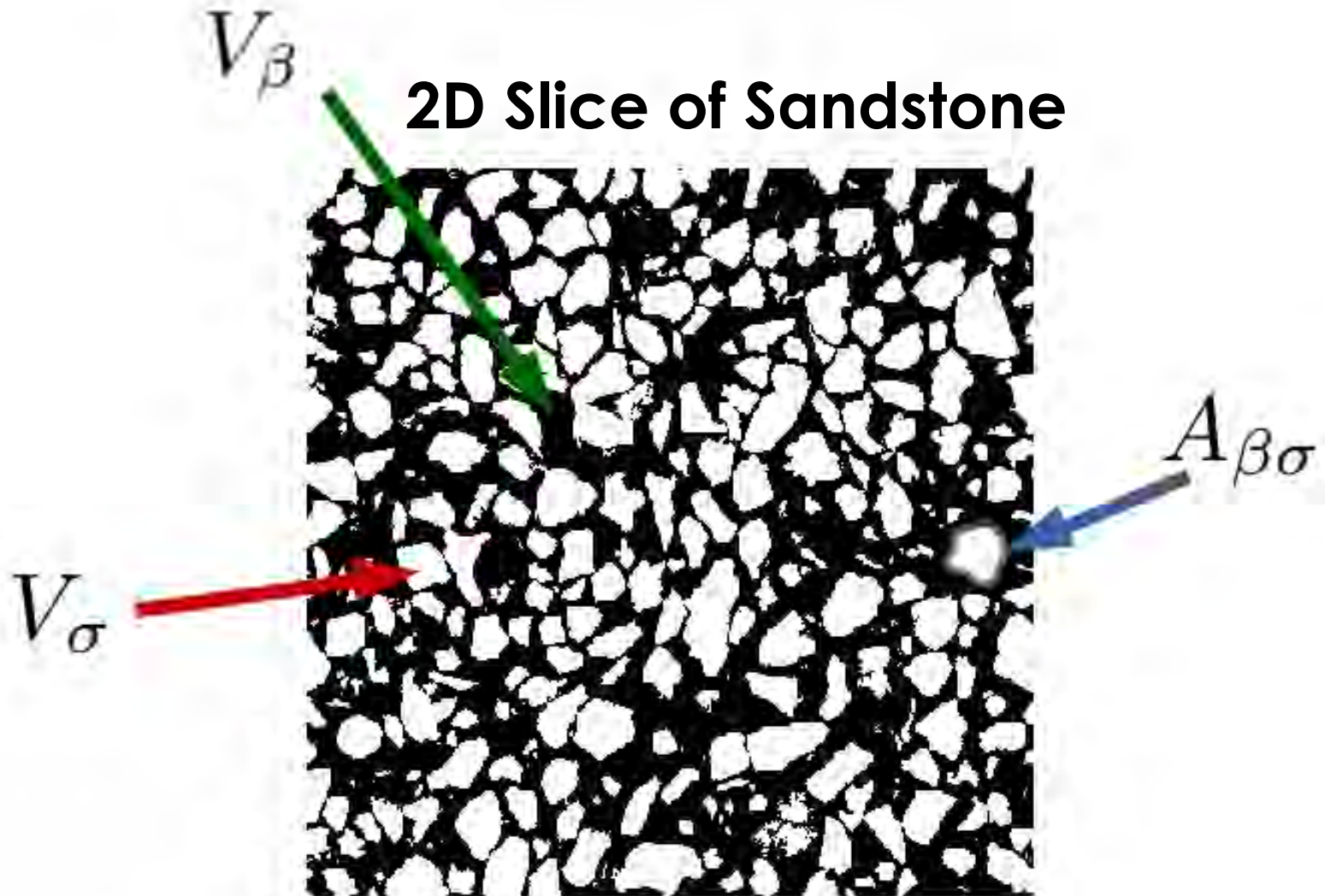
$$\underbrace{\rho \left(\underbrace{\frac{\partial \mathbf{u}}{\partial t}}_{\text{variation}} + \underbrace{\mathbf{u} \cdot \nabla \mathbf{u}}_{\text{convection}} \right)}_{\text{Inertial acceleration}} = \underbrace{\left(\underbrace{-\nabla p}_{\text{pressure}} + \underbrace{\mu \nabla^2 \mathbf{u}}_{\text{diffusion}} + \underbrace{\mathbf{f}}_{\text{body forces}} \right)}_{\text{Forces}}$$

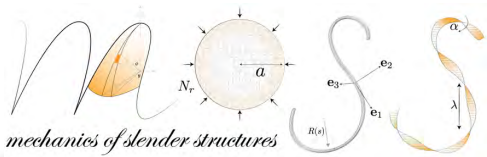
Continuity Equation (mass conservation)

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{u}) = 0$$



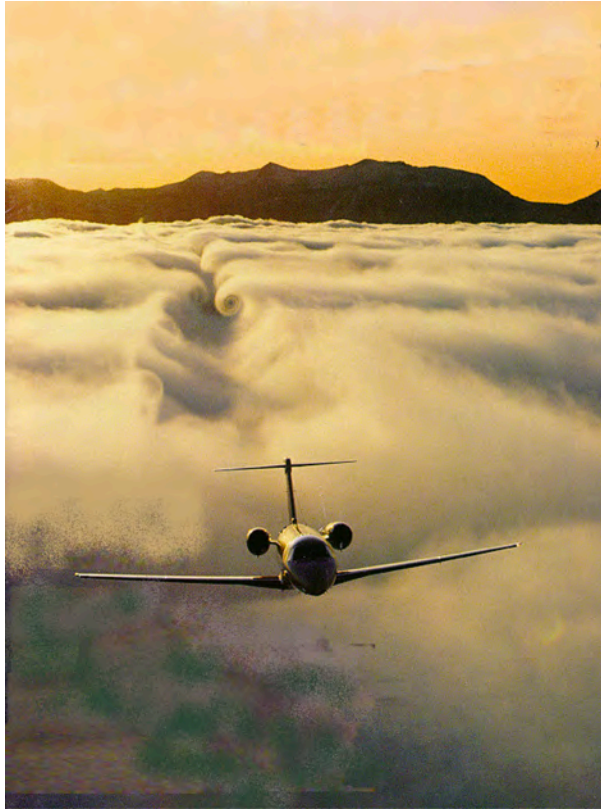
Porous Media





Fluid Behavior

Inertial Forces



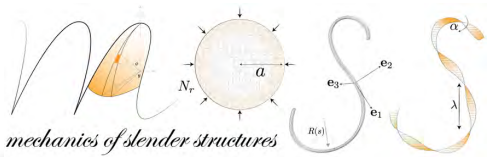
Trailing airplane vortices

Viscous Forces



Coiling honey

Reynolds Number: inertial/viscous



Fluid Behavior

Inertial Forces



Reynolds Number: inertial/viscous

Inertial acceleration

$$\underbrace{\rho \left(\frac{\partial \mathbf{u}}{\partial t} + \mathbf{u} \cdot \nabla \mathbf{u} \right)}_{\text{variation}} = \underbrace{-\nabla p + \underbrace{\mu \nabla^2 \mathbf{u}}_{\text{diffusion}} + \mathbf{f}}_{\text{Forces}}$$

Viscous Forces



Steady inertial forcing due to the convective derivative:

- Time dependence arises from U_0

$$\rho \mathbf{u} \cdot \nabla \mathbf{u}$$

Linear unsteady term sets the inertial time scale to **establish steady flows**:

$$\rho \partial \mathbf{u} / \partial t$$

Time scale estimated by balancing unsteady **inertial force density** with **viscous force density**

$$f_u \sim \rho U_0 / \tau_i$$

$$f_v \sim \mu U_0 / L_0^2$$

$$\tau_i \sim \frac{\rho L_0^2}{\mu}$$

Time required for a vorticity to diffuse a distance L_0 , with a diffusivity $\nu = \mu / \rho$, $\tau_i \sim 10\text{ms}$

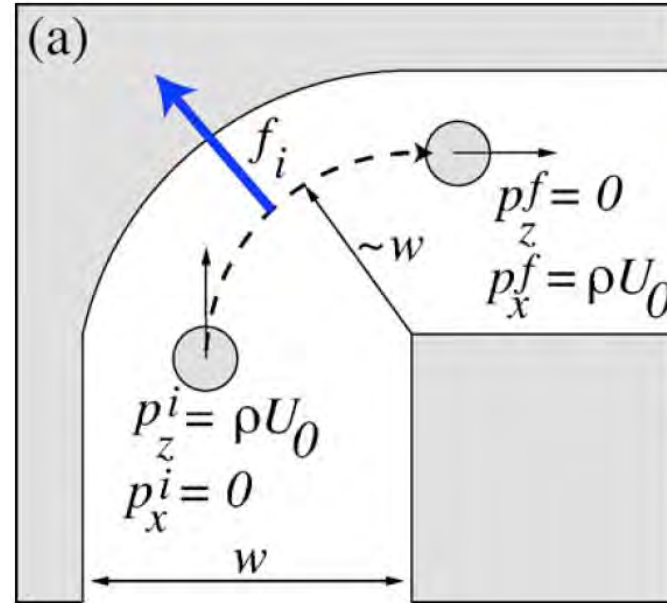
Inertial Forces



Viscous Forces



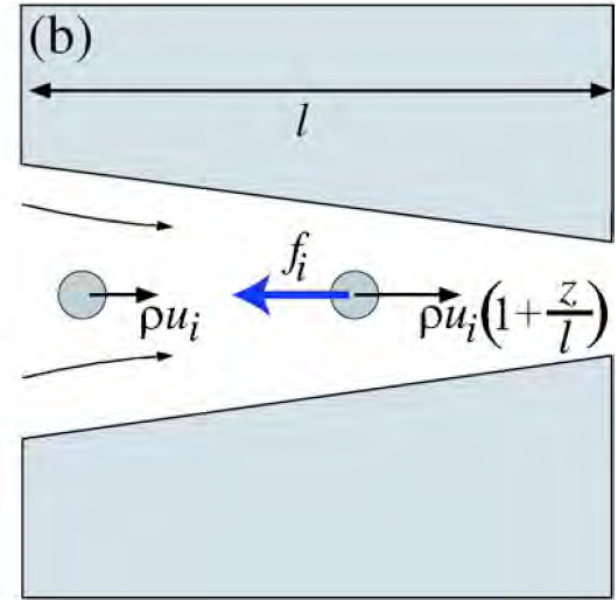
Reynolds Number: inertial/viscous



Fluid element accelerating around curve.

- During a turn time:
 $\tau_0 \sim w/U_0$
- Loss of momentum density:
 ρU_0
- By exerting an inertial centrifugal force density:

$$f_i \sim \rho U_0 / \tau_0 = \rho U_0^2 / w$$



Fluid element in a channel of contracting length.

- By mass conservation, velocity increases as:
 $u \sim U_0(1 + z/l)$
- Gain momentum at a rate:

$$f_i \sim \rho \frac{du}{dt} = \rho U_0 \frac{du}{dz} \sim \frac{\rho U_0^2}{l}$$

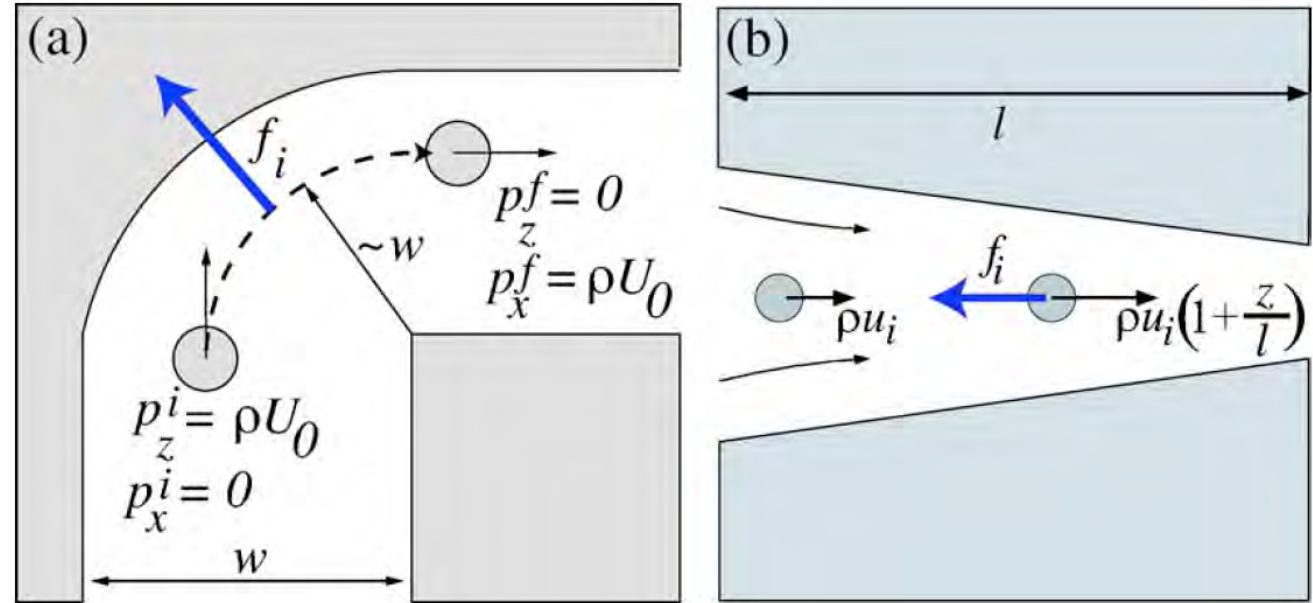
Inertial Forces



Viscous Forces



Reynolds Number: inertial/viscous



Inertial Forces $f_i \sim \rho U_0 / \tau_0 = \rho U_0^2 / w$

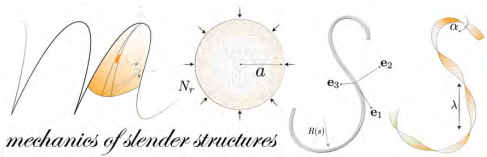
$$f_i \sim \rho \frac{du}{dt} = \rho U_0 \frac{du}{dz} \sim \frac{\rho U_0^2}{l}$$

Viscous Forces

- Viscous force densities result from gradients in viscous stress: $f_v \sim \mu U_0 / L_0^2$

Ratio of these two force densities is the **Reynolds number**:

$$\frac{f_i}{f_v} = \frac{\rho U_0 L_0}{\mu} \equiv \mathcal{R}$$



Fluid Behavior

Inertial Forces



Viscous Forces



Reynolds Number: inertial/viscous $\frac{f_i}{f_v} = \frac{\rho U_0 L_0}{\mu} \equiv \mathcal{R}$

Estimation of Reynolds numbers for common microfluidic devices.

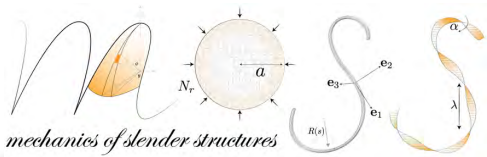
- Typical fluid – water
 - Viscosity: 1.025 cP @ 25°C
 - Density: 1 g/mL
- Typical channel dimensions
 - Radius/height (smaller than width): 1 – 100 μm
- Typical velocities
 - Average velocity: 1 $\mu\text{m/s}$ – 1 cm/s

Typical Reynolds number:

$$\mathcal{R} \sim \mathcal{O}(10^{-6}) - \mathcal{O}(10^1)$$

Low Reynolds number: viscous forces > inertial forces

- Flows are **linear**.
- Nonlinear terms in Navier-Stokes disappear
 - Linear, predictable **Stokes flow**

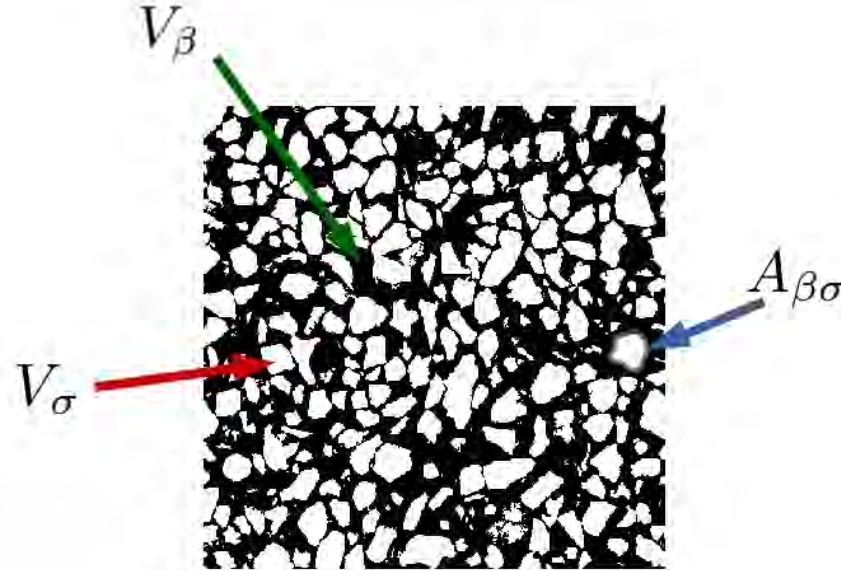


Fluid Behavior

Inertial Forces



Reynolds Number: inertial/viscous $\frac{f_i}{f_v} = \frac{\rho U_0 L_0}{\mu} \equiv \mathcal{R}$



Viscous Forces

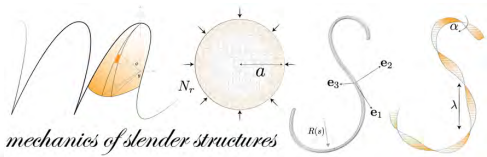


Typical Reynolds number:

$$\mathcal{R} = \frac{\rho U_0 d_{30}}{\mu} \leq 1$$

Low Reynolds number: viscous forces > inertial forces

- Flows are **linear**.
- Nonlinear terms in Navier-Stokes disappear
 - Linear, predictable **Stokes flow**



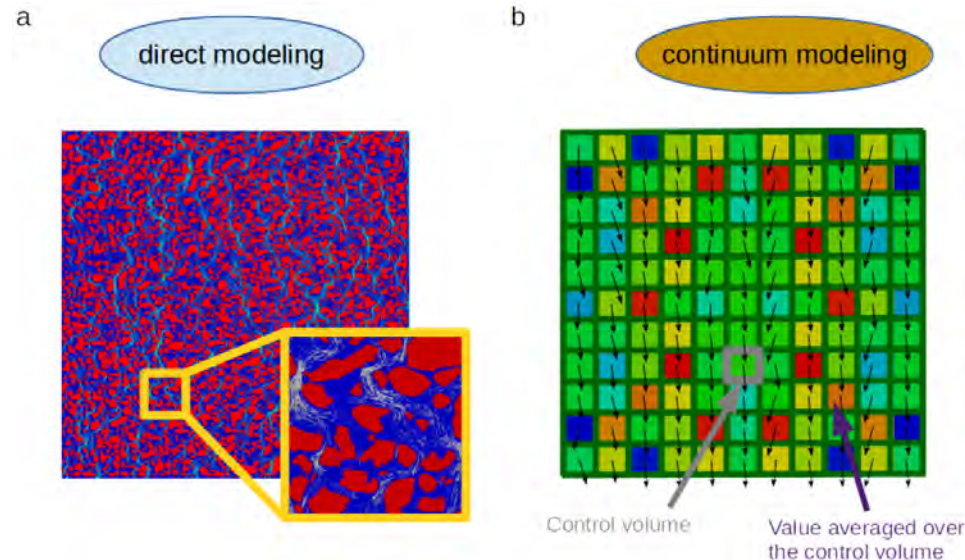
Stokes Equations

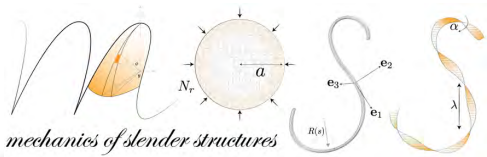
Stokes Equations (momentum & mass conservation)

$$0 = \nabla \cdot \mathbf{u}_\beta$$

$$0 = -\nabla p_\beta + \rho_\beta \mathbf{g} + \mu_\beta \nabla^2 \mathbf{u}_\beta$$

...but, in order to proceed, we need to prescribe BC's on each grain...





Stokes Equations (momentum & mass conservation)

$$0 = \nabla \cdot \mathbf{u}_\beta$$

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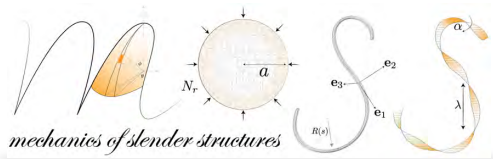
...but, in order to proceed, we need to prescribe BC's on each grain...

Darcy-Brinkman-Stokes Equations

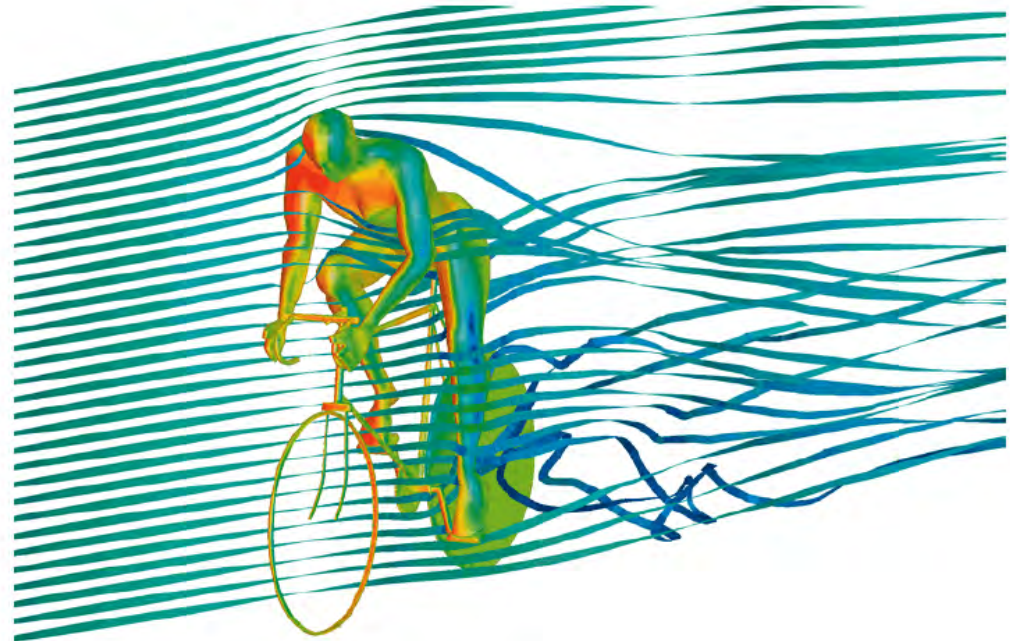
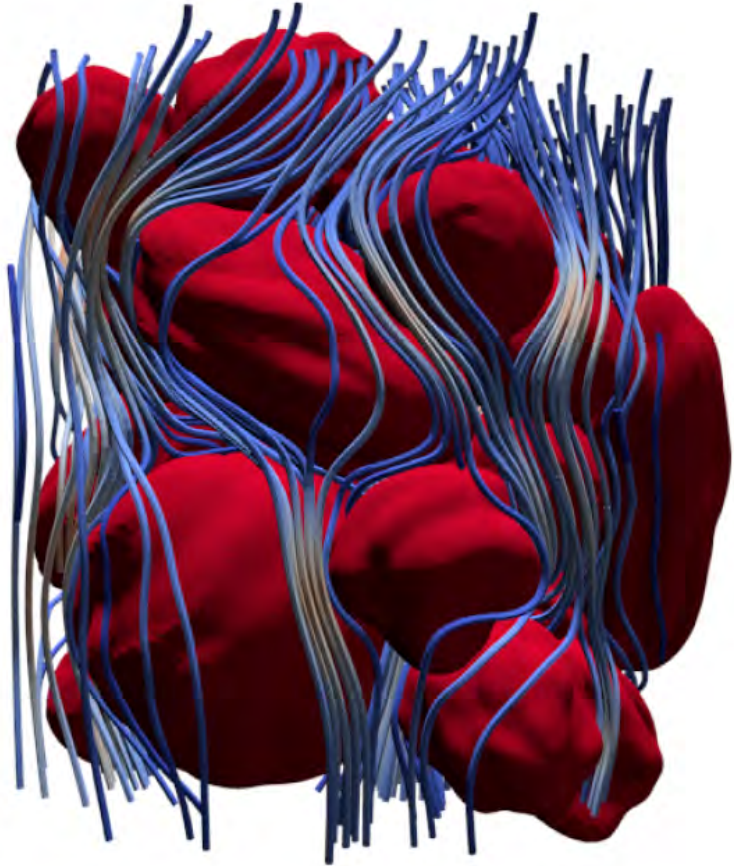
Averaging over pressures and velocities.

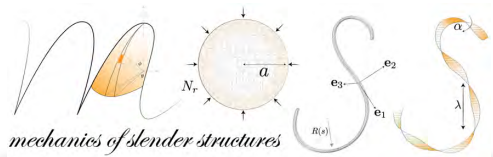
$$0 = -\nabla \langle p_\beta \rangle + \rho_\beta \mathbf{g} + \mu_\beta \nabla^2 \langle \mathbf{u}_\beta \rangle - \underbrace{\mu_\beta k_{ij}^{-1}}_{\text{Viscous friction}} \cdot \langle \mathbf{u}_\beta \rangle$$

Void space: $\langle \phi_\beta \rangle = \frac{1}{V} \int_{V_\beta} dV$



Viscous Drag





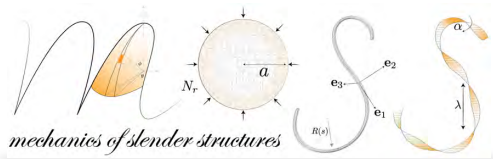
Darcy-Brinkman-Stokes Equations

Averaging over pressures and velocities.

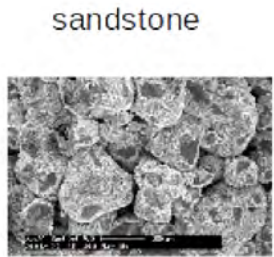
$$0 = -\nabla \langle p_\beta \rangle + \rho_\beta \mathbf{g} + \underbrace{\mu_\beta \nabla^2 \langle \mathbf{u}_\beta \rangle}_{\text{Often negligible}} - \underbrace{\mu_\beta k_{ij}^{-1}}_{\text{Viscous friction}} \cdot \langle \mathbf{u}_\beta \rangle$$

Darcy's law (volumetric flux, isotropic medium)

$$\mathbf{q} = -\frac{k}{\mu} (\nabla \langle p_\beta \rangle - \rho_\beta \mathbf{g})$$



Porous Media



200 μm

Column of adsorbent beads



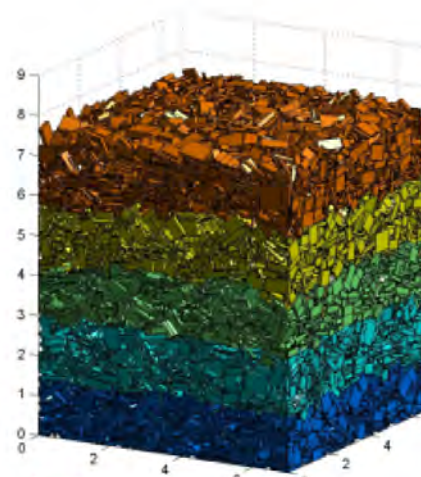
2 mm

Structured packing



2 cm

Eco-Shale



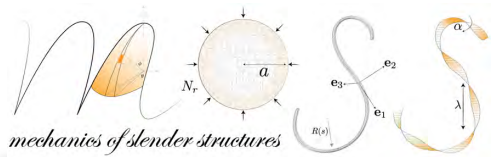
2 m

Canopy



m





Biot Poroelasticity

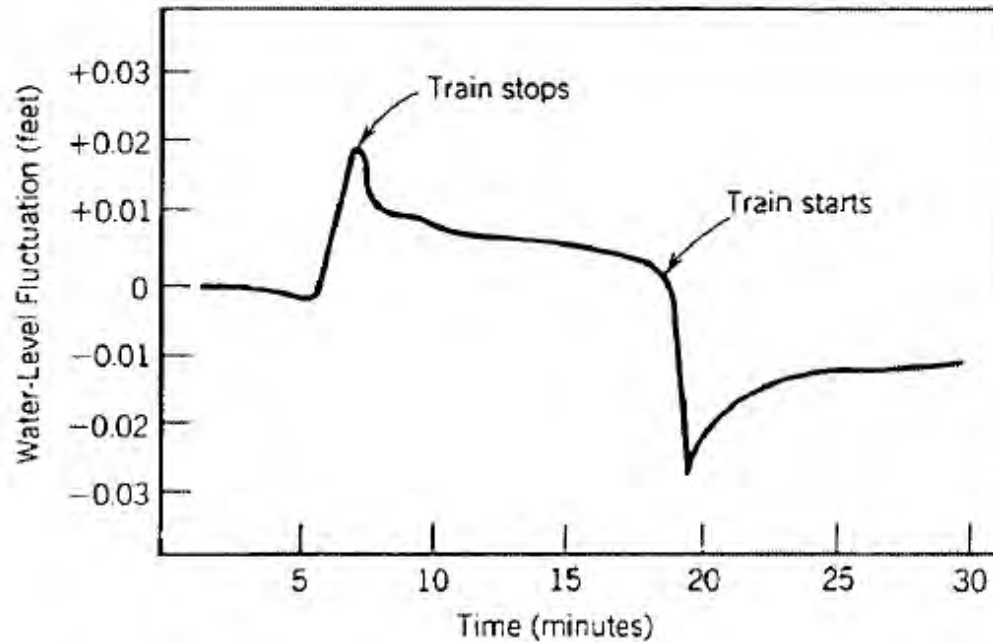
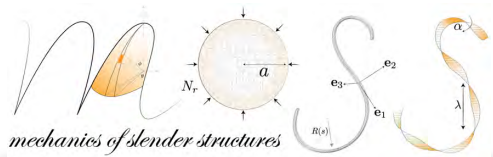


Figure 1.1: Water-level fluctuations due to a passing train. An approaching train compresses the aquifer, which quickly raises the pore pressure in the affected region. Fluid then flows away from the high-pressure region. As the train departs, the aquifer expands, thereby quickly reducing the pore pressure in the affected region. Fluid again flows in response to the pressure differences, but this time it builds up the pore pressure. The approximately equal and opposite behaviors demonstrate that the aquifer is elastic (Domenico and Schwartz, 1998, p. 65; Jacob, 1940).



Biot Poroelasticity

Coupled problem:

Pore pressure has time dependence, as does poroelastic stresses/strains.

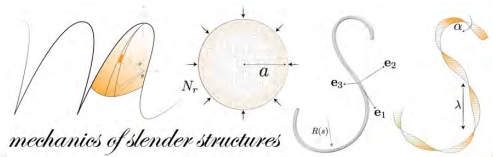
Poroelasticity:

Cannot solve fluid flow problem independent of stress field.

- Stress changes in fluid-saturated porous media typically produce significant changes in pore pressure.

Increment in **total work** associated with **strain increment** and **fluid content**.

$$dW = \sigma_{ij} d\varepsilon_{ij} + p d\zeta$$



Biot Poroelasticity

Time dependence work is related to the **fluid flux** through **Darcy's law**.

$$\alpha \frac{\partial \varepsilon}{\partial t} + S \frac{\partial p}{\partial t} = \frac{k}{\mu} \nabla^2 p$$

Squeeze the soil –
how much water
comes out?

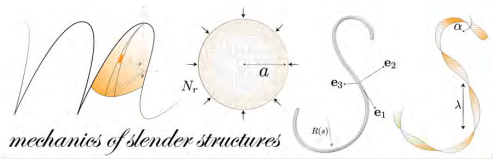
Pressurize the water –
how much water will
go in the soil?

Compression of the medium (e.g. soil) includes compression of pore fluid and particles plus the fluid expelled from an element by flow.

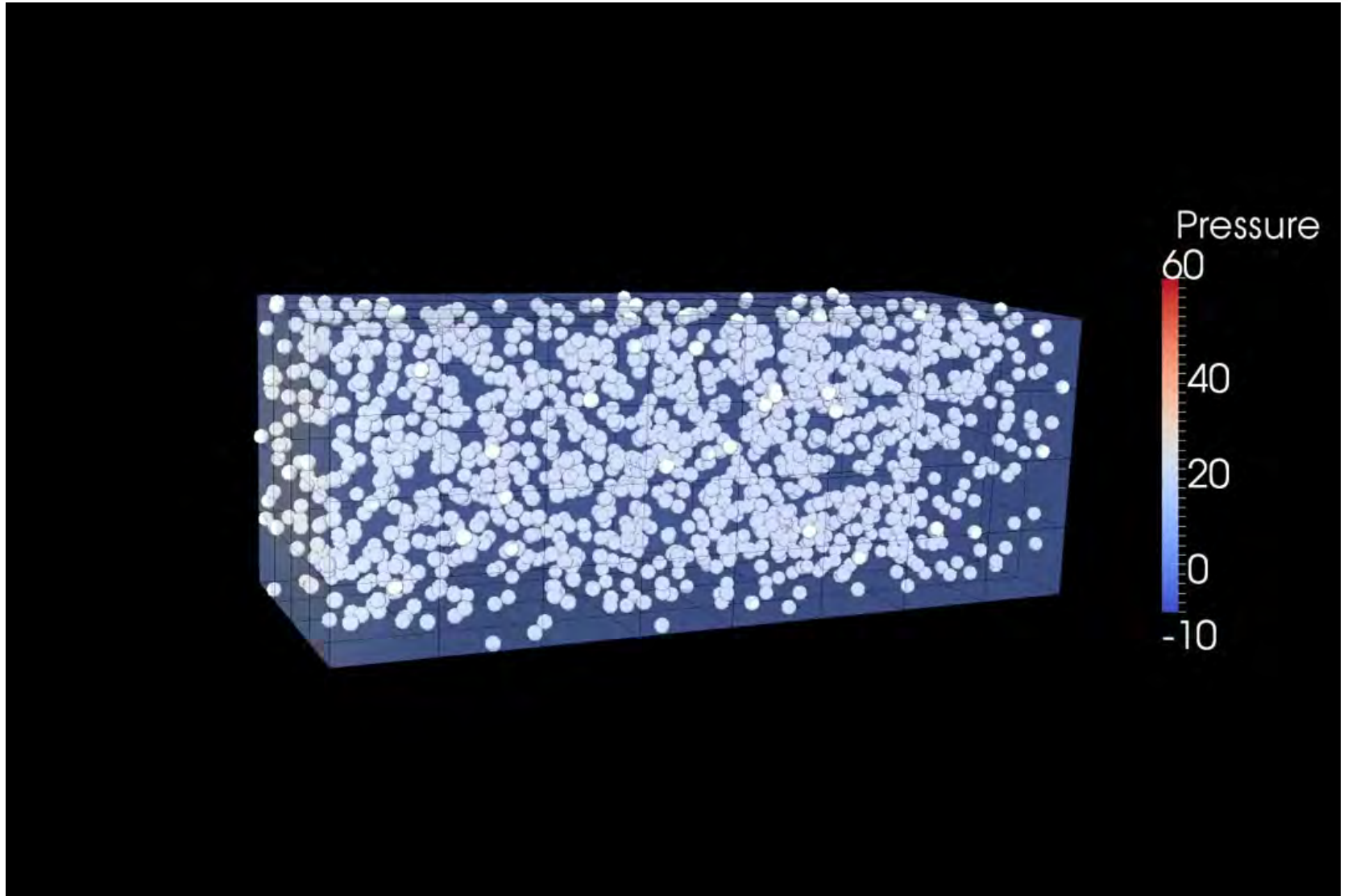
Resistance of medium defined by bulk and shear moduli.

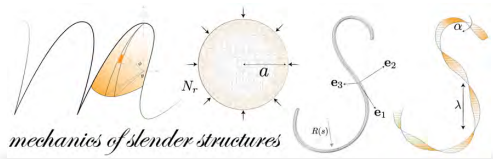
$$\left(K + \frac{4}{3} G \right) \nabla^2 \varepsilon = \alpha \nabla^2 p$$

Stress caused by (1.) hydrostatic pressure of water filling pores, and (2.) average stress in porous network. **Stresses** in the soil **carried** in part by the **fluid** and in part by **solid**.



Biot Poroelasticity



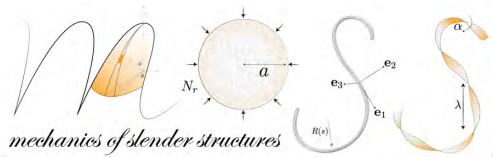


mechanics of slender structures

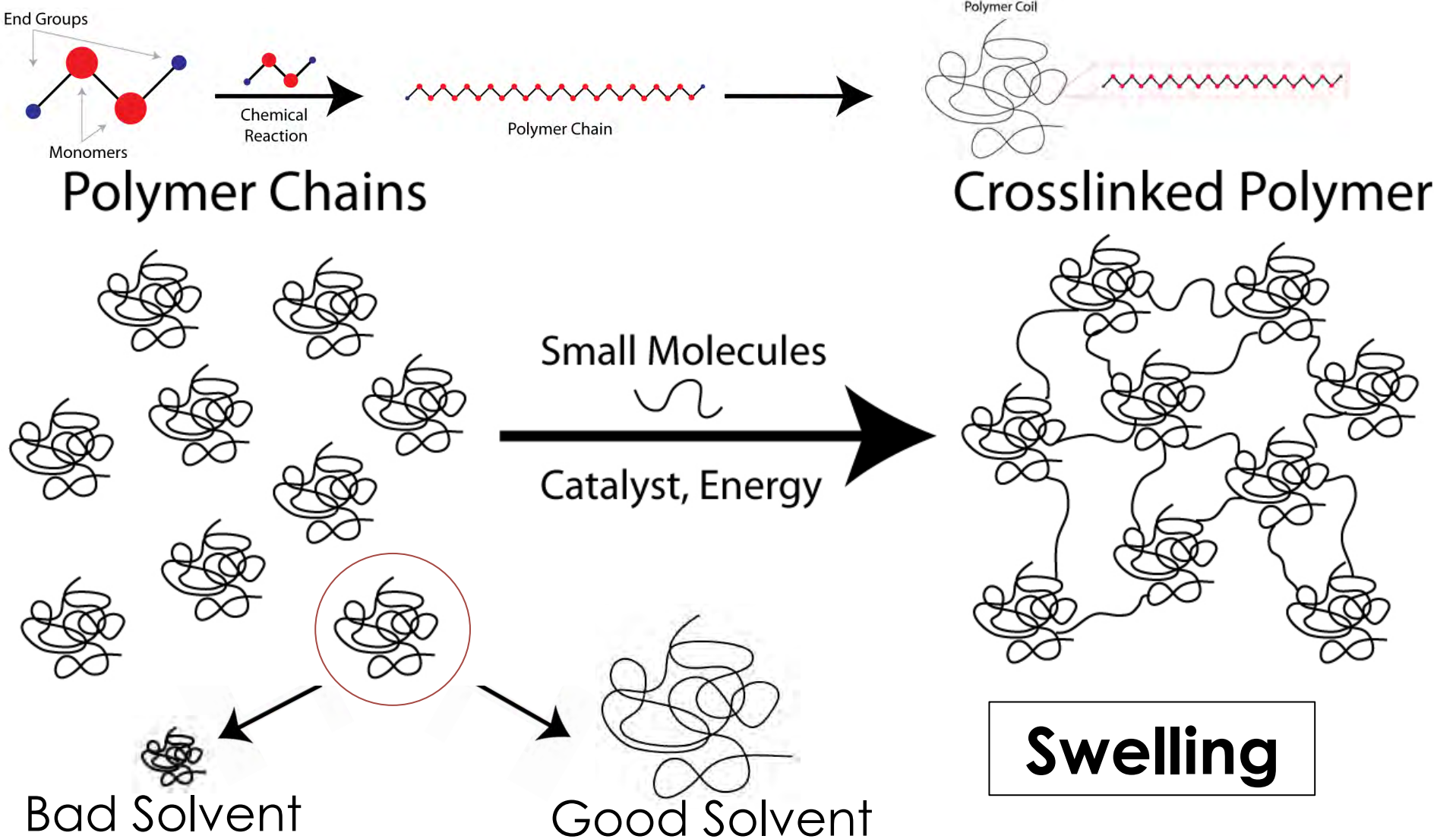
Swelling Spheres

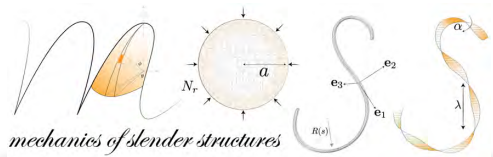


Igor30, "Play with Water Balz Balls Jumbo Polymer Hydrogel", <https://www.youtube.com/watch?v=GX2PRQi6Tdk>, 2014.



Polymers & Swelling





Gibbs Free Energy of Dilution

$$\Delta \mathcal{G} = \underbrace{\Delta H}_{\text{Heat}} - T \underbrace{\Delta S}_{\text{Entropy}}$$

Equilibrium Swelling $\Delta \mathcal{G} = 0$

At constant pressure

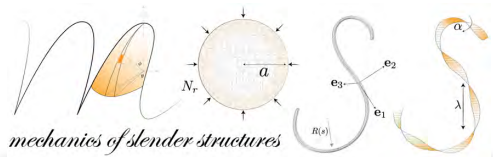
Determination via Osmotic Pressure

Excess pressure required to keep mixed phase in equilibrium with the pure liquid.

$$\Pi = -\frac{RT}{V} \ln \left(\frac{p}{p_0} \right)$$

p : Vapor pressure of liquid in equilibrium with mixture.

p_0 : Saturation pressure



Gibbs Free Energy of Dilution

Enthalpy \sim Internal Energy

Entropy of dilution (Boltzmann)

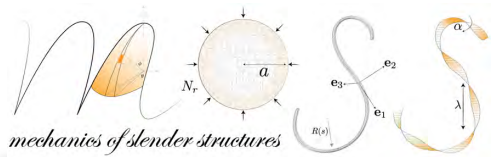
Flory-Huggins Equation

Free, long polymer chains

$$\Delta \mathcal{G} = RT \left[\ln(1 - v_2) + v_2 + \chi v_2^2 \right]$$

Flory-Huggins Chi parameter: dimensionless, polymer/fluid interactions.

Good solvents: $\chi \sim 0.1 - 0.5$



Flory-Rehner Equation

Crosslinked polymer networks

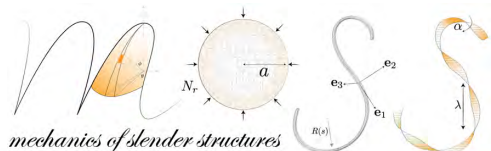
- The **entropy** change caused by **mixing** of polymer and solvent.
- The **entropy** change caused by **reduction** in numbers of possible chain **conformations** on swelling.
- The **heat of mixing** of polymer and solvent, which may be positive, negative, or zero.

Equilibrium swelling of a crosslinked network:

$$\ln(1 - v_2) + v_2 + \chi v_2^2 + \frac{\rho V_s}{M_c} v_2^{1/3} = 0$$

Approximate equilibrium stretch:

$$\lambda_{eq} \approx \left(\frac{RT}{V_s} \frac{1/2 - \chi}{G} \right)^{1/5}$$

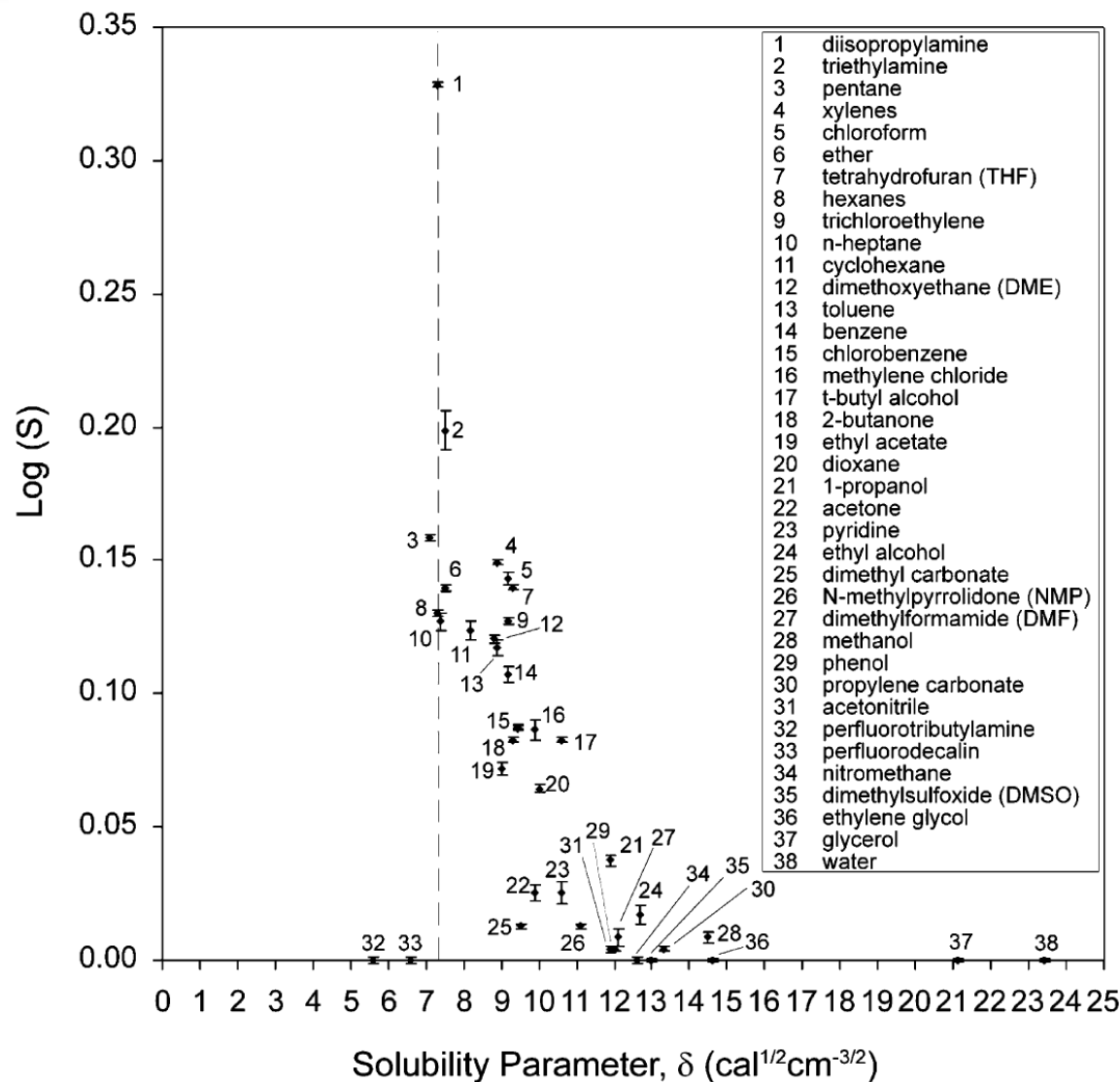


PDMS & Swelling

Table 1. Solubility Parameters, Swelling Ratios, and Dipole Moments of Various Solvents Used in Organic Synthesis

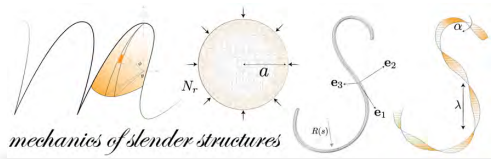
solvent	δ^a	S^b	μ (D)	ref ^c	rank ^d
perfluorotributylamine	5.6	1.00	0.0	10	32
perfluorodecalin	6.6	1.00	0.0	10	33
pentane	7.1	1.44	0.0	10	3
poly(dimethylsiloxane)	7.3	∞	0.6-0.9	8, 14	
diisopropylamine	7.3	2.13	1.2	10	1
hexanes	7.3	1.35	0.0	10	8
n-heptane	7.4	1.34	0.0	10	10
triethylamine	7.5	1.58	0.7	8,10	2
ether	7.5	1.38	1.1	10	6
cyclohexane	8.2	1.33	0.0	10	11
trichloroethylene	9.2	1.34	0.9	10	9
dimethoxyethane (DME)	8.8	1.32	1.6	10	12
xylenes	8.9	1.41	0.3	10	4
toluene	8.9	1.31	0.4	10	13
ethyl acetate	9.0	1.18	1.8	8,10	19
benzene	9.2	1.28	0.0	10	14
chloroform	9.2	1.39	1.0	10	5
2-butanone	9.3	1.21	2.8	10	18
tetrahydrofuran (THF)	9.3	1.38	1.7	10	7
dimethyl carbonate	9.5	1.03	0.9	8,10	25
chlorobenzene	9.5	1.22	1.7	10	15
methylene chloride	9.9	1.22	1.6	10	16
acetone	9.9	1.06	2.9	8,12	22
dioxane	10.0	1.16	0.5	10	20
pyridine	10.6	1.06	2.2	10	23
N-methylpyrrolidone (NMP)	11.1	1.03	3.8	10	26
tert-butyl alcohol	10.6	1.21	1.6	8,12	17
acetonitrile	11.9	1.01	4.0	10	31
1-propanol	11.9	1.09	1.6	8,10	21
phenol	12.0	1.01	1.2	8,12	29
dimethylformamide (DMF)	12.1	1.02	3.8	8,10	27
nitromethane	12.6	1.00	3.5	10	34
ethyl alcohol	12.7	1.04	1.7	8,12	24
dimethyl sulfoxide (DMSO)	13.0	1.00	4.0	10	35
propylene carbonate	13.3	1.01	4.8	10	30
methanol	14.5	1.02	1.7	8,12	28
ethylene glycol	14.6	1.00	2.3	8,12	36
glycerol	21.1	1.00	2.6	13,15	37
water	23.4	1.00	1.9	8,12	38

^a δ in units of $\text{cal}^{1/2} \text{cm}^{-3/2}$, ^b S denotes the swelling ratio that was measured experimentally: $S = D/D_0$, where D is the length of PDMS in the solvent and D_0 is the length of the dry PDMS. ^c References refer to literature values of δ and μ . ^d Rank refers to the order of the solvent in decreasing swelling ability (see Figure 1).

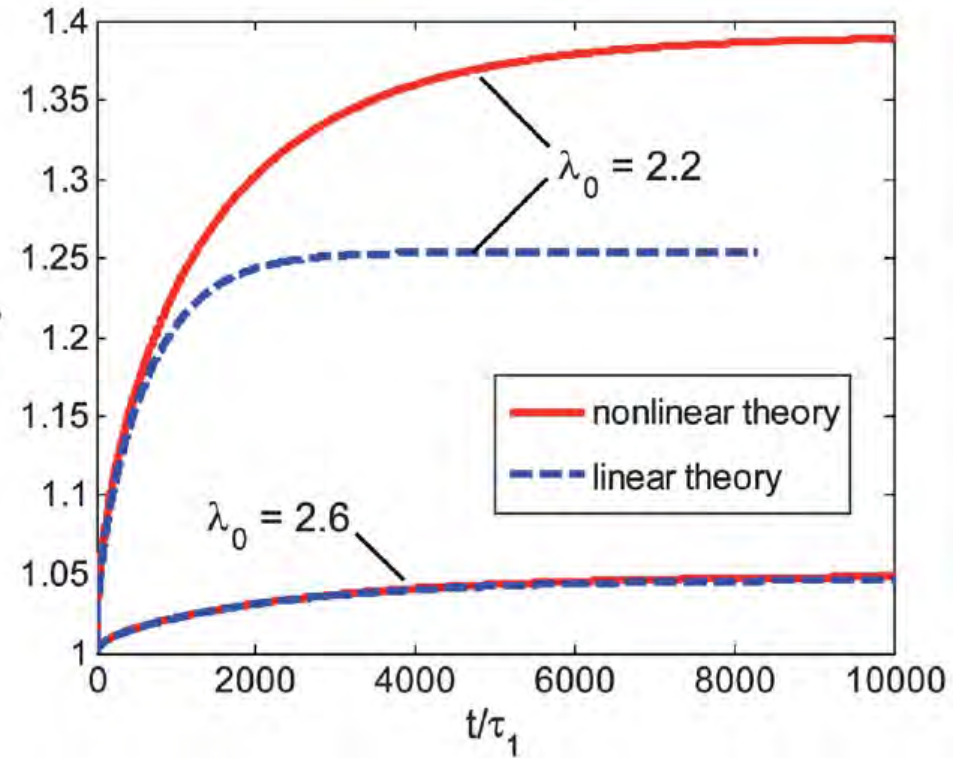
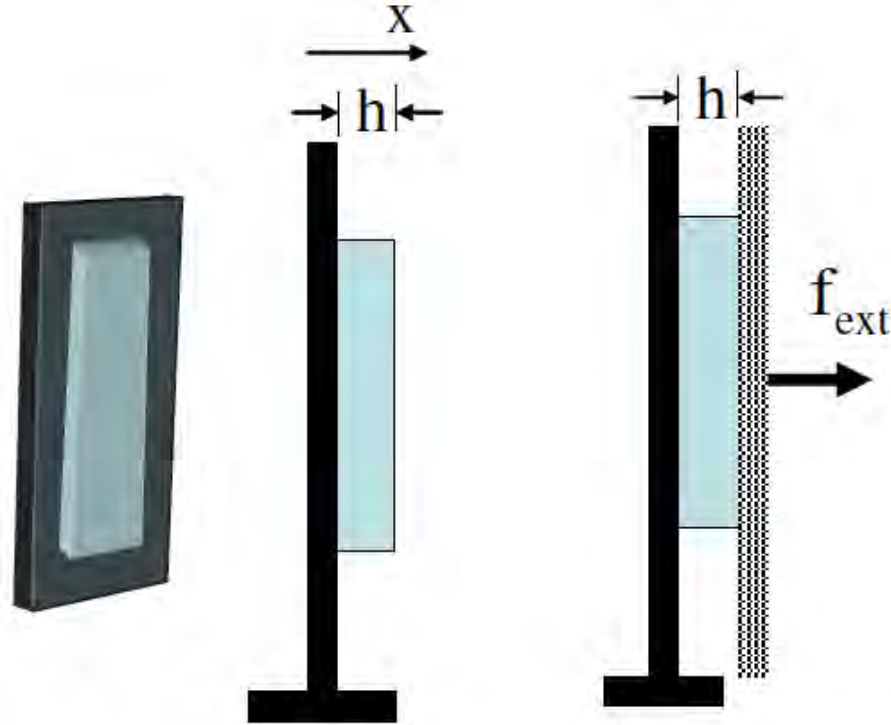


- 1 diisopropylamine
- 2 triethylamine
- 3 pentane
- 4 xylenes
- 5 chloroform
- 6 ether
- 7 tetrahydrofuran (THF)
- 8 hexanes
- 9 trichloroethylene
- 10 n-heptane
- 11 cyclohexane
- 12 dimethoxyethane (DME)
- 13 toluene
- 14 benzene
- 15 chlorobenzene
- 16 methylene chloride
- 17 t-butyl alcohol
- 18 2-butanone
- 19 ethyl acetate
- 20 dioxane
- 21 1-propanol
- 22 acetone
- 23 pyridine
- 24 ethyl alcohol
- 25 dimethyl carbonate
- 26 N-methylpyrrolidone (NMP)
- 27 dimethylformamide (DMF)
- 28 methanol
- 29 phenol
- 30 propylene carbonate
- 31 acetonitrile
- 32 perfluorotributylamine
- 33 perfluorodecalin
- 34 nitromethane
- 35 dimethylsulfoxide (DMSO)
- 36 ethylene glycol
- 37 glycerol
- 38 water

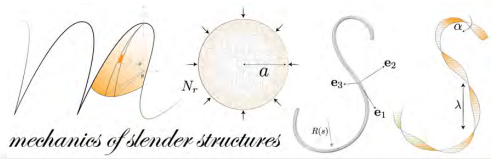




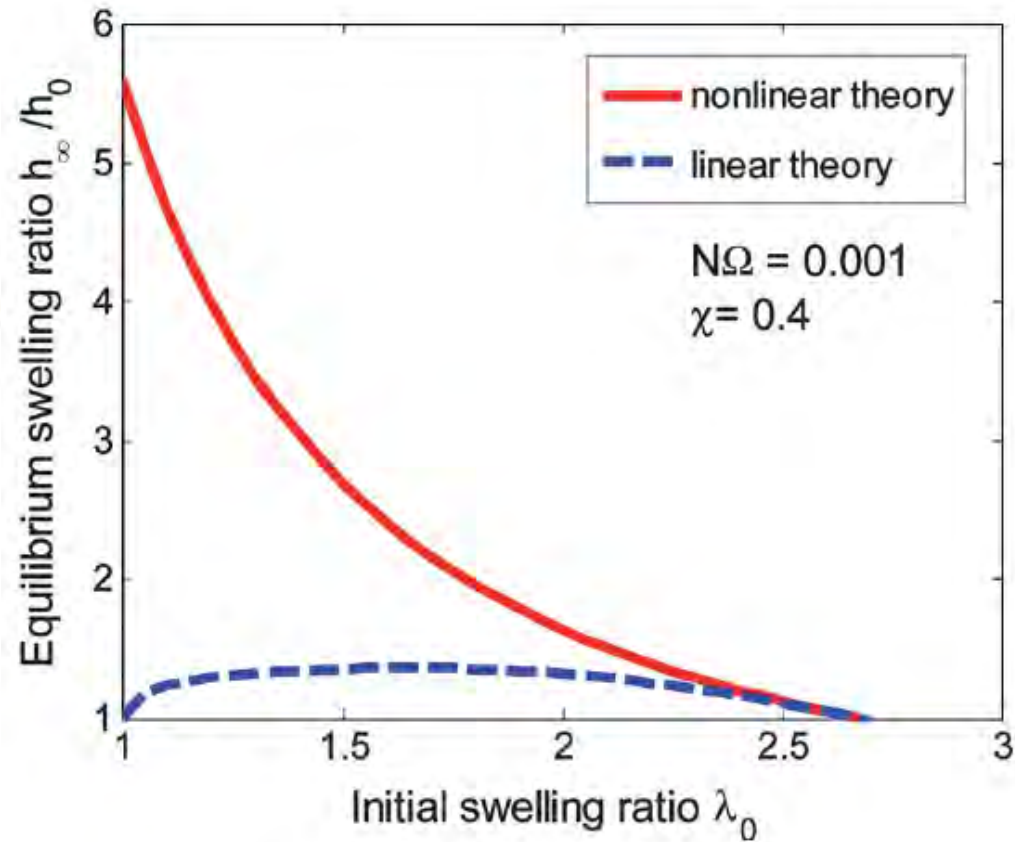
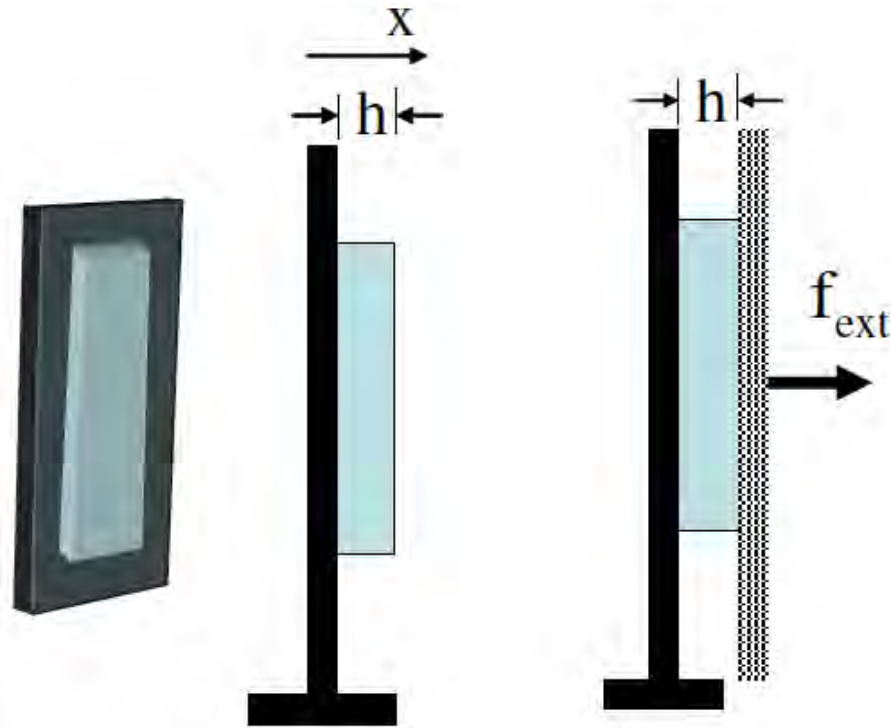
Swelling a Disk

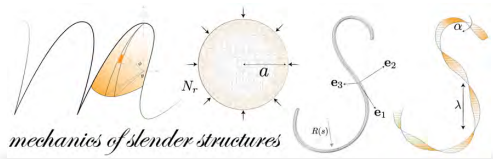


λ is the initial swelling ratio.



Swelling a Disk





Swelling a Sphere

Swelling Dynamics

Linearized, similar to poroelasticity

$$\left(K + \frac{4}{3}G \right) \nabla \nabla \cdot \mathbf{u} + G \nabla^2 \mathbf{u} = \nabla^2 p$$

Incompressibility & Darcy's law

$$\nabla \cdot \mathbf{u} = k \nabla^2 p$$

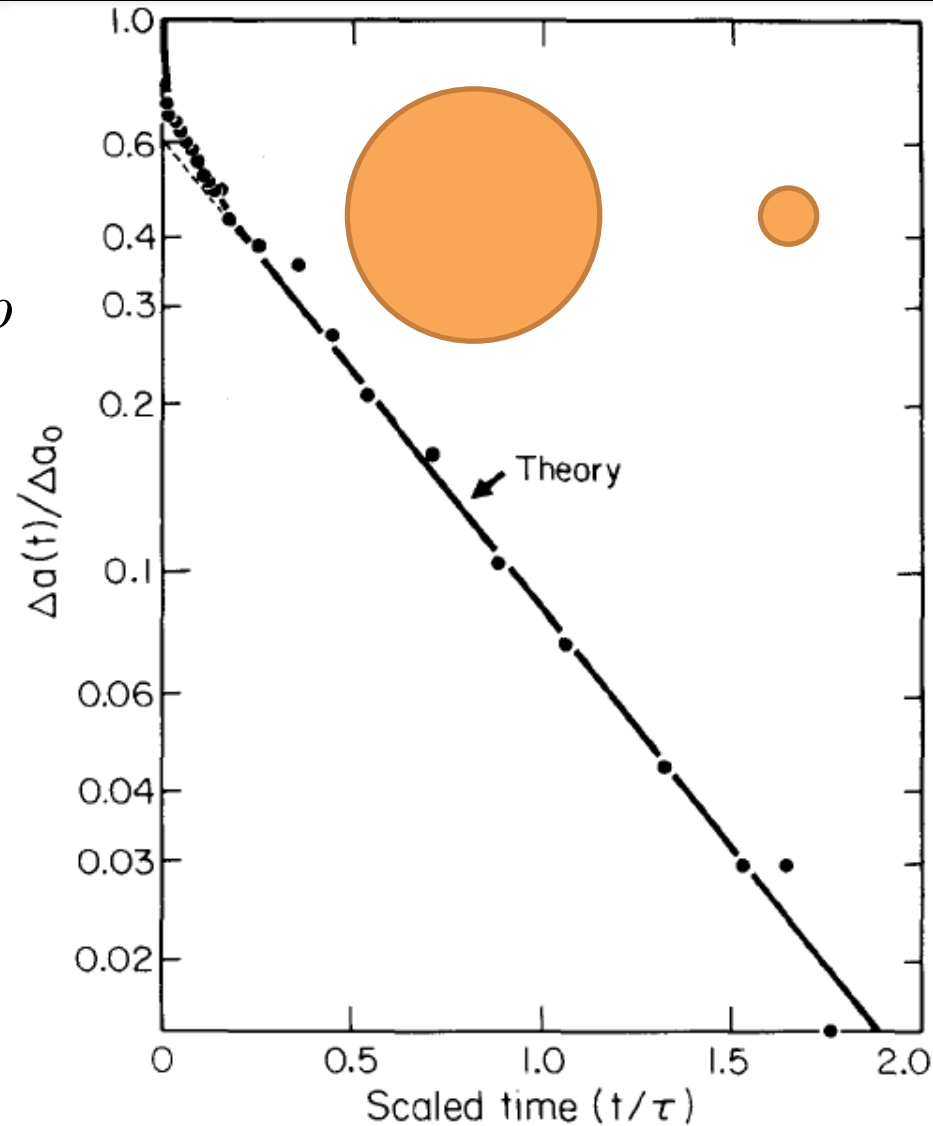
Volume change (e.g. sphere)

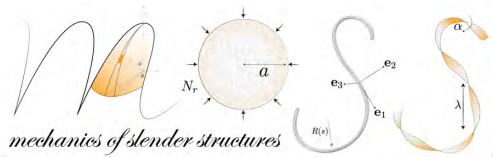
$$\underbrace{\alpha(\mathbf{x}, t)}_{\text{Volume change}} = \nabla \cdot \mathbf{u}$$

Volume change

Satisfied by diffusion relation:

$$\frac{\partial \alpha}{\partial t} = D \nabla^2 \alpha \quad D \equiv \left(K + \frac{4}{3}G \right) k$$





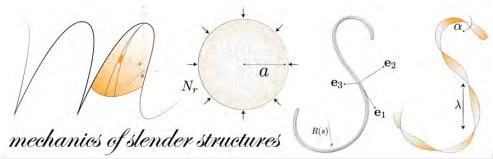
Swelling of Elastic Materials

Fluids Deforming Solids

Douglas P. Holmes

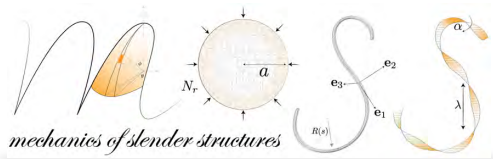
Mechanical Engineering
Boston University



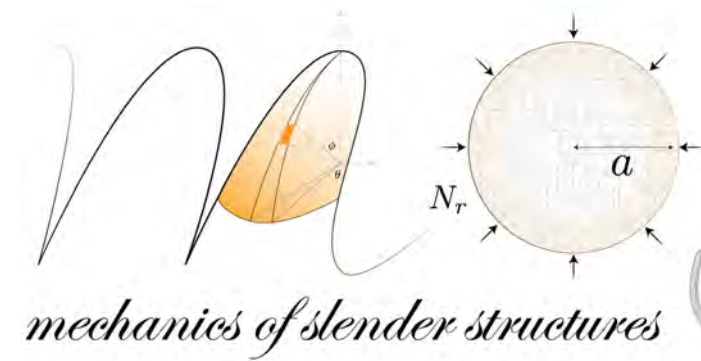


Gelatin cubes dropped onto solid surface High Speed Video 6200 fps

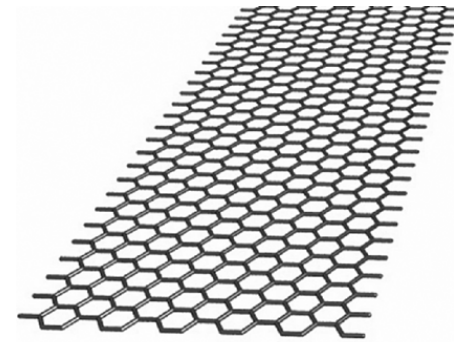
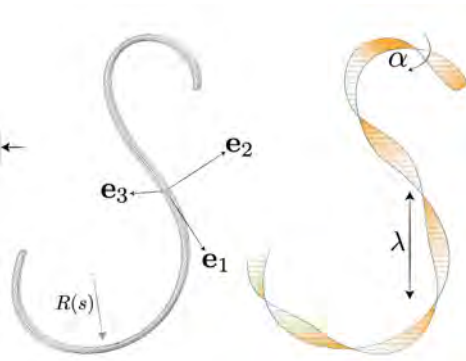
ModernistCuisine, "Gelatin cubes dropped onto solid surface High Speed Video 6200 fps", <https://www.youtube.com/watch?v=4n5AfHYST6E>, 2011.



mechanics of slender structures



mechanics of slender structures

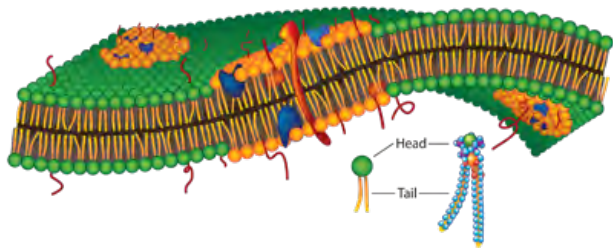


(a)



(b)

Source: Kreupl et al. (2004)



<http://www.lanl.gov/science/1663/august2011/images/Cell-Wave-Final.png>



<http://isabelleleo.deviantart.com/art/Just-hair-292904304>

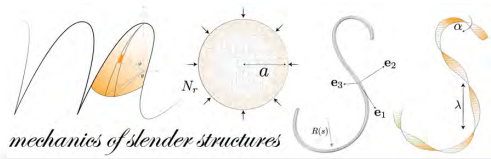


<http://www.contactlensescomparison.com/wp-content/themes/smallbiz/images/lens.jpg>

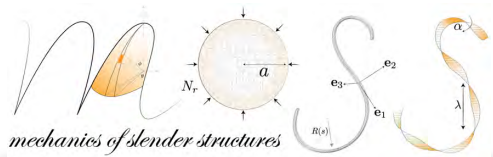
Geometric Non-linearities:

- **Buckling**
- Wrinkling
- **Folding**
- Creasing
- **Snapping**

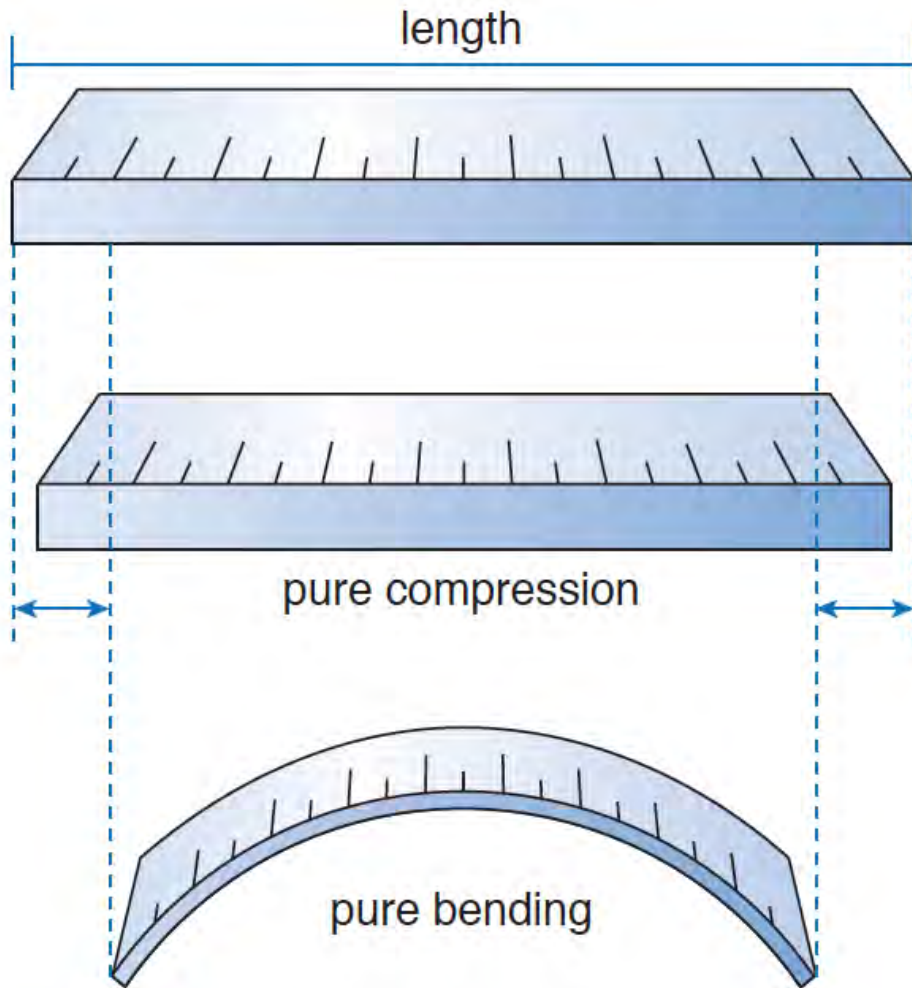
How do objects change shape?



How do you
“grow” a **structure** into
a desired shape?



Thin Structures



Bending vs. Stretching

E – Elastic Modulus

h – thickness

$\varepsilon_{\alpha\beta}$ – in-plane strain

κ – curvature

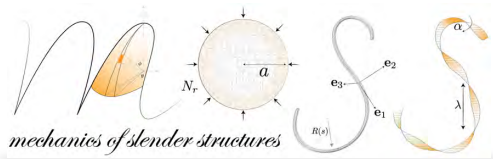
$$U_m \sim E h \varepsilon_{\alpha\beta}^2$$

Energy in Compression \sim thickness

$$U_b \sim E h^3 \kappa^2$$

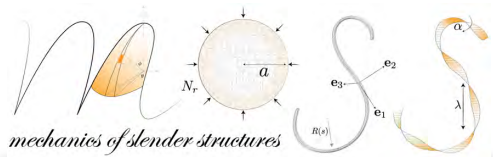
Energy in Bending \sim thickness³

Thin structures deform by **bending** & avoid **stretching**



A still photo is a kind of lie

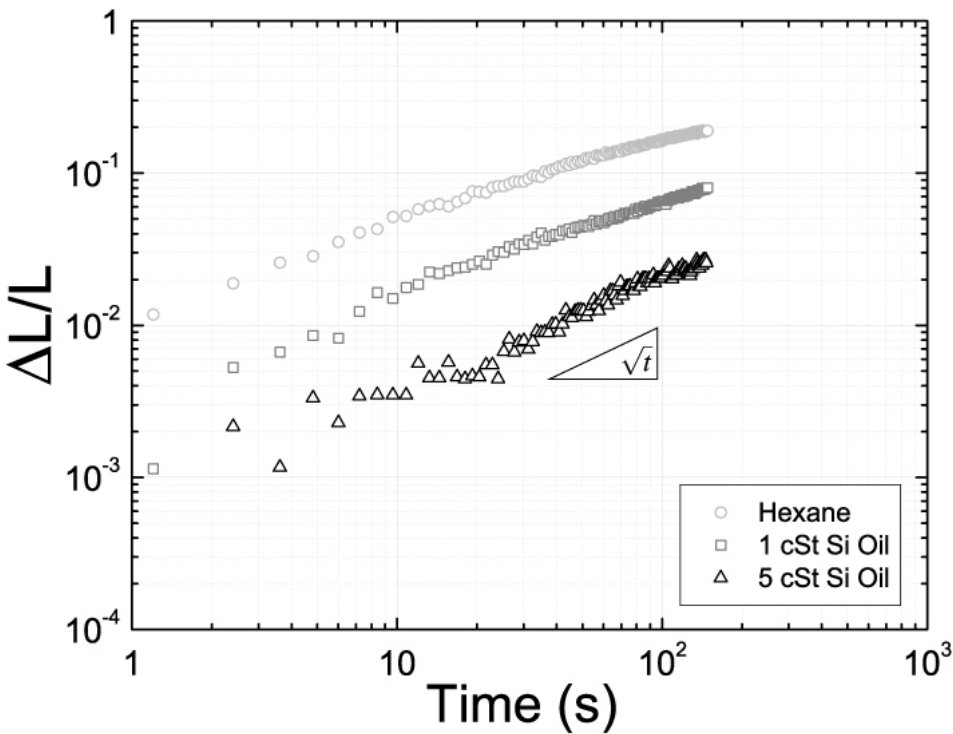
PETER DOIG



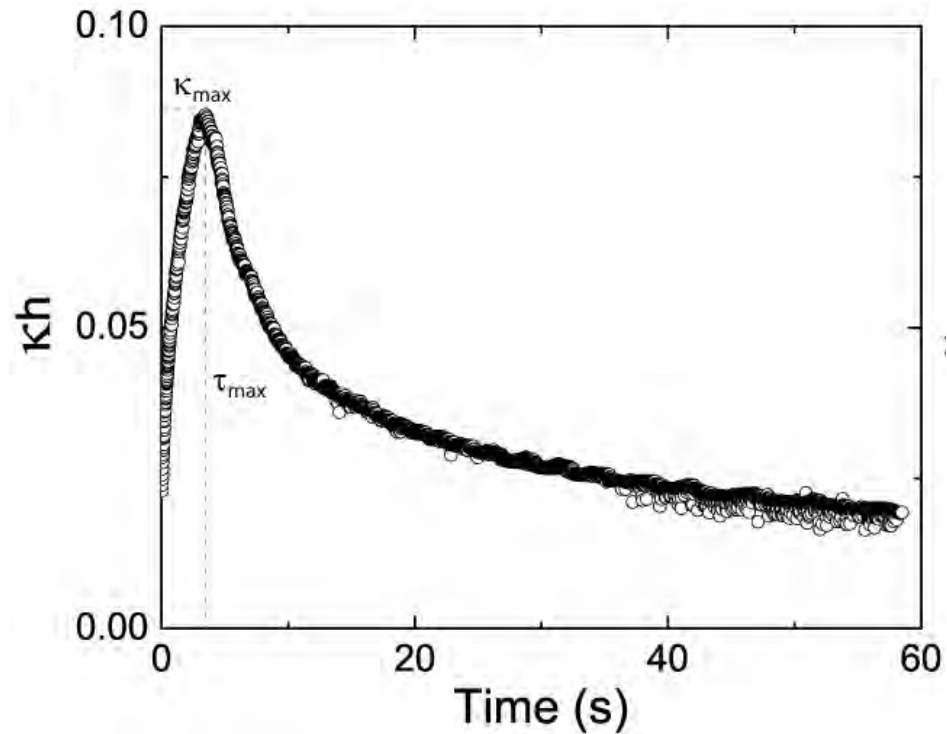
Swelling Dynamics

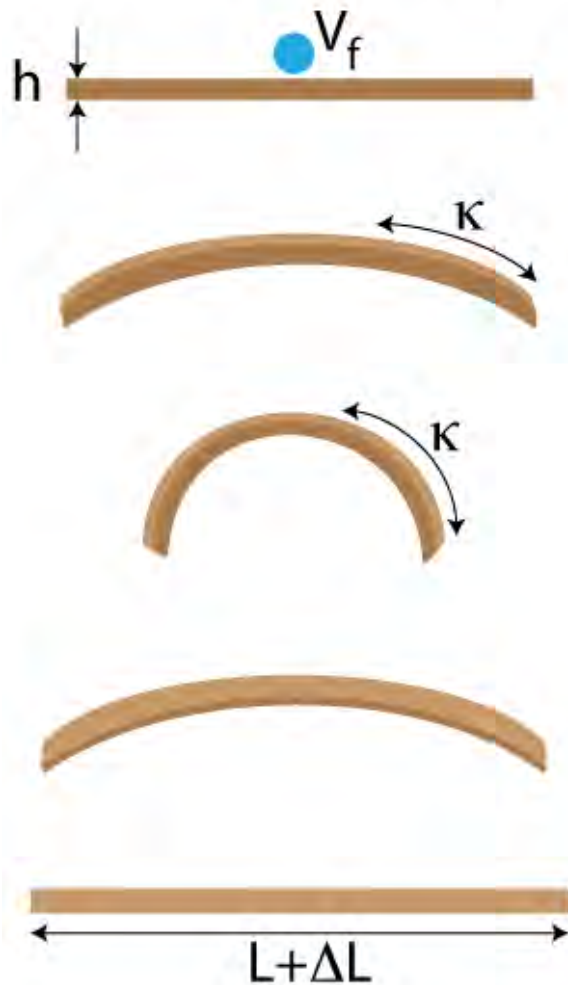


Diffusive-like Dynamics



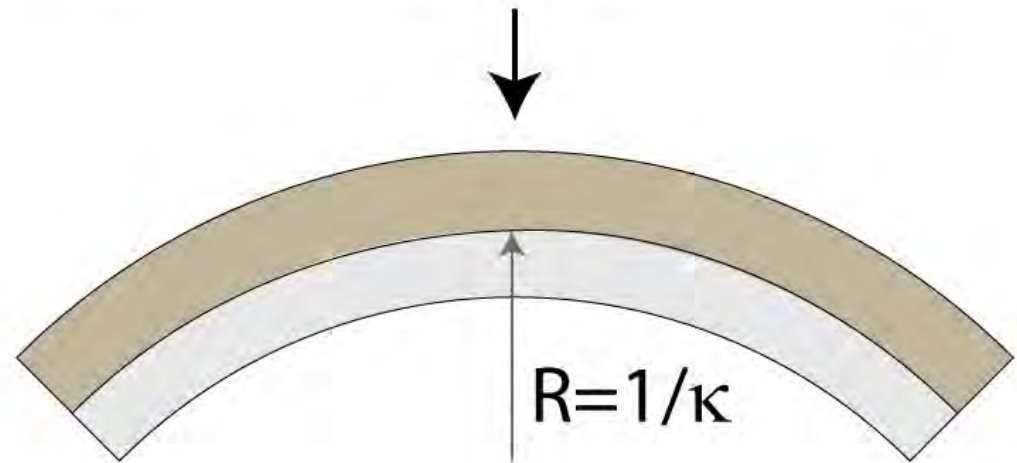
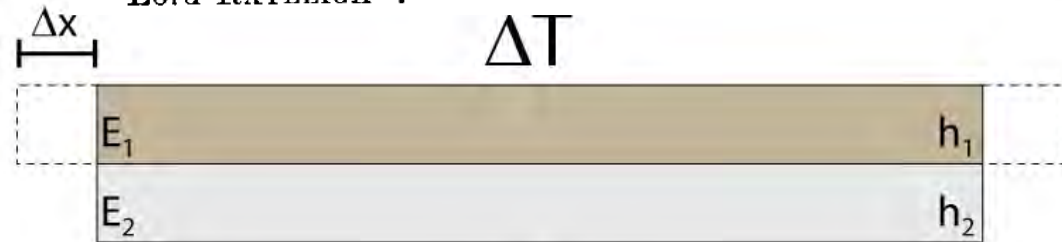
Beam Bending



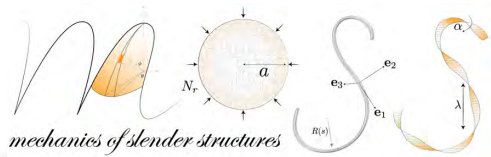


FEBRUARY 1901.

XV. *On the Stresses in Solid Bodies due to unequal Heating, and on the Double Refraction resulting therefrom.* By Lord RAYLEIGH*.

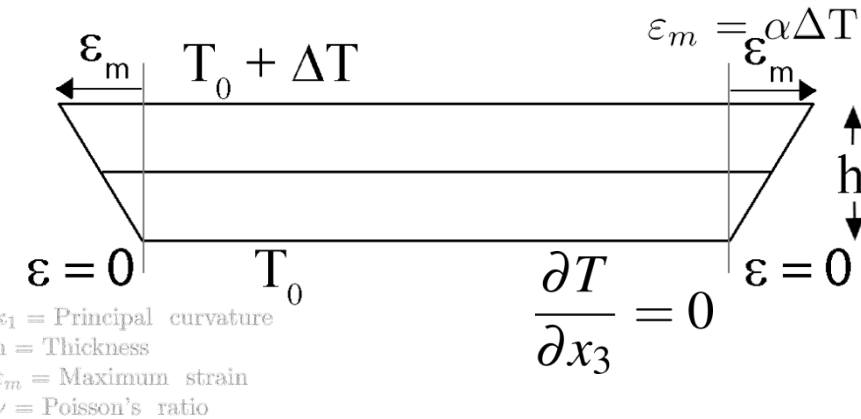


Dynamics of bending : Diffusion of temperature



Diffusion

- Thermal diffusion through the beam thickness.
- Shape obtained by minimizing the bending moment in the beam.
- Beam curvature as temperature diffuses.



Beam curvature as solvent diffuses:

$$\frac{\kappa_1 h}{\varepsilon_m (1 + \nu)} = 1.33e^{-\frac{\pi^2 t / \tau}{4}} - 0.77e^{-\frac{9\pi^2 t / \tau}{4}} + \dots$$

Poroeelastic time scale:

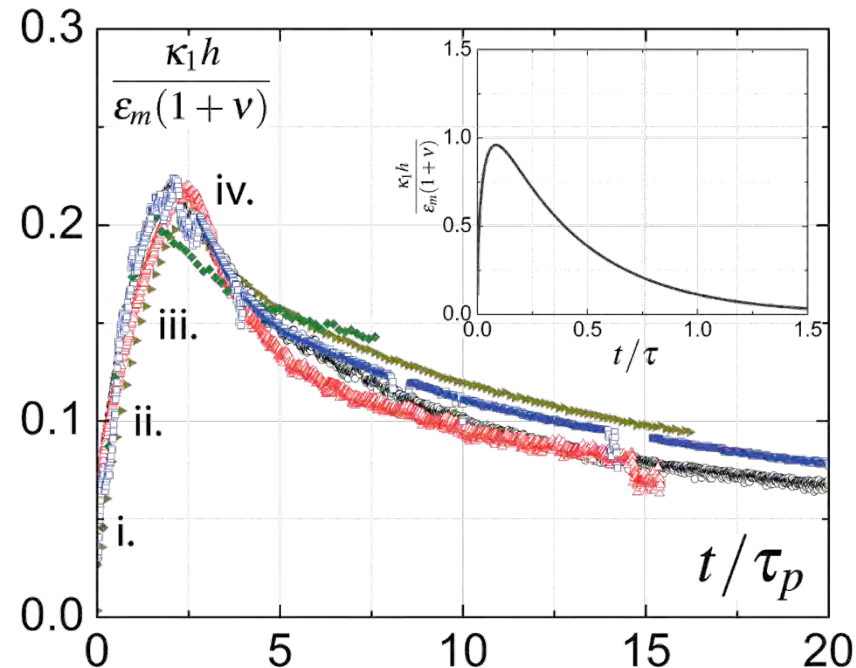
$$\tau_p \approx \frac{\mu h^2}{kE}$$

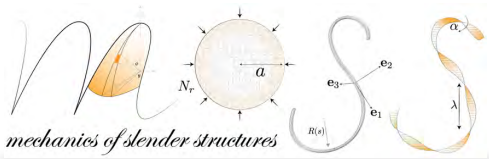
μ = Solvent viscosity

h = Thickness

k = Permeability ($k \approx 10^{-18} \text{ m}^2/\text{s}$)

E = Elastic modulus ($E = 10^6 \text{ Pa}$)





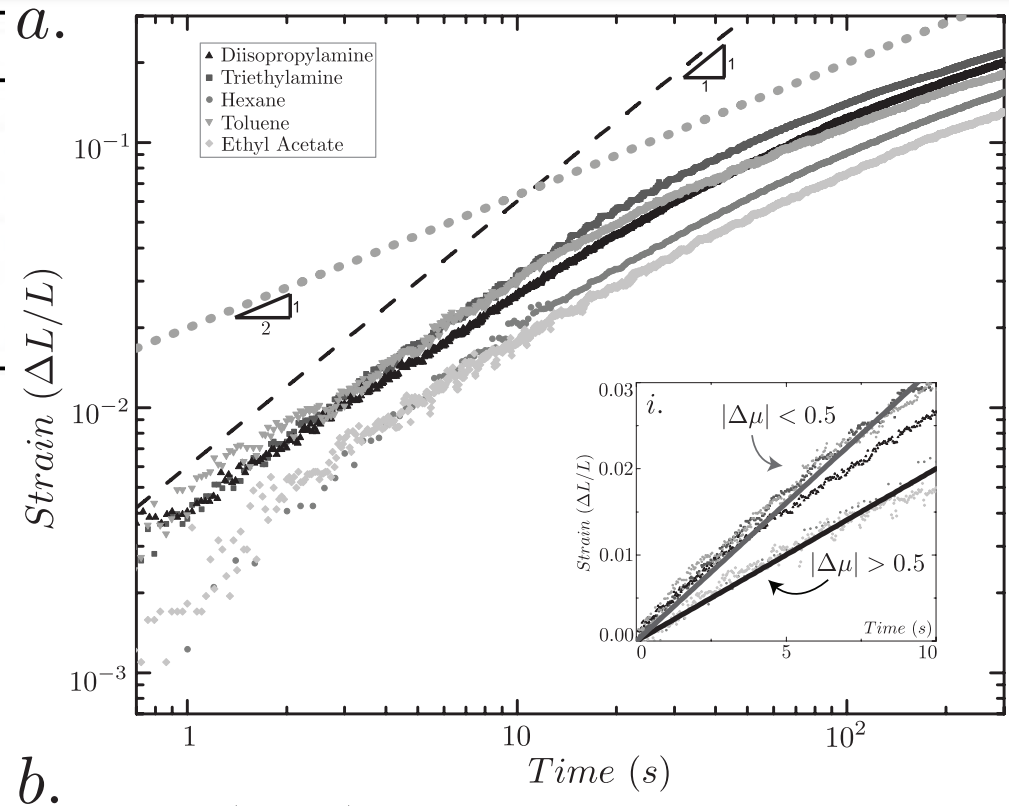
Swelling

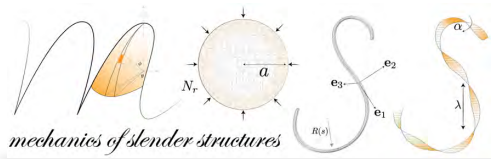
Material	δ_s ($cal^{1/2}cm^{-3/2}$)	μ (D)	ϵ_{eq}
PDMS	7.3	0.6-0.9	–
Diisopropylamine	7.3	1.2	1.13
Triethylamine	7.5	0.7	0.58
Hexanes	7.3	0.0	0.35
Toluene	8.9	0.4	0.31
Ethyl acetate	9.0	1.8	0.18

δ_s = Solubility parameter

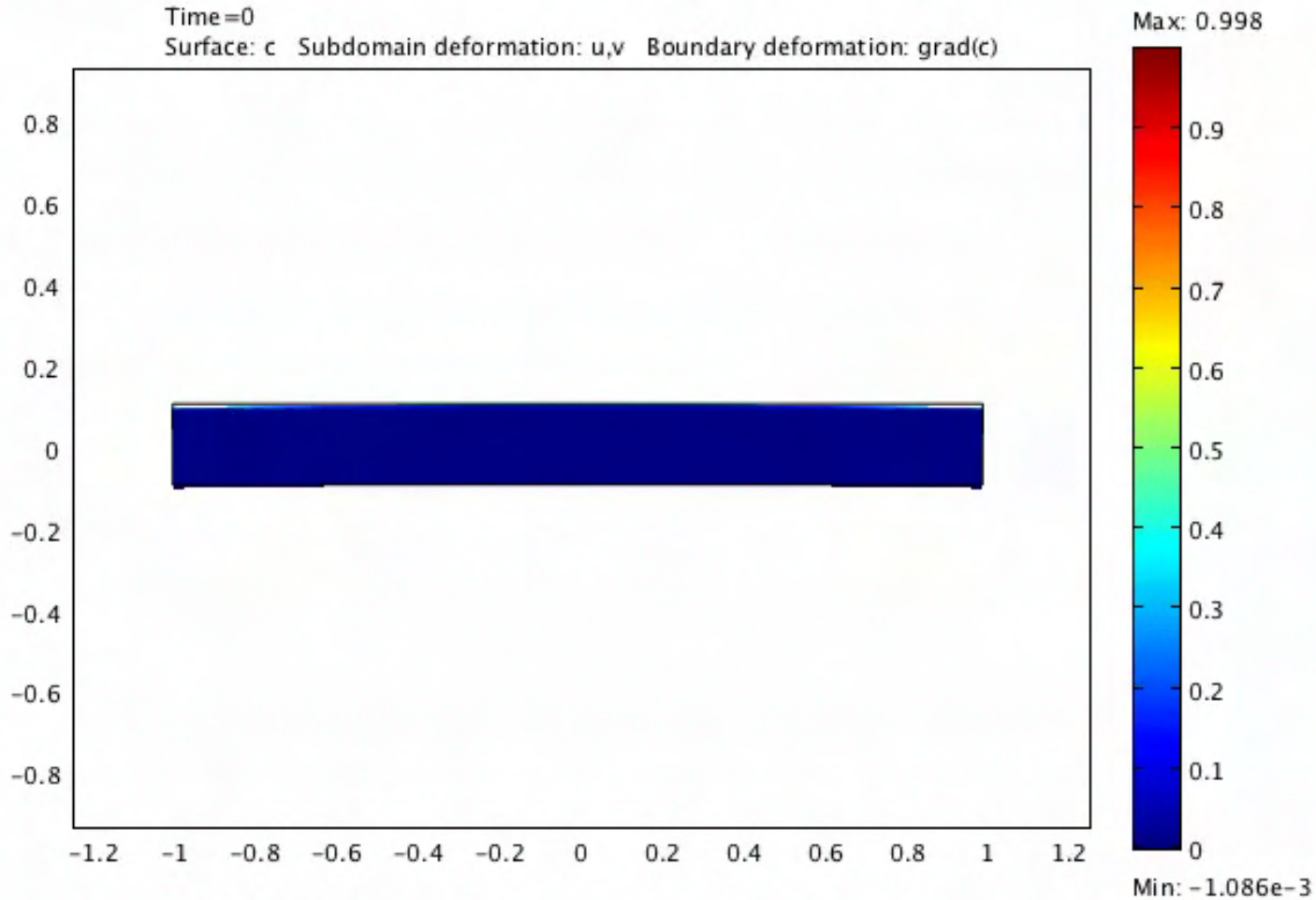
μ = Solvent polarity

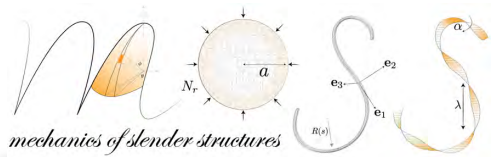
ϵ_{eq} = Strain at equilibrium



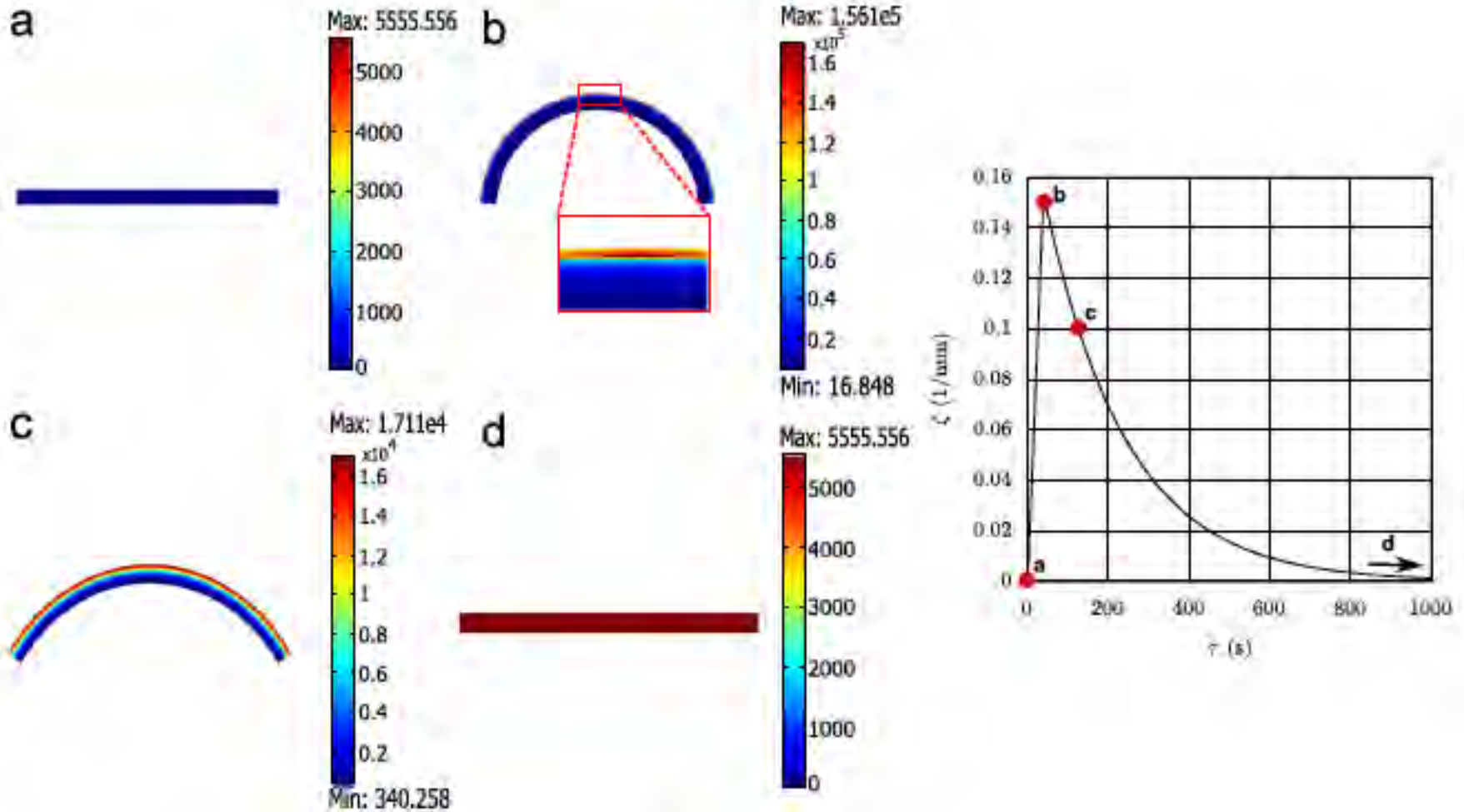


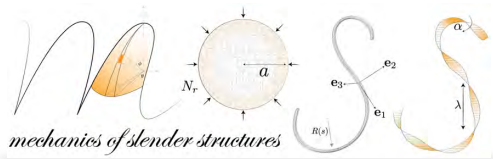
Nonlinear Swelling



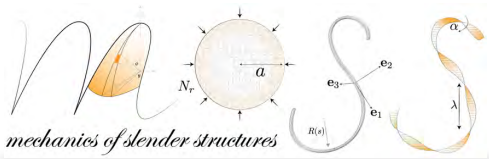


Nonlinear Swelling

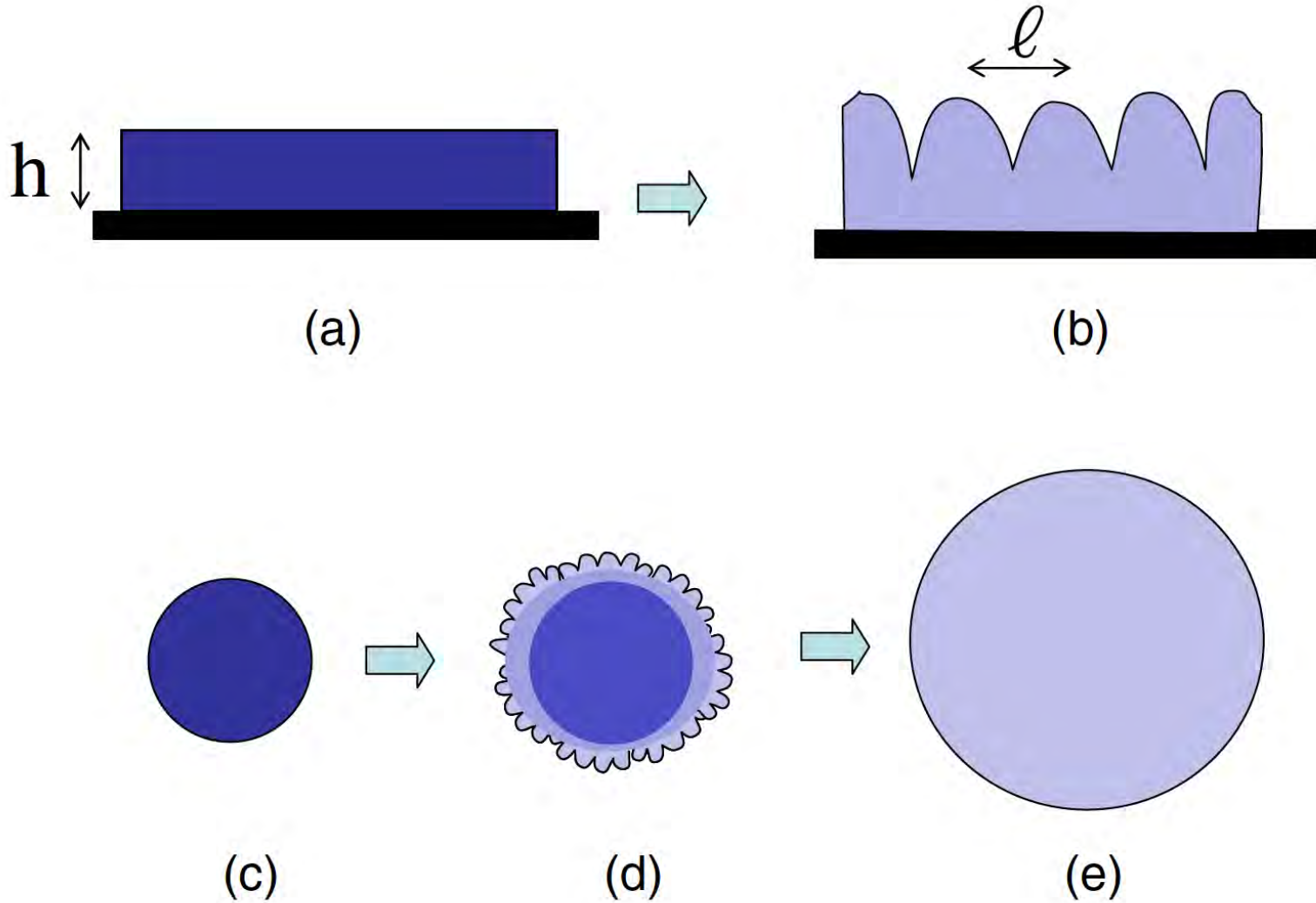


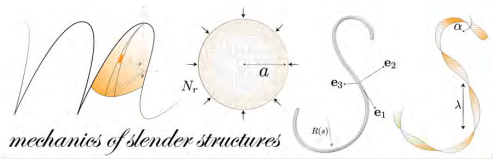


What happens when you
swell a thicker beam?

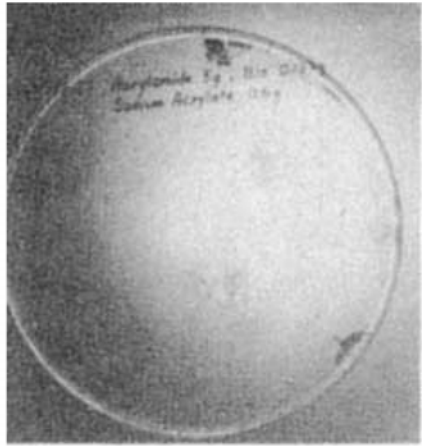


Mechanical Instability

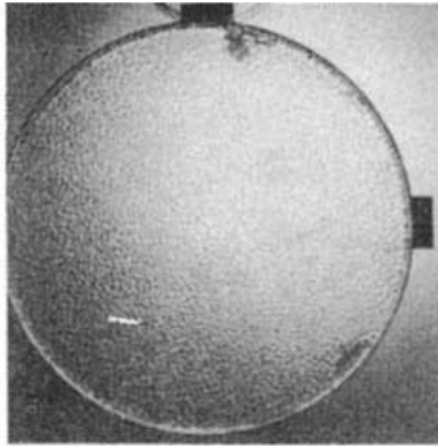




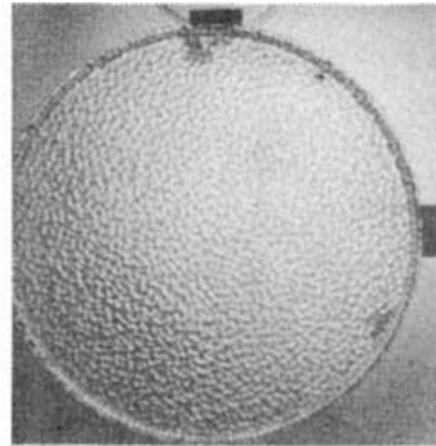
Mechanical Instability



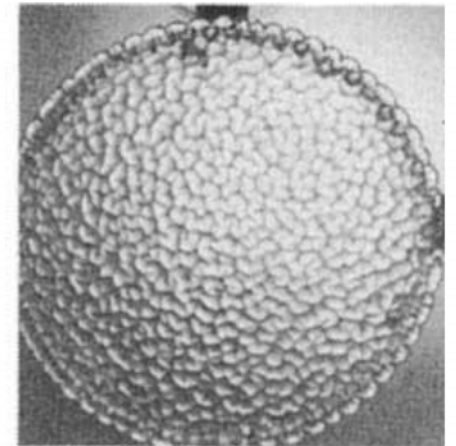
a



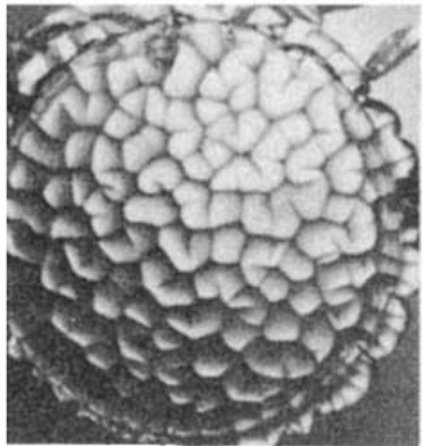
b



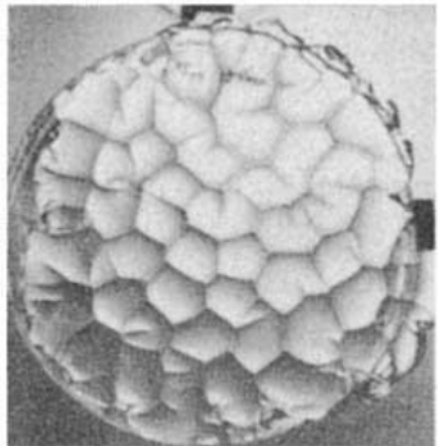
c



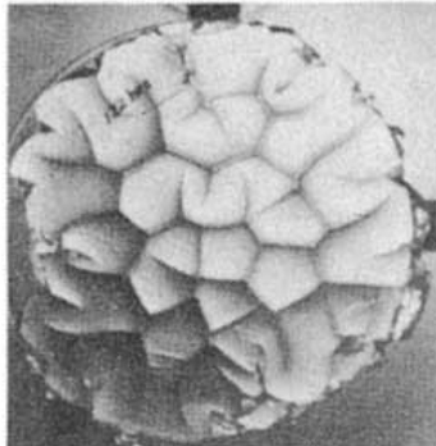
d



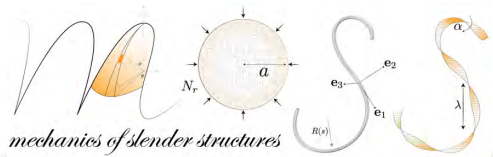
e



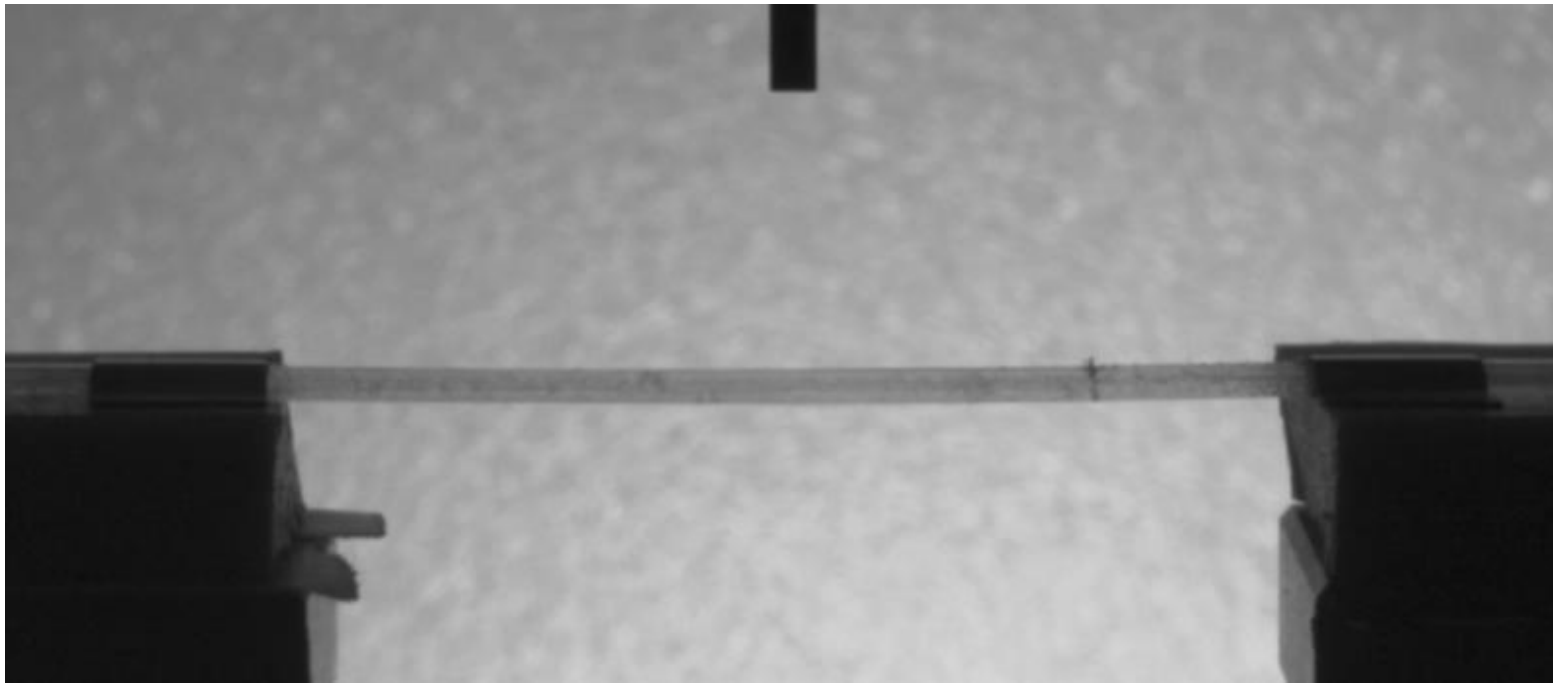
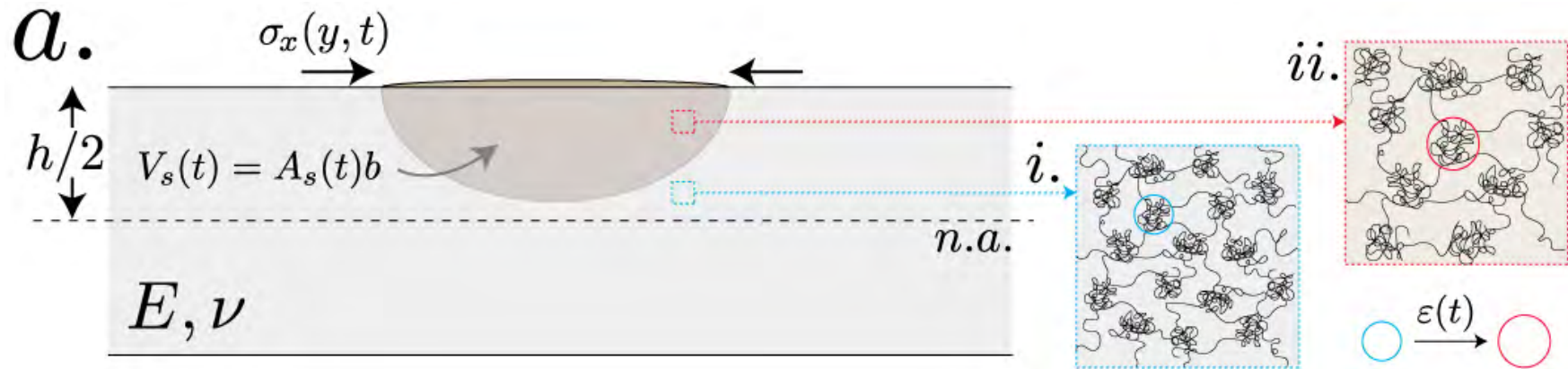
f



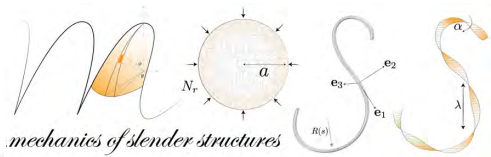
g



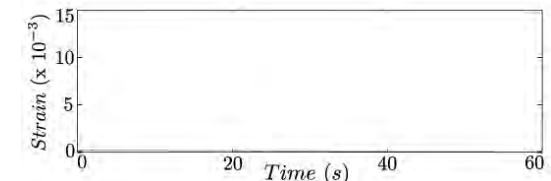
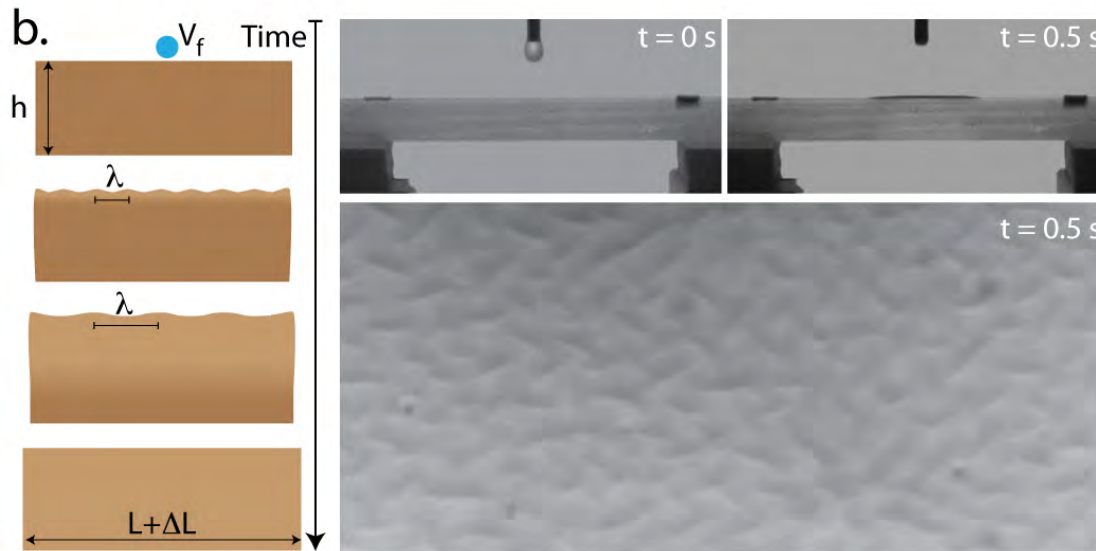
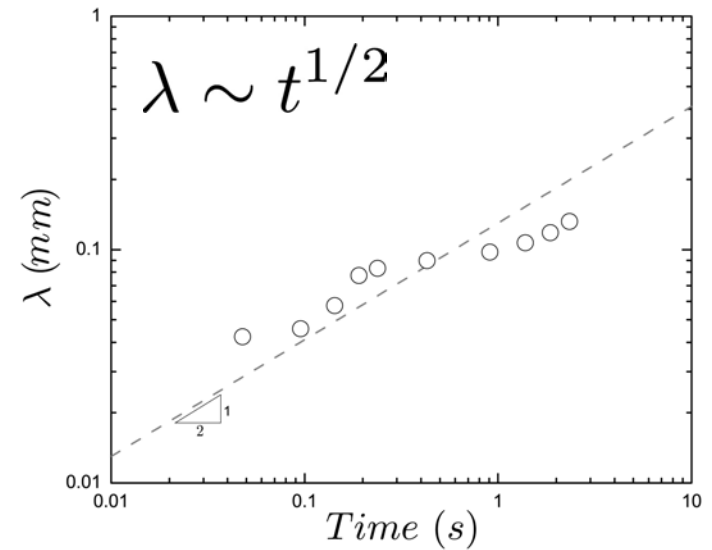
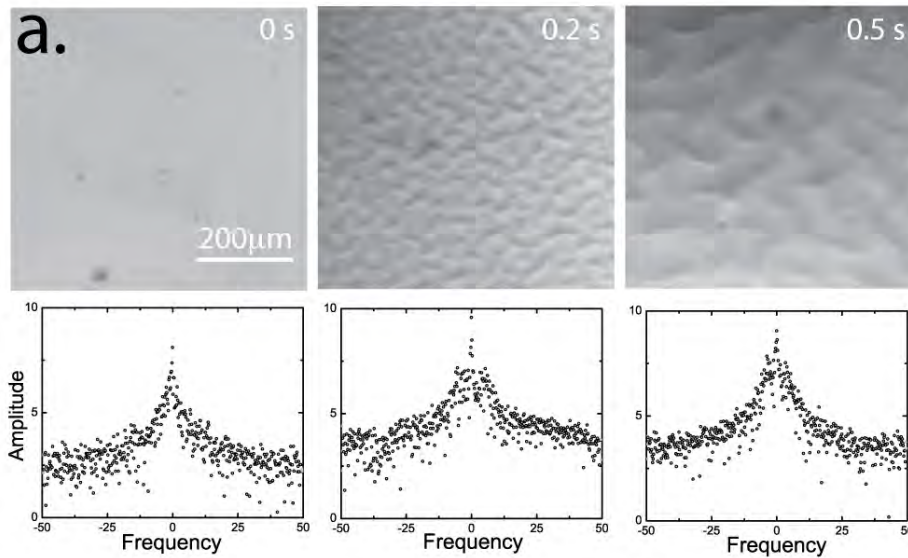
Bending and Buckling



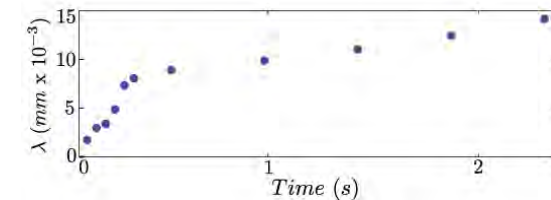
A. Pandey and D.P. Holmes. "Swelling-Induced Deformations: A Materials-Defined Transition from Structural Instability to Surface Instability," *Soft Matter*, **9**, 5524, (2013).



Bending and Buckling



No Structural Deformation



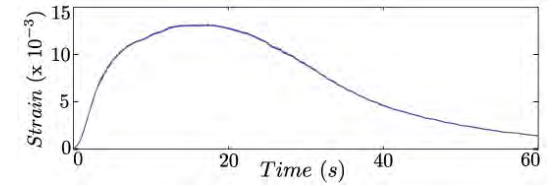
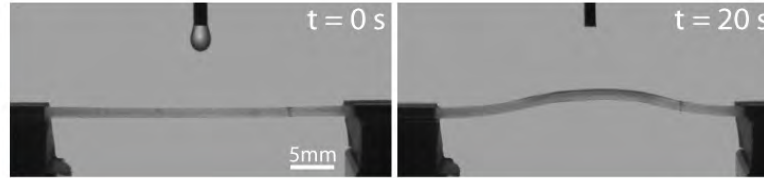
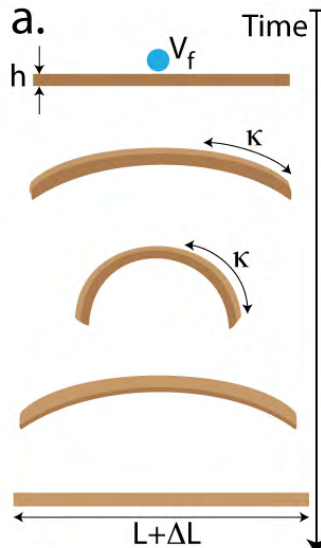
Surface Deformation

A. Pandey and D.P. Holmes. "Swelling-Induced Deformations: A Materials-Defined Transition from Structural Instability to Surface Instability," *Soft Matter*, **9**, 5524, (2013).

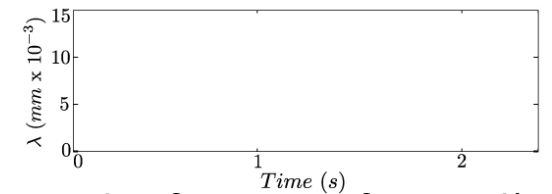
H. Tanaka, H. Tomita, A. Takasu, T. Hayashi, and T. Nishi. *Physical Review Letters*, **68**, 18, 1992.



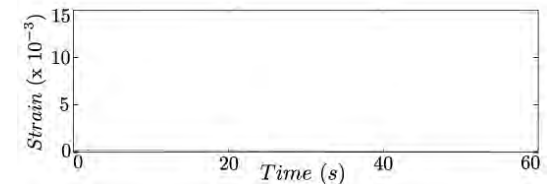
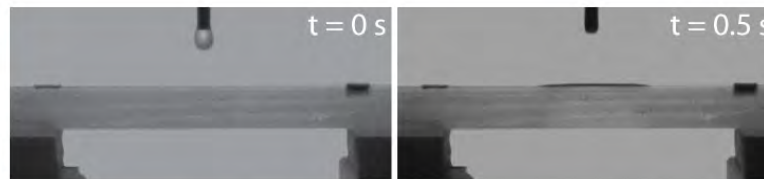
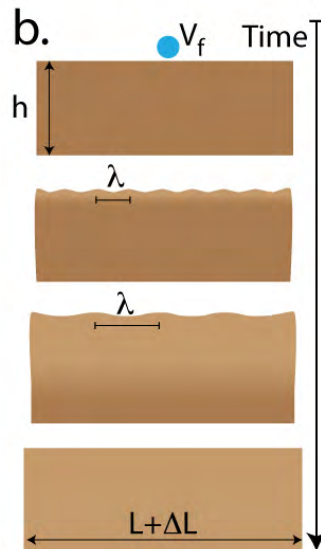
Bending and Buckling



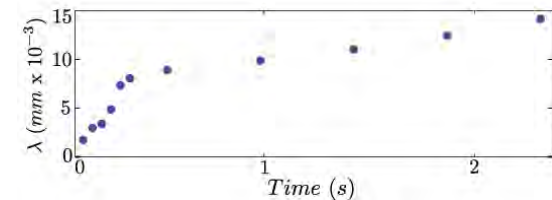
Structural Deformation



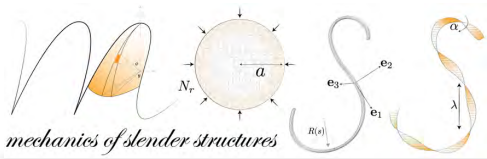
No Surface Deformation



No Structural Deformation



Surface Deformation



Bending vs. Swelling

Can the fluid bend the structure?

Bending

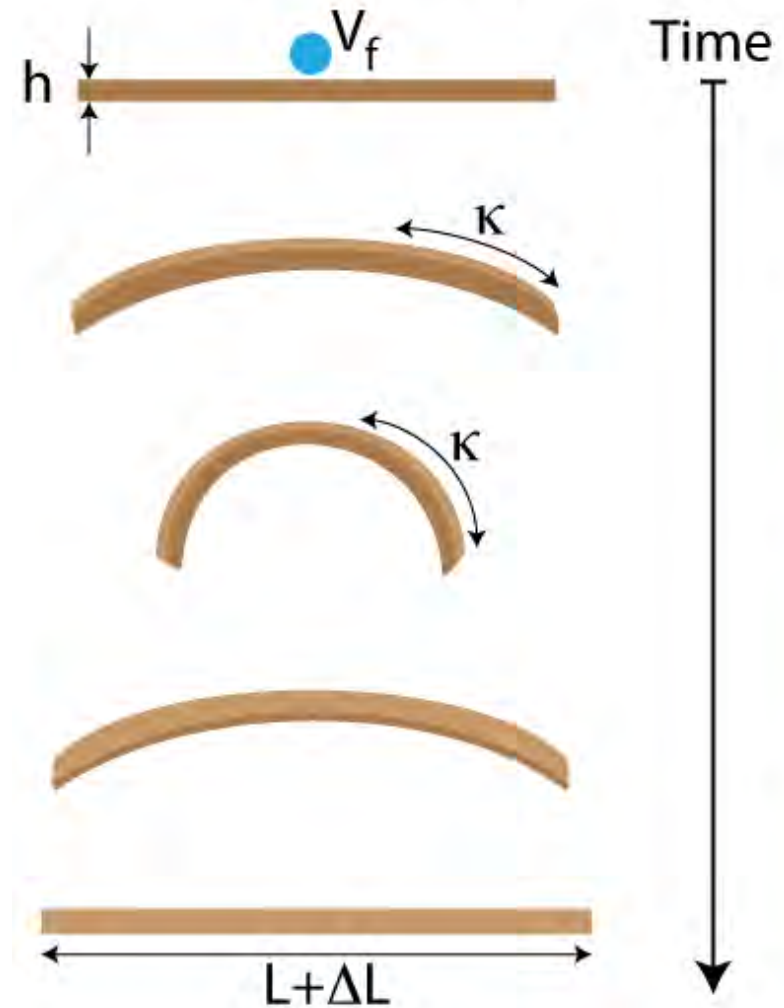
$$U_b = \frac{B}{2} \int_L \theta'(s)^2 ds \sim \bar{E} h^3$$

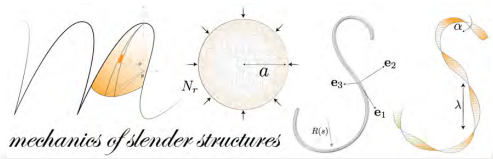
Swelling

$$U_s = \int_{V_f} \sigma \varepsilon_{eq} dV_f \sim E \varepsilon_{eq}^2 V_f$$

Length scale:

$$l_{es} \sim (\varepsilon_{eq}^2 V_f)^{1/3}$$

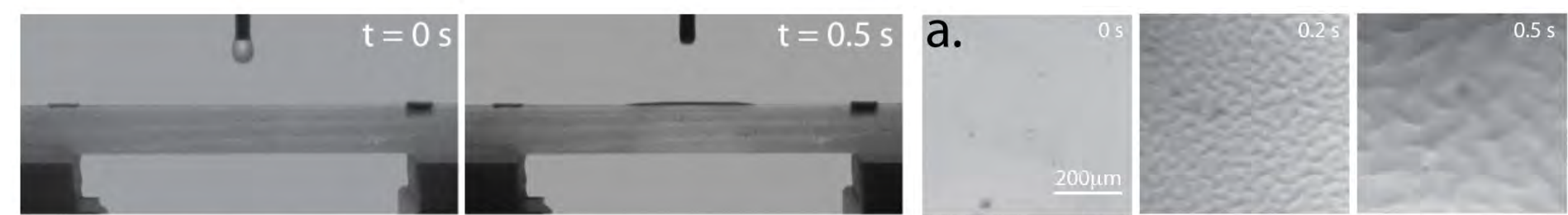




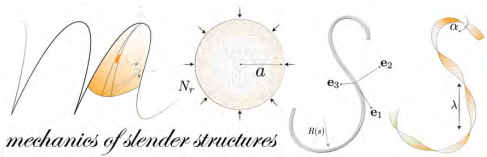
Thin structures bend...



Thick structures stay flat, while their surface creases...



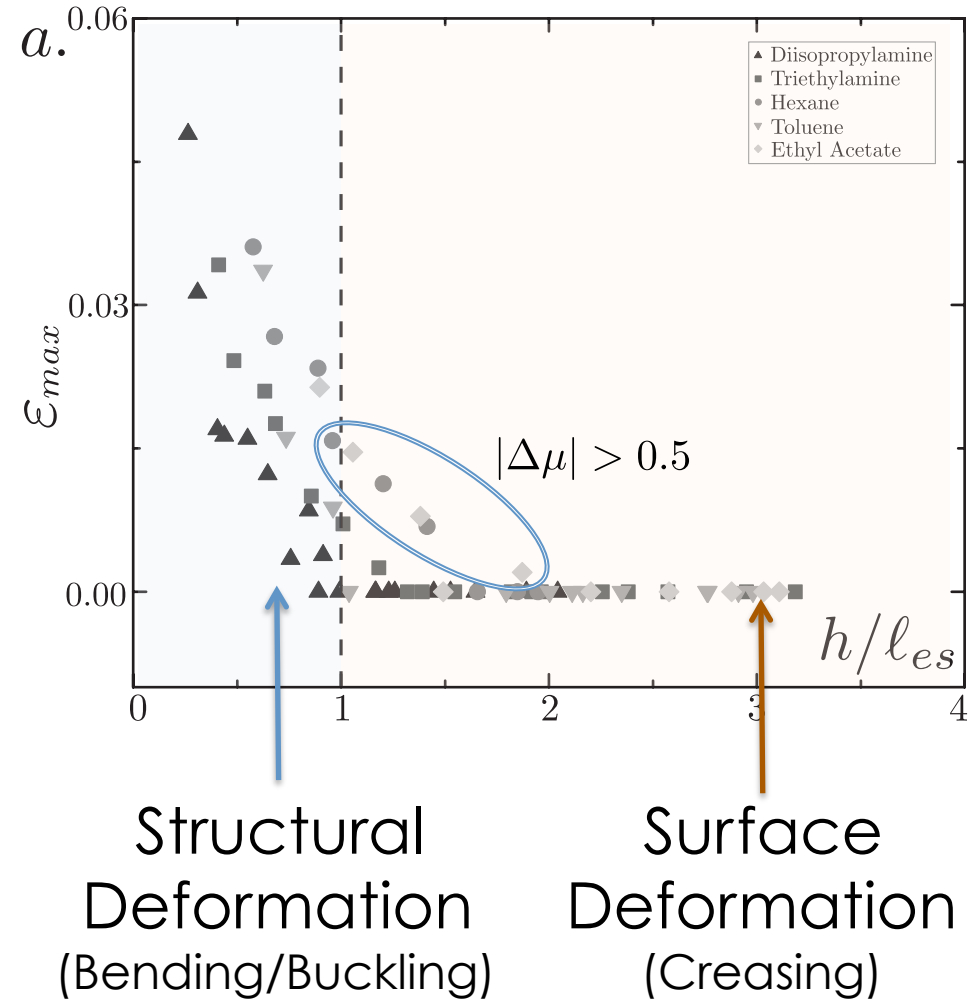
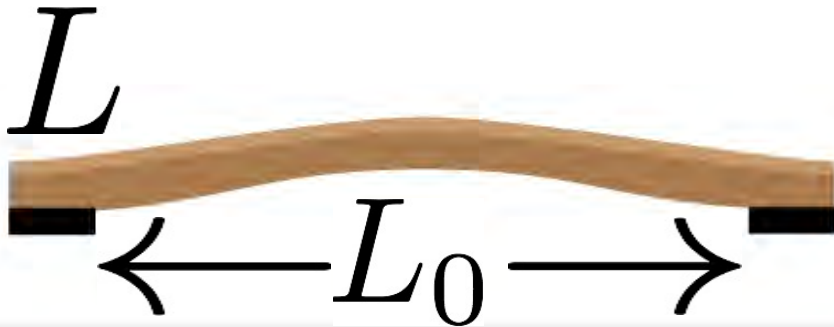
“Elastoswelling” length $l_{es} \sim (\epsilon_{eq}^2 V_f)^{1/3}$

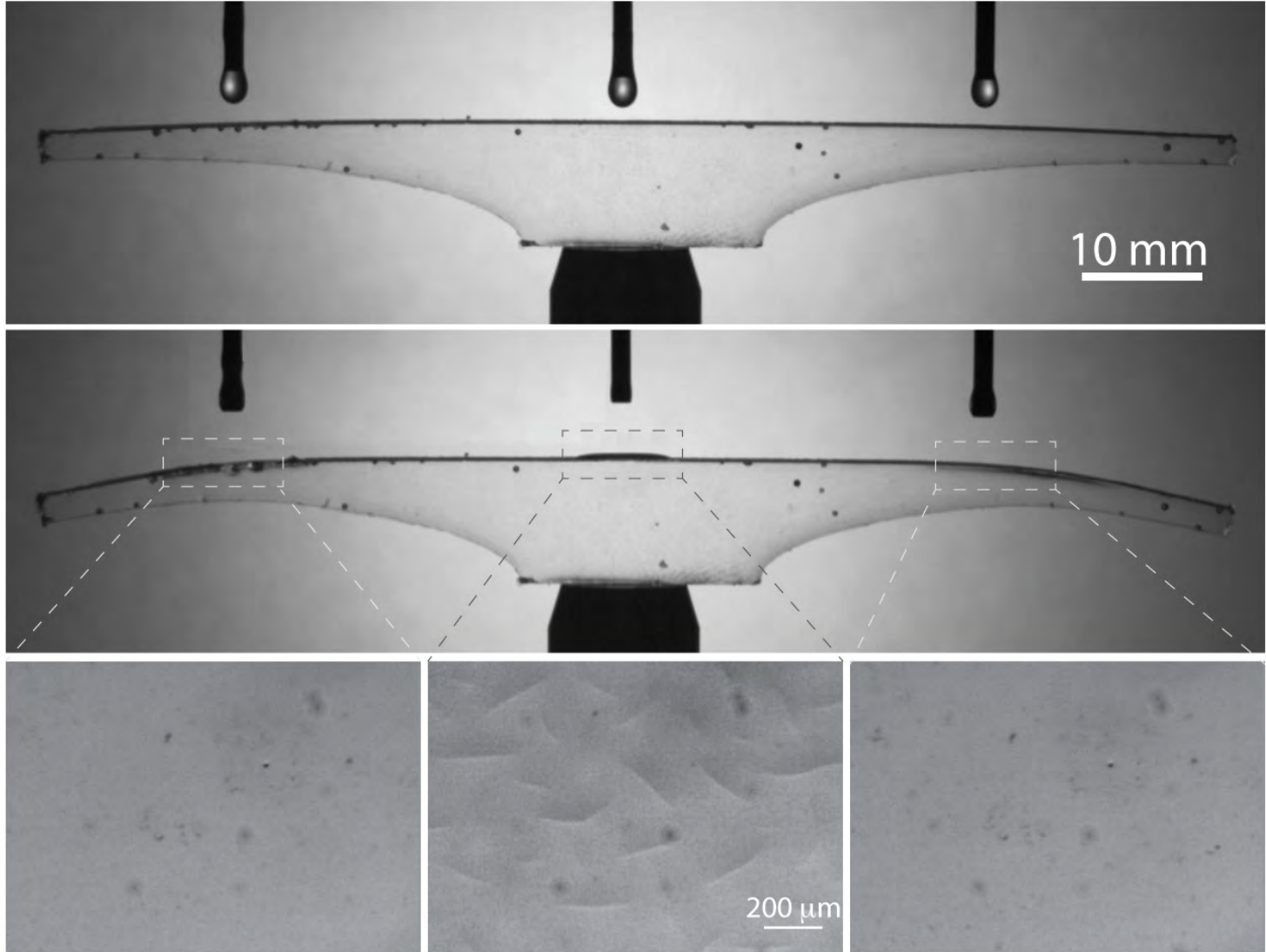
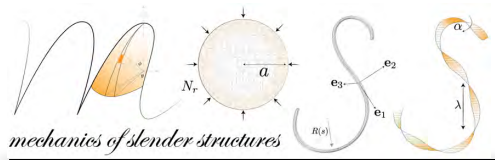


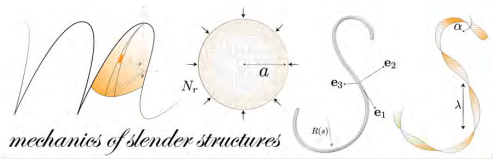
Deformation Transition

Material	δ_s ($cal^{1/2}cm^{-3/2}$)	μ (D)	ϵ_{eq}
PDMS	7.3	0.6-0.9	–
Diisopropylamine	7.3	1.2	1.13
Triethylamine	7.5	0.7	0.58
Hexanes	7.3	0.0	0.35
Toluene	8.9	0.4	0.31
Ethyl acetate	9.0	1.8	0.18

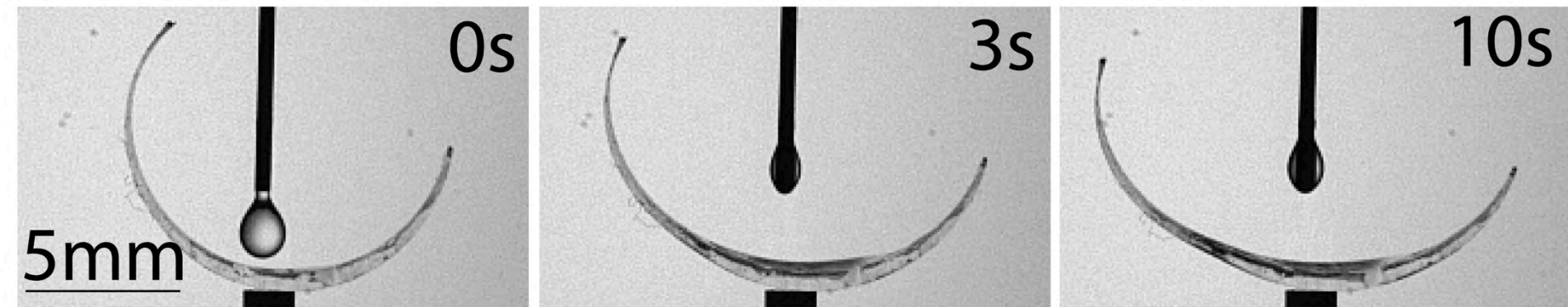
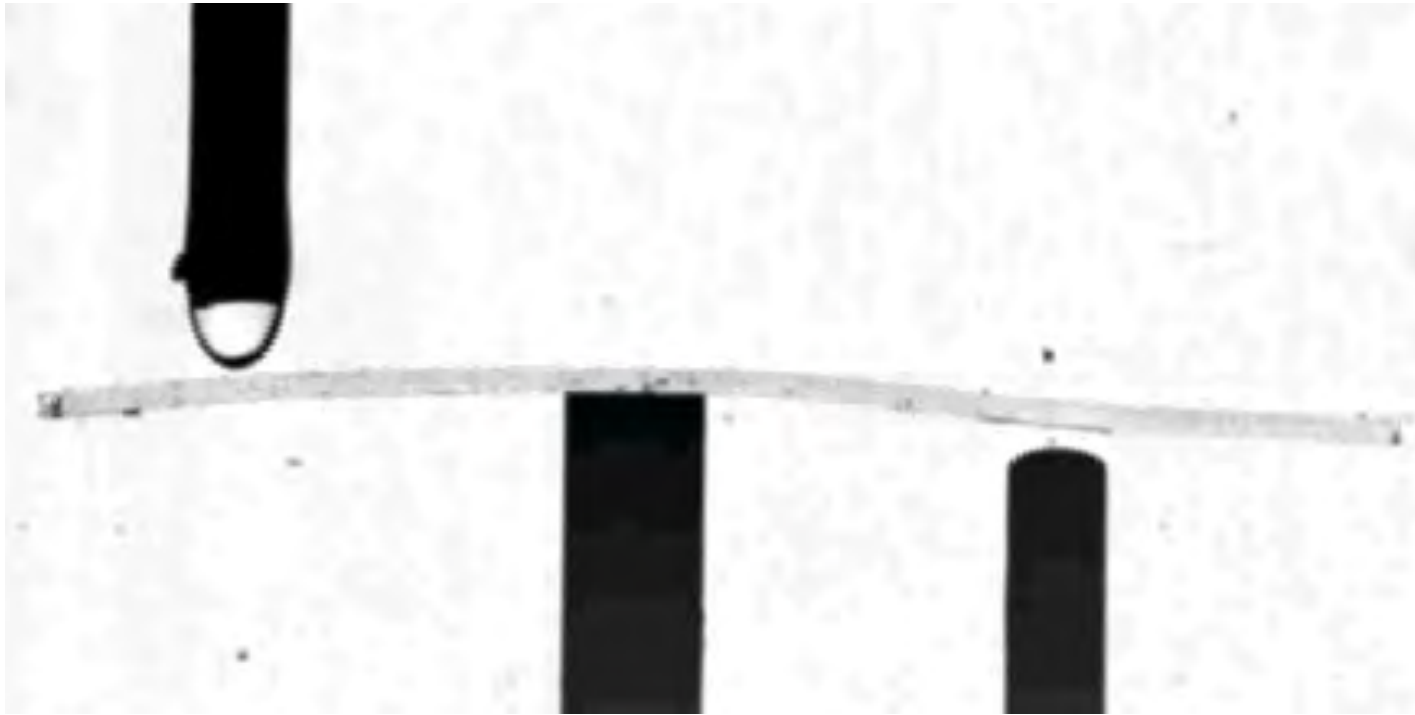
$$l_{es} \sim (\epsilon_{eq}^2 V_f)^{1/3}$$



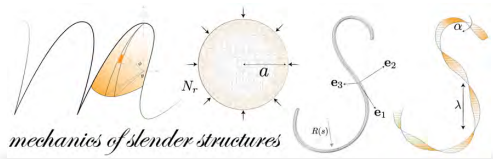




Controlling Shape

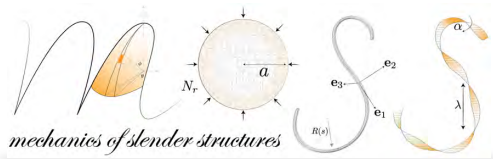


D.P. Holmes, M. Roché, T. Sinha, and H.A. Stone. "Bending and Twisting of Soft Materials by Non-homogenous Swelling" *Soft Matter*, **7**, 5188, 2011.

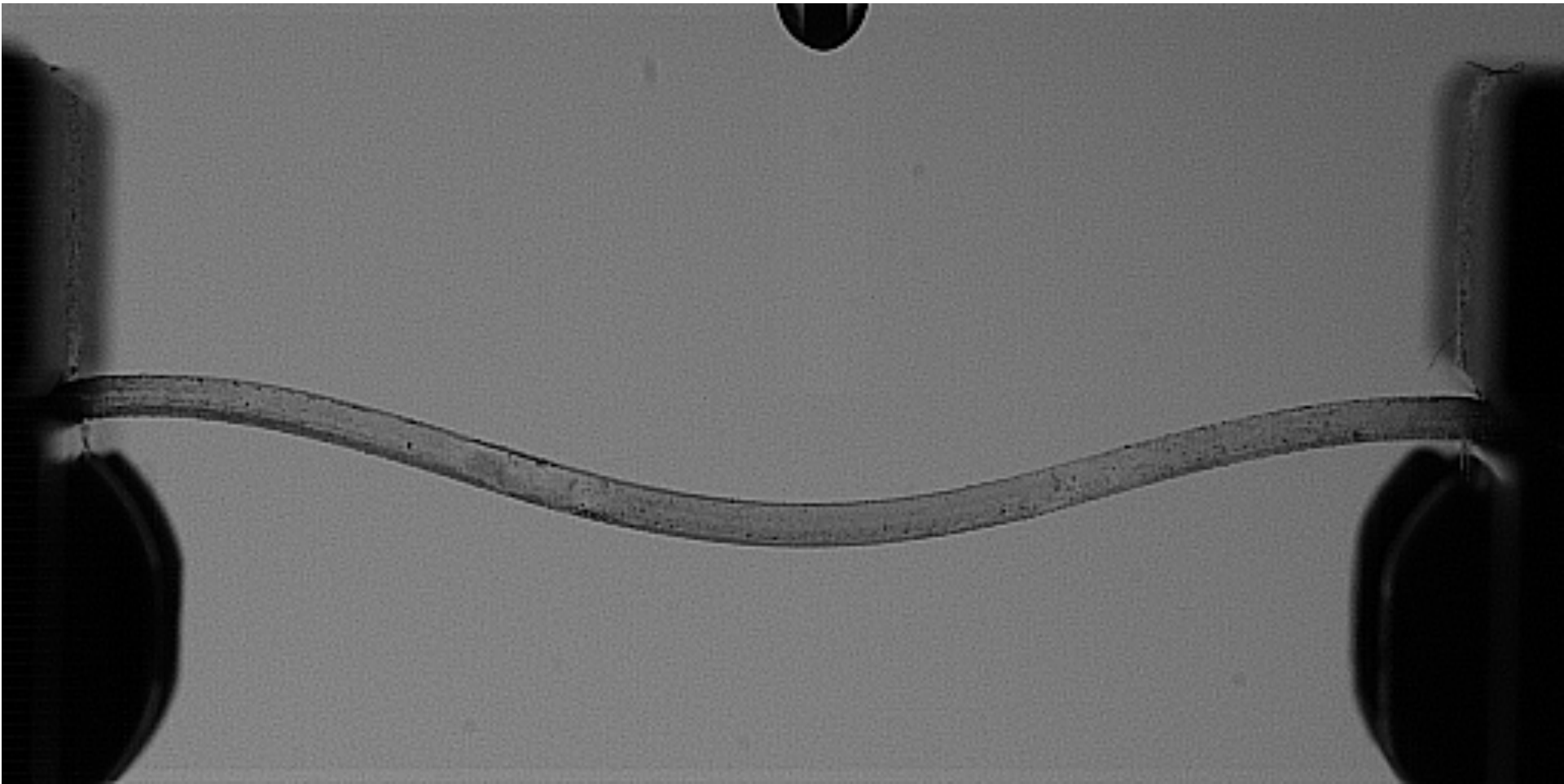


Microfluidic Swelling

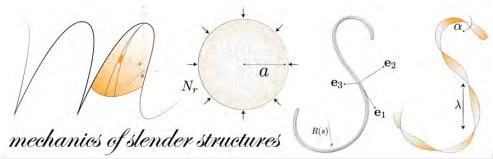




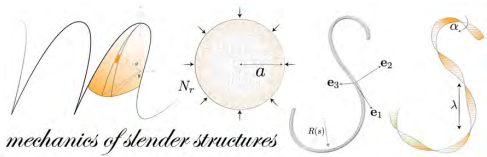
Controlling Shape



———— 10 mm



What about wetting?



Fluid Behavior

Viscous Forces



Coiling honey

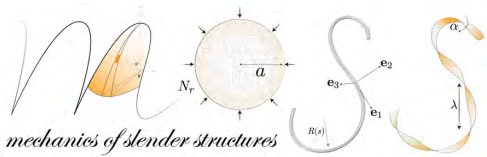
Interfacial Forces



Wetting of water on a textured surface

Capillary Number: viscous/interfacial

- Honey: <http://www.honeyassociation.com/webimages/honey-dipper.jpg>
- Droplet: <http://www.rycobel.be/en/technical-info/articles/1337/measuring-dynamic-absorption-and-wetting>

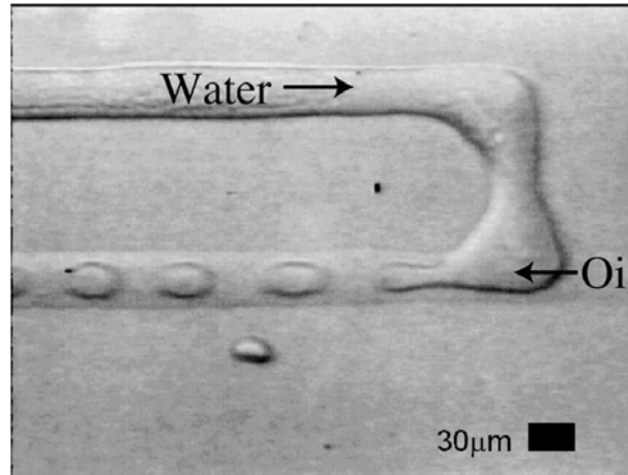


Fluid Behavior

Viscous Forces



Capillary Number: viscous/interfacial



Monodisperse droplet generation

- Droplet emulsions in immiscible fluids
- Injection of water into stream of oil

Interfacial tension prevents the fluids from flowing alongside each other.

Interfacial Forces



Surface tension acts to reduce the interfacial area.

$$\sigma_c \sim \gamma/R$$

Viscous stresses act to extend and drag the interface downstream.

$$\sigma_v \sim \mu U_0/h$$

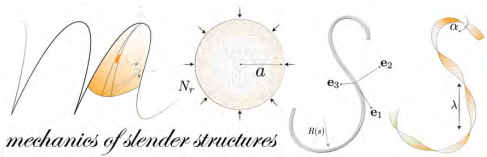
Characteristic droplet size:

$$R \sim \frac{\gamma}{\mu U_0} h = \frac{h}{\mathcal{C}}$$

Capillary number:

$$\mathcal{C} \equiv \frac{\mu U_0}{\gamma}$$

- Squires, Todd M., and Stephen R. Quake. "Microfluidics: Fluid physics at the nanoliter scale." *Reviews of modern physics* 77.3 (2005): 977.
- Thorsen, Todd, et al. "Dynamic pattern formation in a vesicle-generating microfluidic device." *Physical review letters* 86.18 (2001): 4163.
- Honey: <http://www.honeyassociation.com/webimages/honey-dipper.jpg>
- Droplet: <http://www.rycobel.be/en/technical-info/articles/1337/measuring-dynamic-absorption-and-wetting>



Fluid Behavior

Viscous Forces



Capillary Number: viscous/interfacial

Large **surface-to-volume** ratios in microfluidic devices

- Makes surface effects increasingly important.
- Important when free fluid surfaces are present.

Surface tensions can exert significant stress

- Result in free surface deformations.
- Can drive fluid motion.

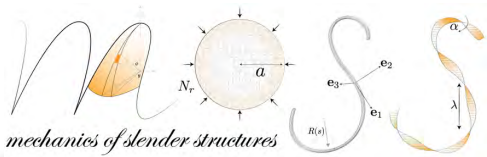
Interfacial Forces



Capillary forces tend to draw fluid into wetting microchannels

- Occurs when **solid-liquid** interfacial **energy** is **lower** than the **solid-gas** interfacial **energy**.

- Squires, Todd M., and Stephen R. Quake. "Microfluidics: Fluid physics at the nanoliter scale." *Reviews of modern physics* 77.3 (2005): 977.
- Honey: <http://www.honeyassociation.com/webimages/honey-dipper.jpg>
- Droplet: <http://www.rycobel.be/en/technical-info/articles/1337/measuring-dynamic-absorption-and-wetting>



Capillary Rise

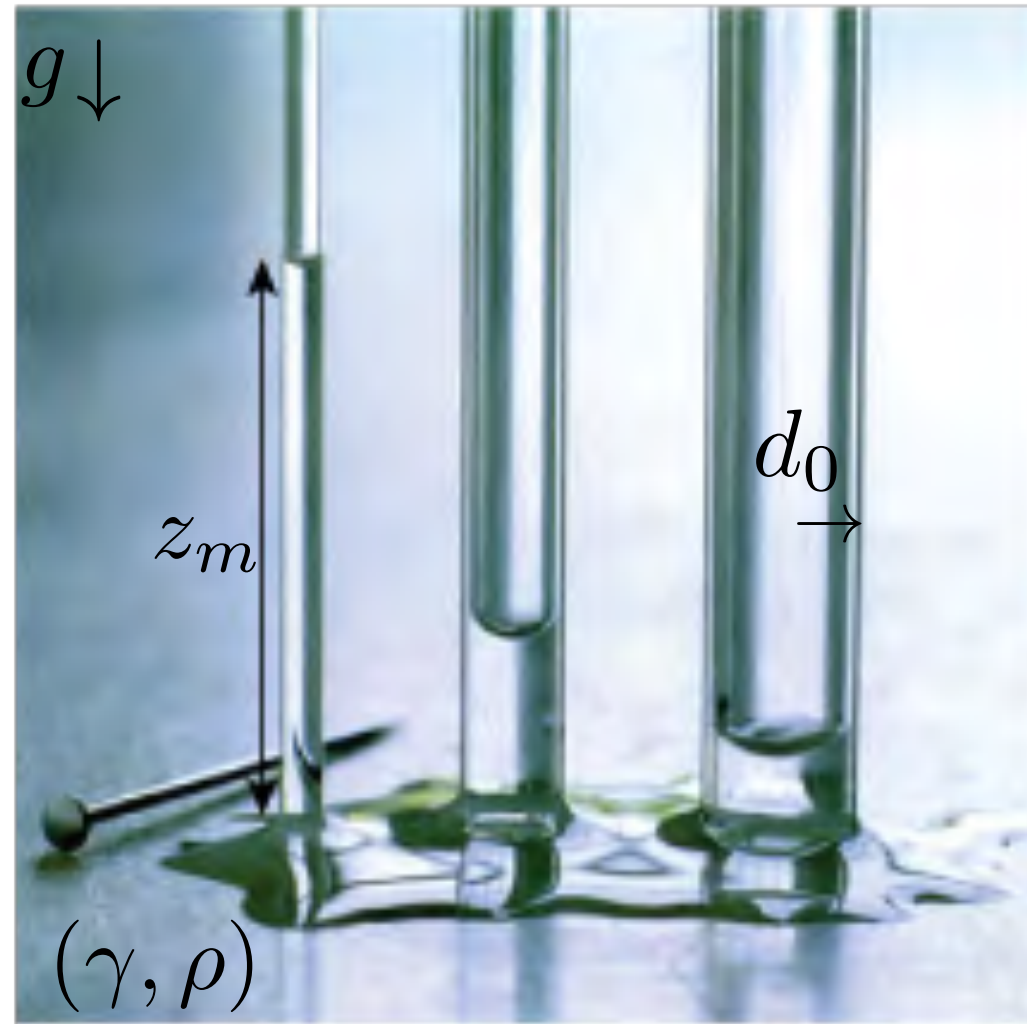
Classical Problem:

- Noted as early as 15th by Leonardo da Vinci.
- Attributed to circulation in plants in 17th century by Montanari.

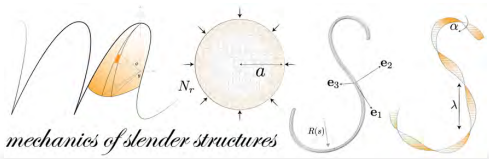
$$U = \underbrace{2\pi\gamma \cos \theta_e r z_m}_{\text{surface energy}} + \underbrace{\frac{1}{2}\rho g \pi r^2 z_m^2}_{\text{gravitational P.E.}}$$

$$\frac{dU}{dz_m} = 0$$

$$l_{cg} \sim \frac{\gamma \cos \theta_e}{\rho g d_0}$$



Balance: Surface Tension & Gravity



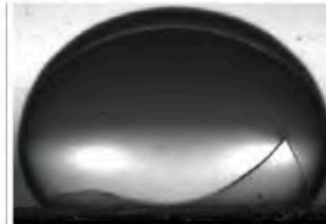
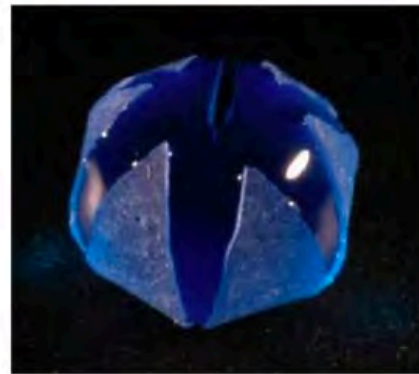
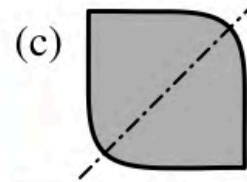
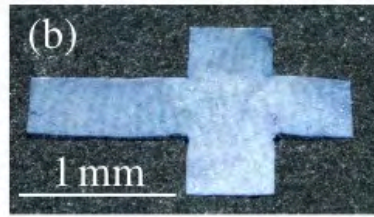
mechanics of slender structures



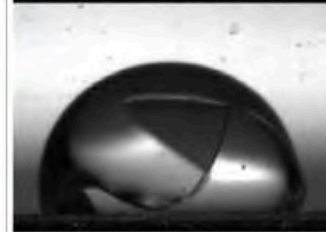
Py, Charlotte, Paul Reverdy, Lionel Doppler, José Bico, Benoit Roman, and Charles N. Baroud. "Capillary origami: spontaneous wrapping of a droplet with an elastic sheet." *Physical Review Letters* 98, no. 15 (2007): 156103.



Elastocapillarity



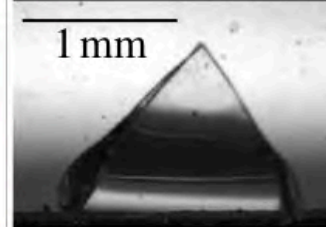
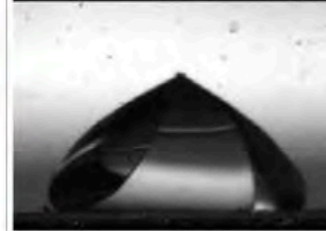
Fluid-structure interaction:



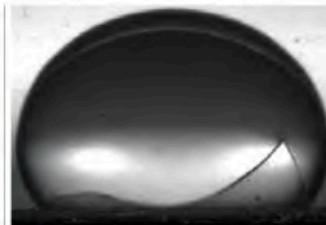
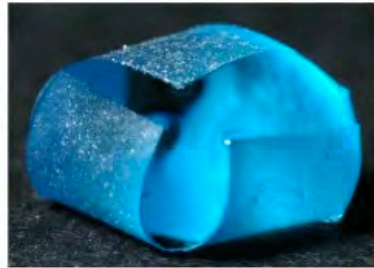
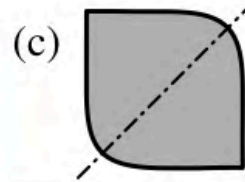
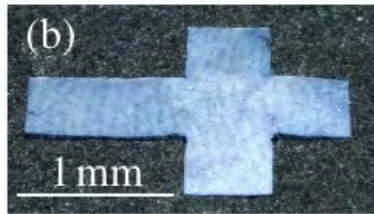
- Droplet bends and folds the sheet.

- Droplet is minimizing the amount of its surface in contact with air.

- Liquid-air surface area is minimized at the expense of bending the sheet.



Elastocapillarity



Fluid-structure interaction:

Elastic energy of a plate – bending:

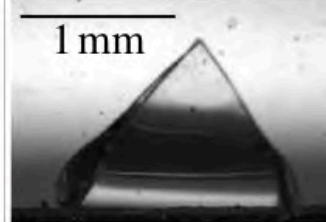
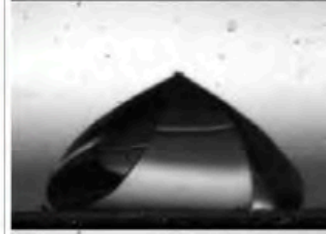
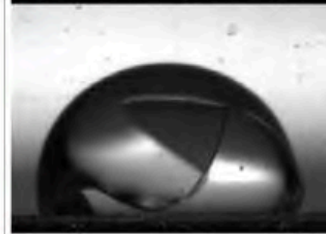
$$\mathcal{U}_e = \frac{1}{2} \iint_P dx dy \int_{-h/2}^{h/2} dz (\sigma_{\alpha\beta} \varepsilon_{\alpha\beta})$$

Relation between in-plane strain to out-of-plane bending:

$$\varepsilon_{\alpha\beta}(x) = z \frac{d^2 w}{dx^2} = \frac{z}{R}$$

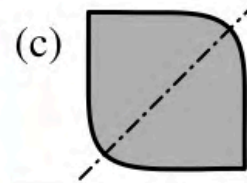
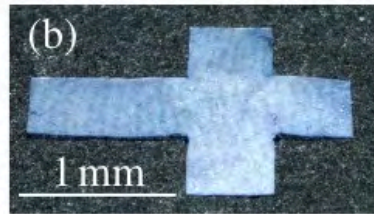
Bending energy:

$$\mathcal{U}_b = \frac{1}{2} \iint_P dx dy \frac{Eh^3}{12} \left(\frac{1}{R} \right)^2$$



$$\mathcal{U}_b \sim Eh^3$$

Elastocapillarity



Fluid-structure interaction:

Bending energy:

$$U_b \sim Eh^3$$

Surface energy:

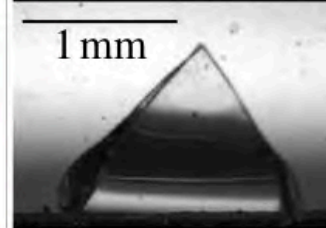
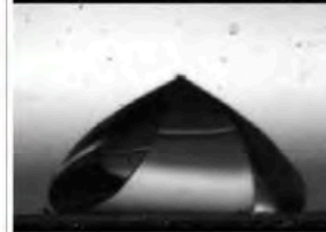
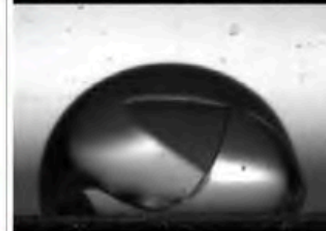
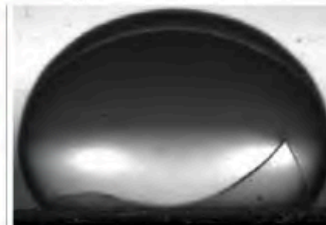
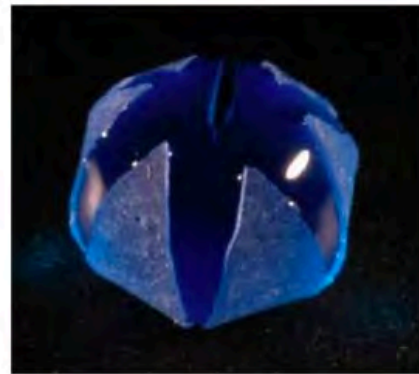
$$U_\gamma \sim \gamma L^2$$

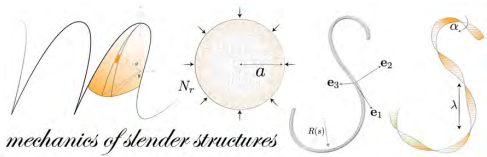
Elastocapillary length:

$$\ell_{ec} \sim \sqrt{\frac{Eh^3}{\gamma}} \sim \sqrt{\frac{B}{\gamma}}$$

Elastocapillary bending of sheet:

$$7\ell_{ec} \leq L \leq 12\ell_{ec}$$





Elastocapillarity

Capillary rise between flexible fibers.

$$\underbrace{B \frac{\partial^4 d}{\partial x^4}}_{\text{Beam bending}} = \underbrace{\gamma \kappa_f}_{\text{Laplace pressure}}$$

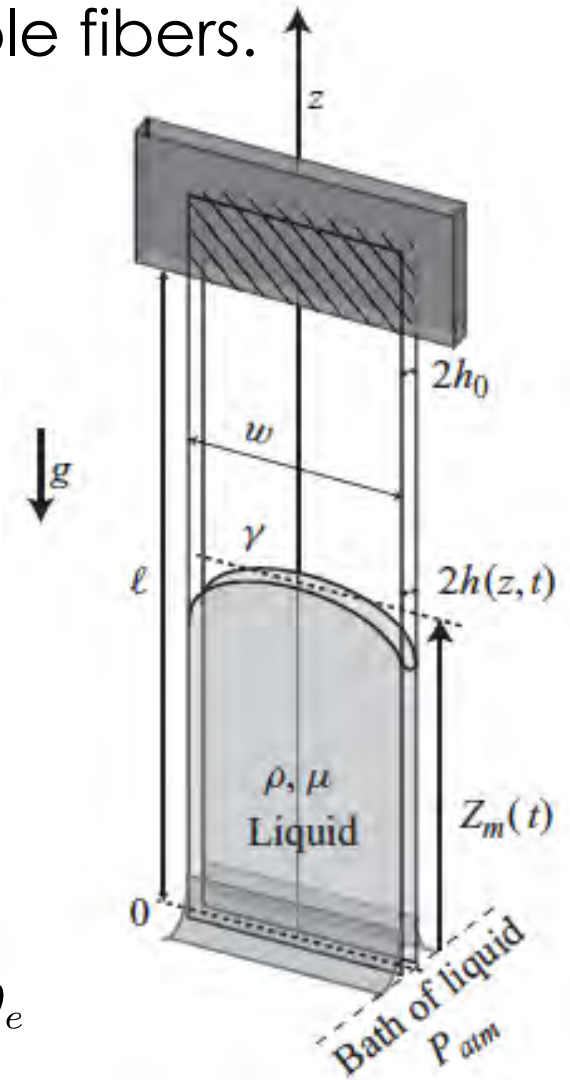
Curvature of meniscus

$$\kappa_f = \frac{1}{d(z_m)} \cos \theta_e + \frac{2}{b} \cos \theta_e$$

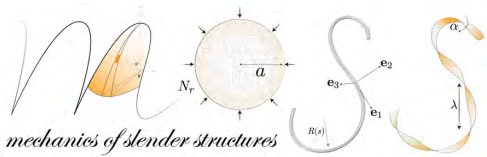
Assume gap is much smaller than width: $d_0 \ll b$

Initial gap: $d(z_m, t) = d_0$

Approximation of meniscus curvature: $\kappa_f \approx \frac{1}{d_0} \cos \theta_e$



J. Bico et al (2004)



Elastocapillarity

Capillary rise between flexible fibers.

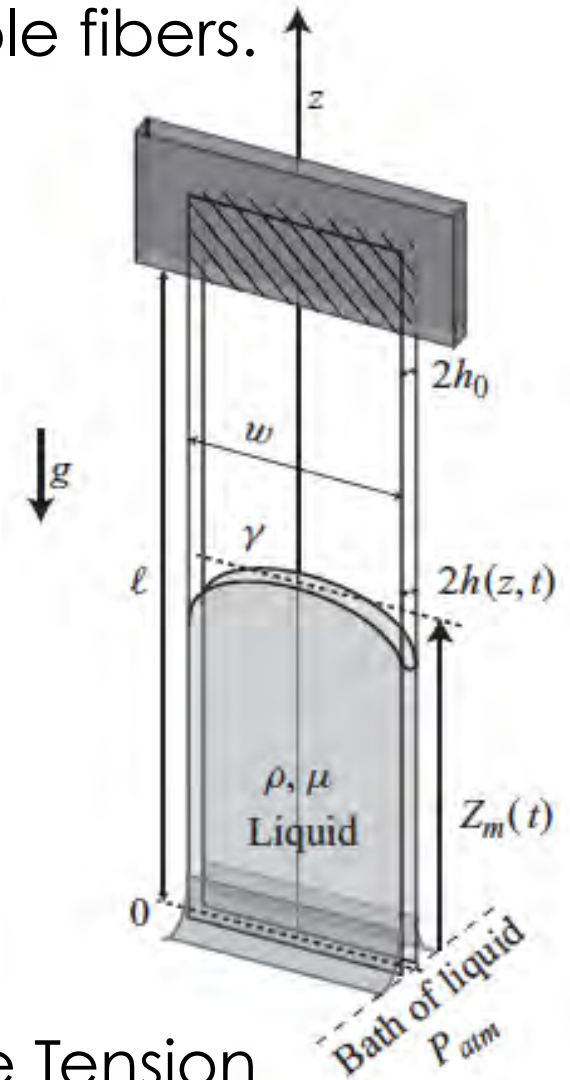
$$\underbrace{B \frac{\partial^4 d}{\partial x^4}}_{\text{Beam bending}} = \underbrace{\gamma \kappa_f}_{\text{Laplace pressure}}$$

Approximation: $\kappa_f \approx \frac{1}{d_0} \cos \theta_e$

Scaling: $B \frac{d_0}{\ell^4} \sim \frac{\gamma}{d_0} \cos \theta_e$

$$\ell_{ec} \equiv \left(\frac{B d_0^2}{\gamma \cos \theta_e} \right)^{1/4}$$

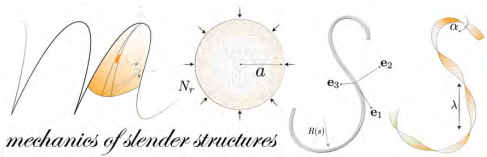
Balance: Bending & Surface Tension



J. Bico et al (2004)



H.-Y. Kim and L. Mahadevan, "Capillary rise between elastic sheets," *J. Fluid Mech.* **548**, 141-150, (2006).
 J.M. Aristoff, C. Duprat, and H.A. Stone, "Elastocapillary Imbibition," *Int. J Nonlinear Mech.* **48**, 648-656, (2011).
 C. Duprat, J.M. Aristoff, and H.A. Stone, "Dynamics of elastocapillary rise," *J. Fluid Mech.* **679**, 641-654, (2011).



Elastocapillarity

Capillary rise between flexible fibers.

Potential energies:

Elastic energy

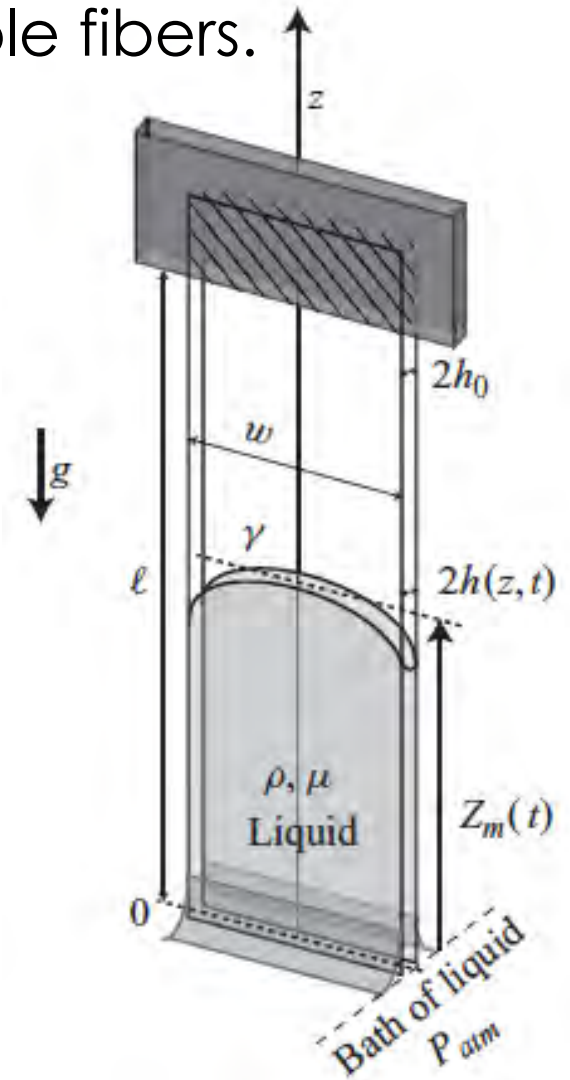
$$U_e \sim \frac{Bwh_0^2}{\ell^3}$$

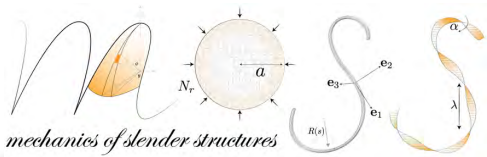
Gravitational potential energy

$$U_g \sim \rho g w h_0 \ell^2$$

Surface energy

$$U_c \sim w \ell \gamma$$





Elastocapillarity

Capillary rise between flexible fibers.

Characteristic length scales:

Elastocapillary length

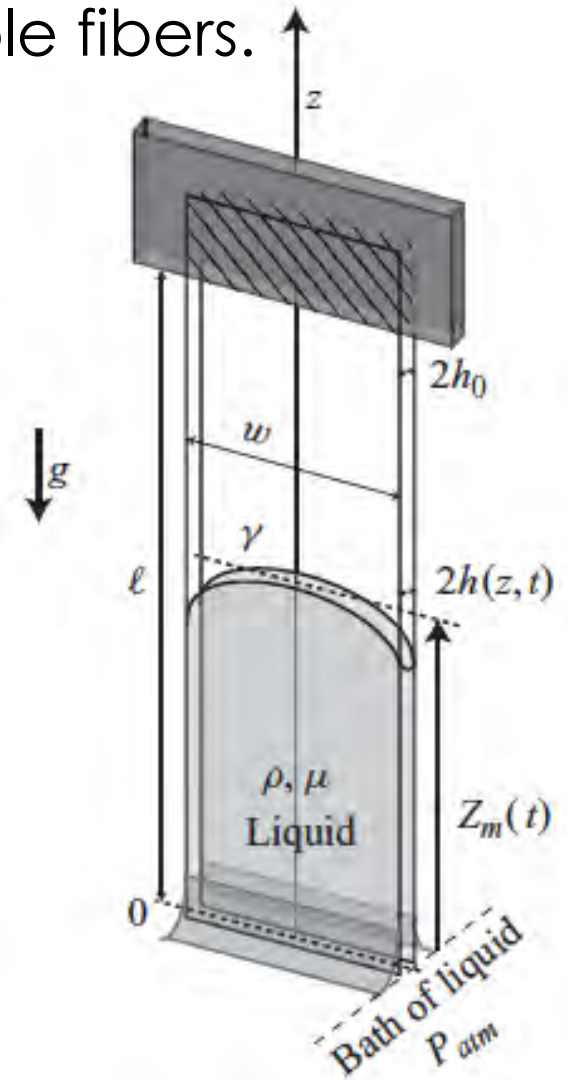
$$l_{ec} = \frac{U_e}{U_c} = \left(\frac{Bh_0^2}{\gamma} \right)^{1/4}$$

Capillary gravity length

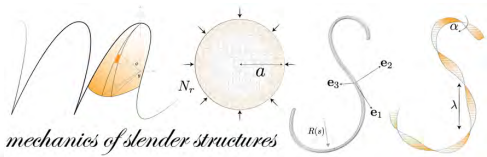
$$l_{cg} = \frac{U_c}{U_g} = \frac{\gamma}{\rho gh_0}$$

Elastogravity length

$$l_{eg} = \frac{U_e}{U_g} = \left(\frac{Bh_0}{\rho g} \right)^{1/5}$$



H.-Y. Kim and L. Mahadevan, "Capillary rise between elastic sheets," *J. Fluid Mech.* **548**, 141-150, (2006).
 J.M. Aristoff, C. Duprat, and H.A. Stone, "Elastocapillary Imbibition," *Int. J Nonlinear Mech.* **48**, 648-656, (2011).
 C. Duprat, J.M. Aristoff, and H.A. Stone, "Dynamics of elastocapillary rise," *J. Fluid Mech.* **679**, 641-654, (2011).



Elastocapillarity

Capillary rise between flexible fibers.

Dimensionless parameters

Bond number:

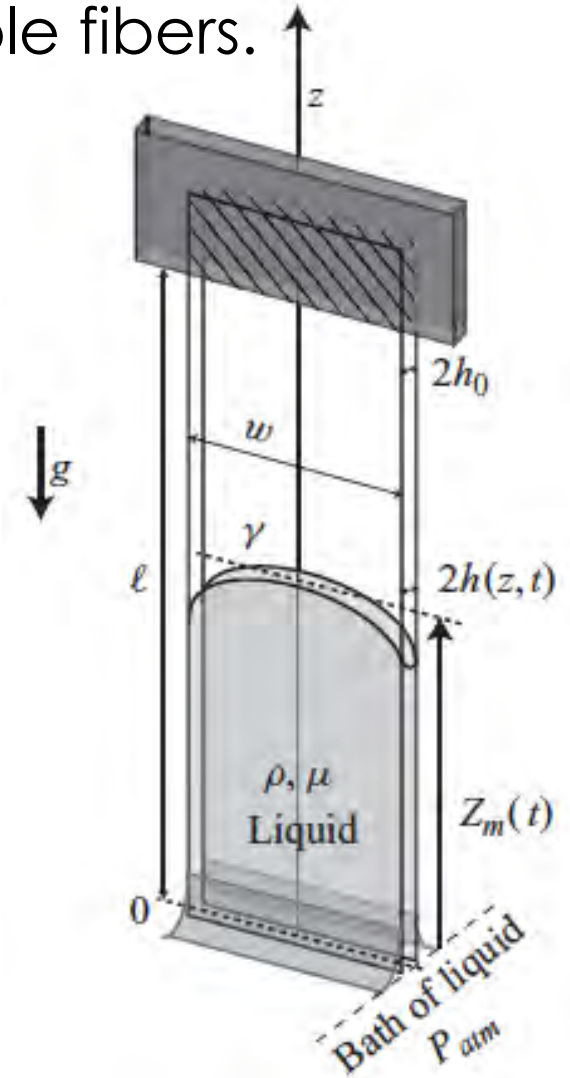
$$\mathcal{B} = \frac{l}{l_{cg}}$$

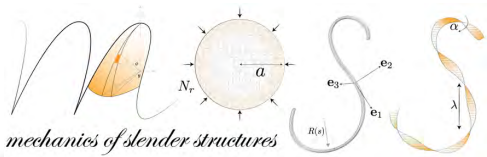
Elastocapillary number

$$\mathcal{E} = \left(\frac{l}{l_{ec}} \right)^4$$

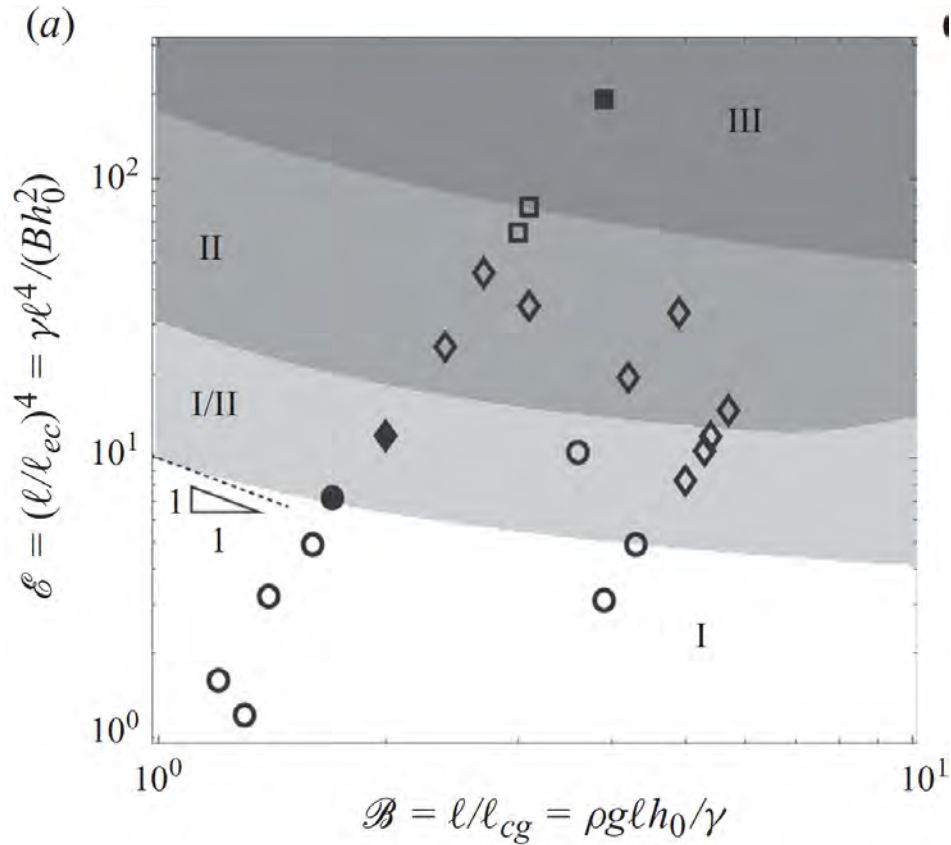
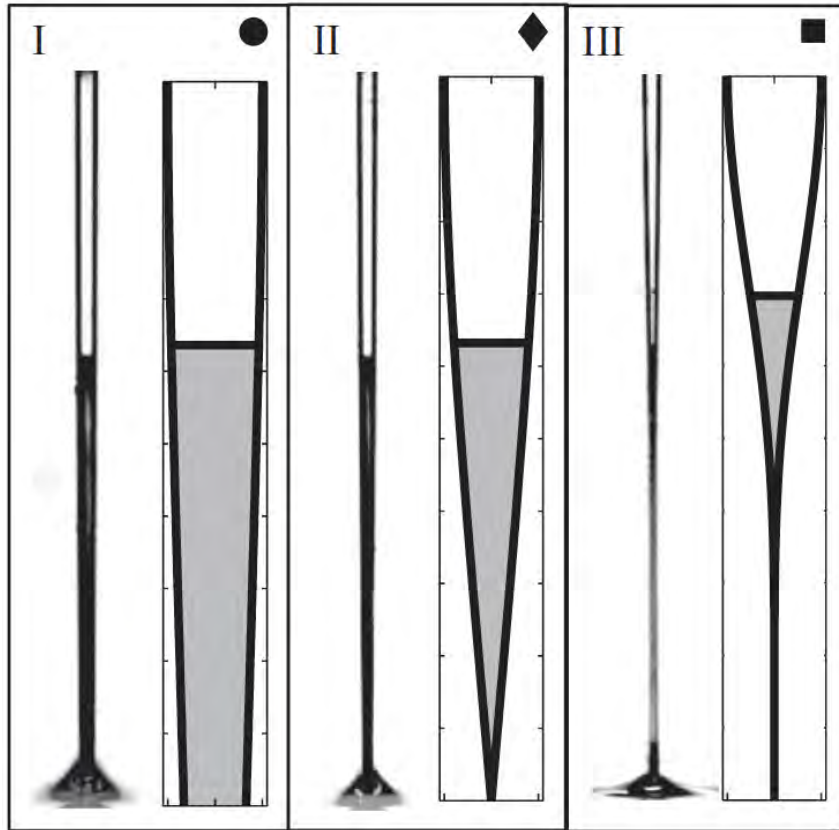
Elastogravity number

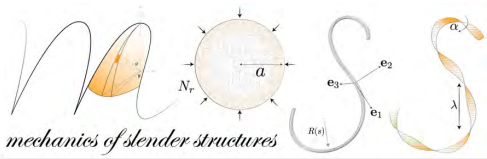
$$\mathcal{G} = \mathcal{B}\mathcal{E} = \left(\frac{l}{l_{eg}} \right)^5$$



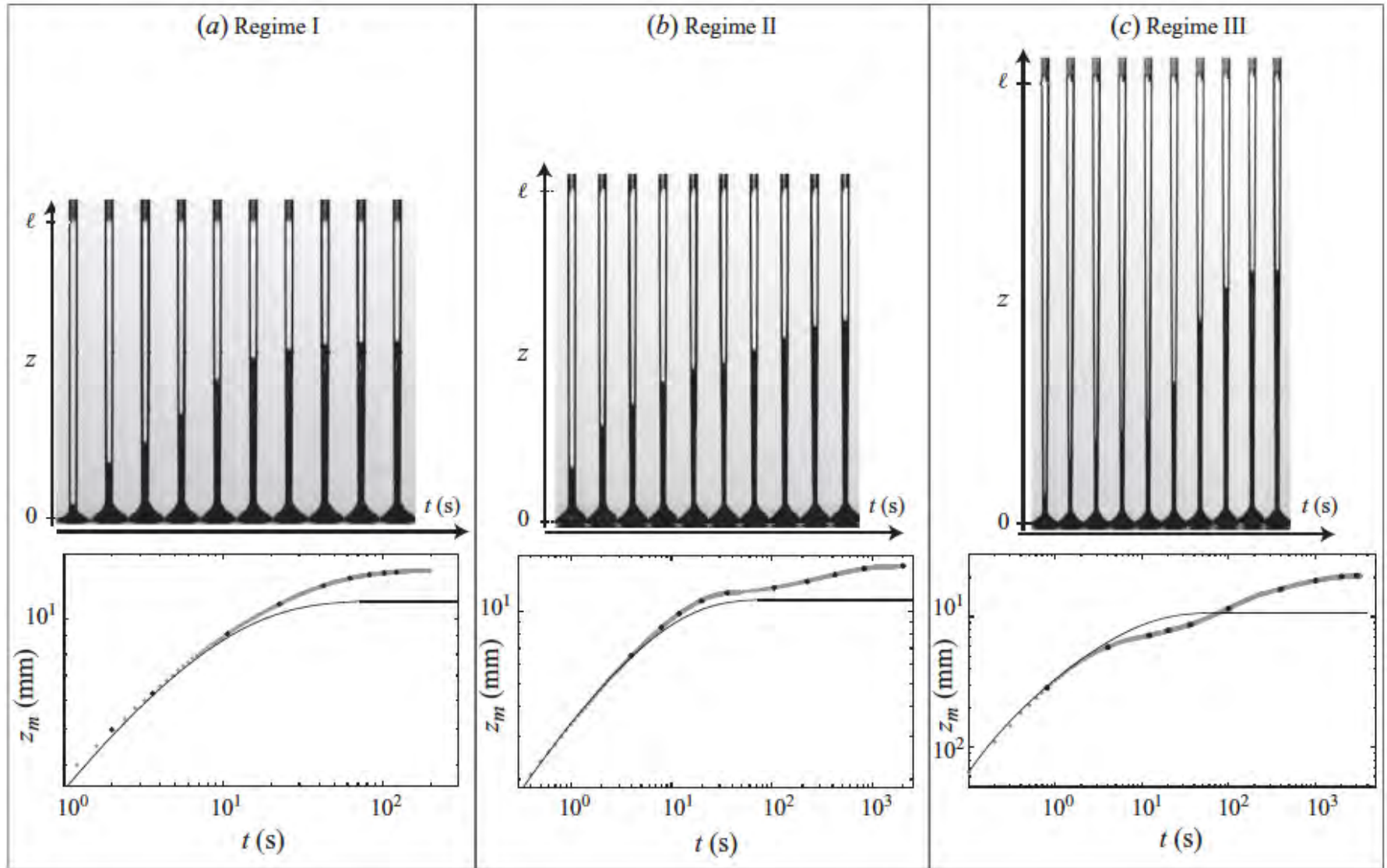


Elastocapillarity

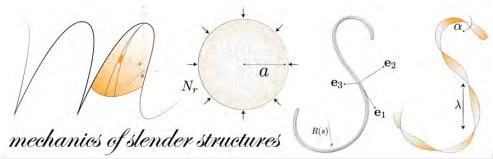




Elastocapillarity



H.-Y. Kim and L. Mahadevan, "Capillary rise between elastic sheets," *J. Fluid Mech.* **548**, 141-150, (2006).
 J.M. Aristoff, C. Duprat, and H.A. Stone, "Elastocapillary Imbibition," *Int. J Nonlinear Mech.* **48**, 648-656, (2011).
 C. Duprat, J.M. Aristoff, and H.A. Stone, "Dynamics of elastocapillary rise," *J. Fluid Mech.* **679**, 641-654, (2011).



Capillarity & Swelling



Solid: Polyvinylsiloxane

Fluid: Silicone Oil (5 cSt)

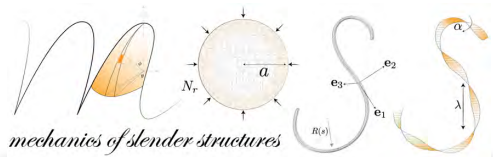
20x faster than real time

$E \approx 1 \text{ MPa}$ (PVS)

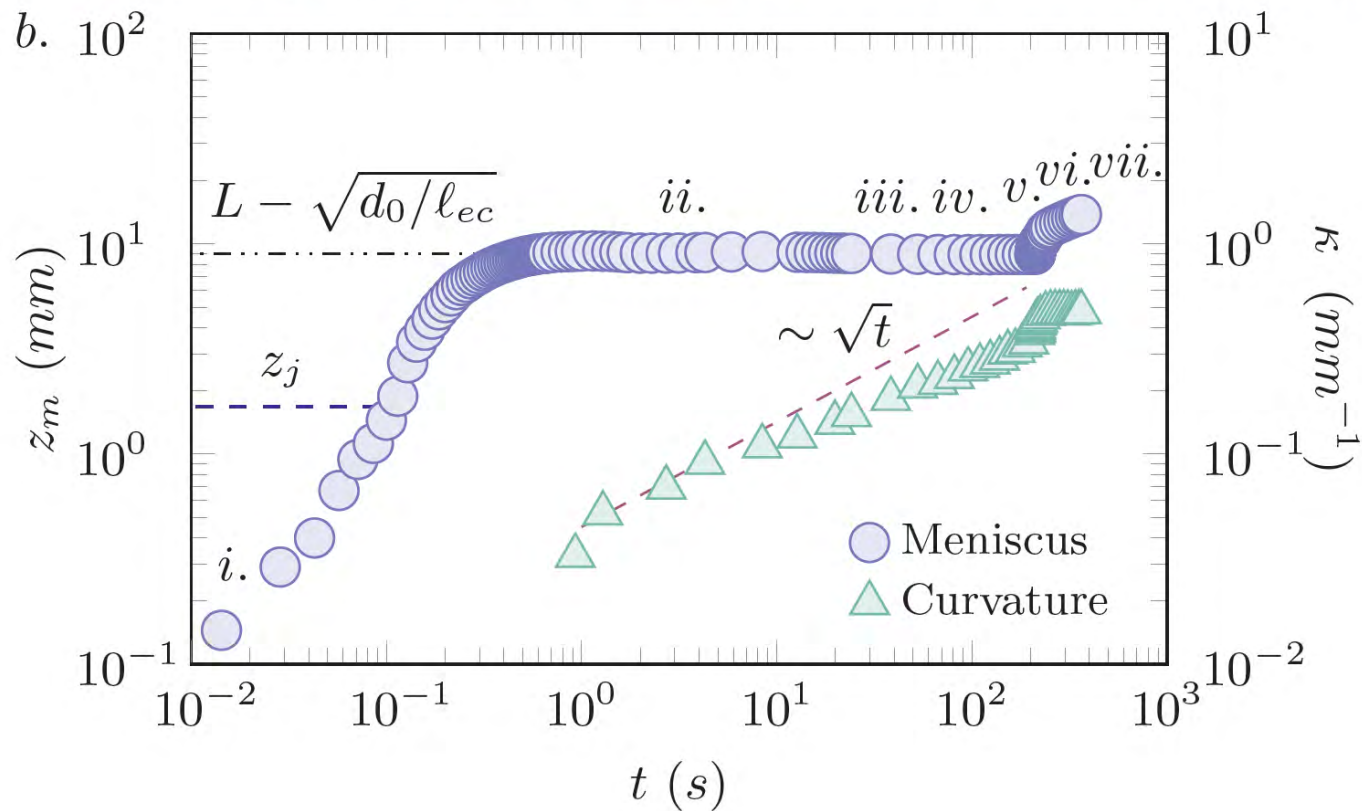
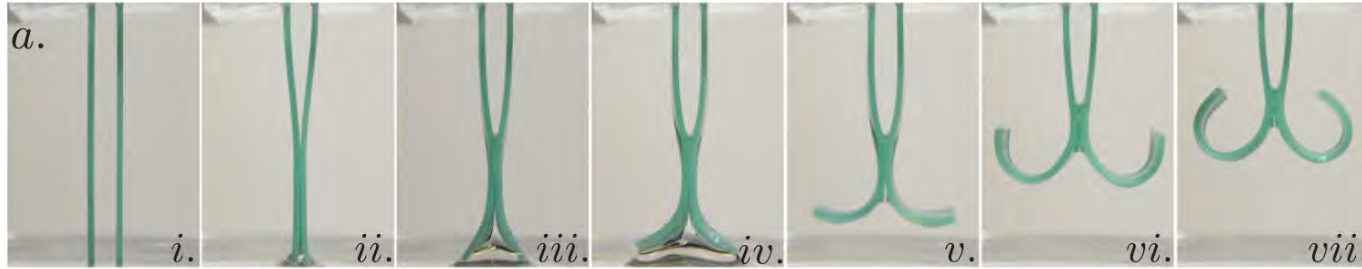
$L = 20 \text{ mm}$

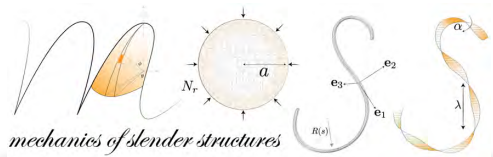
$d \approx 2 \text{ mm}$

$h \approx 0.5 \text{ mm}$

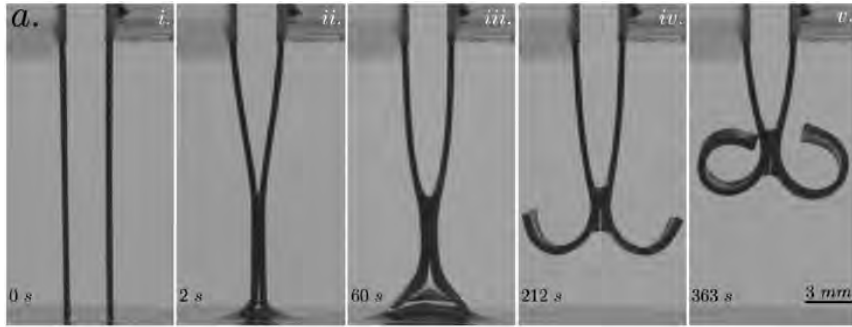


Capillarity & Swelling





Capillarity & Swelling



1. Elastocapillary rise between flexible fibers.

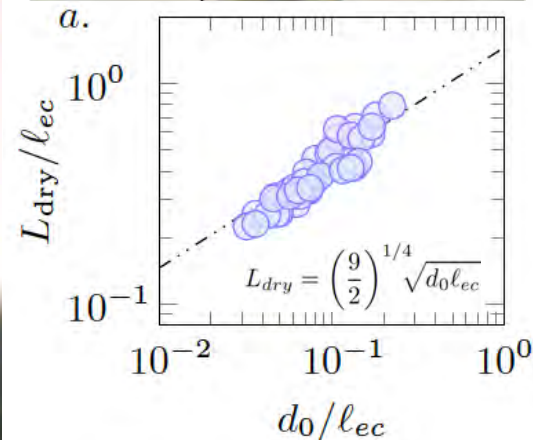
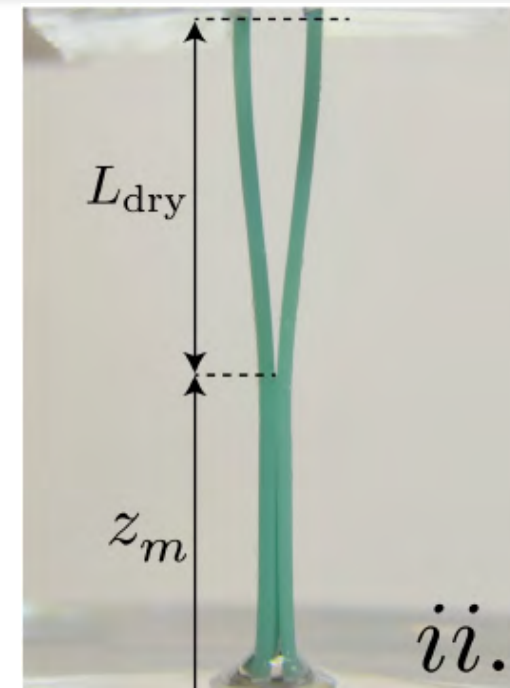
At short times, elastocapillary rise dominates the deformation.

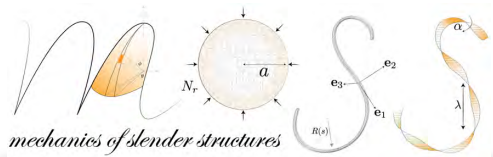
2. Swelling-induced bending.

Bending is constrained by surface tension, as the beam bends with a lower curvature than a free swelling beam.

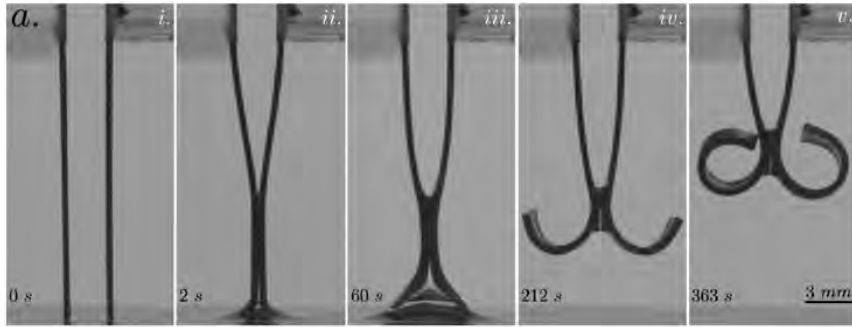
3. Bending dominates surface tension.

Separation occurs as the "natural" curvature of the beam exceeds the fluids ability to confine it.





Capillarity & Swelling



1. Elastocapillary rise between flexible fibers.

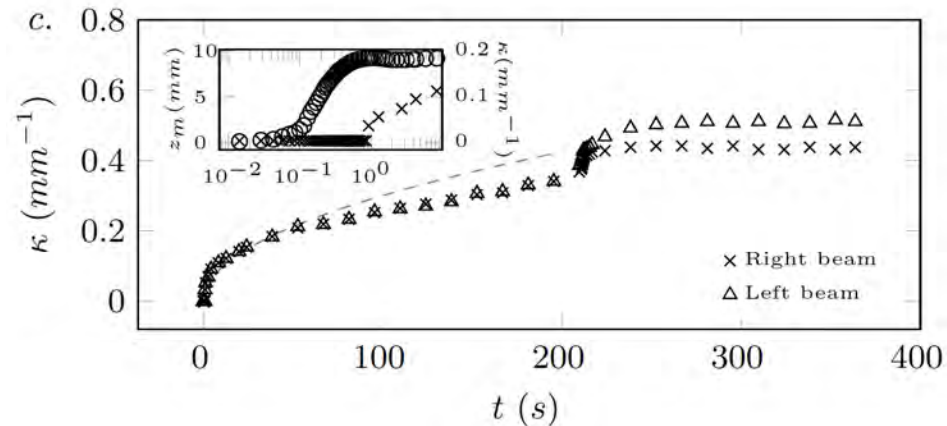
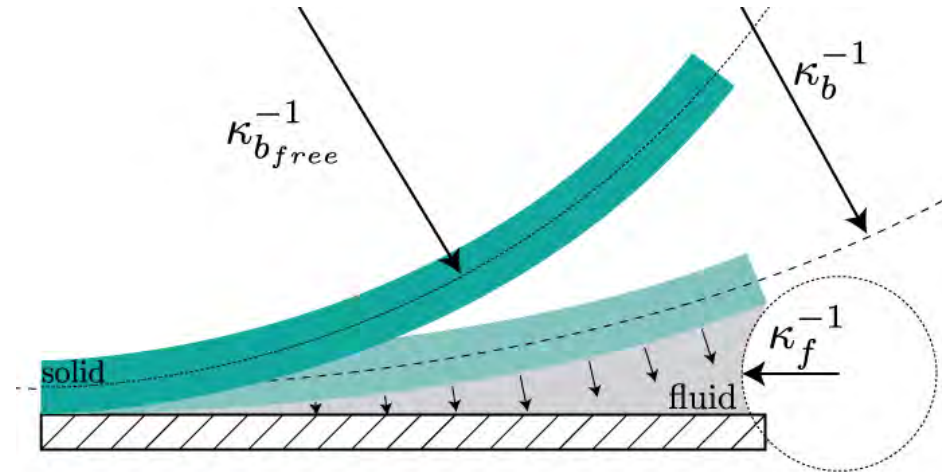
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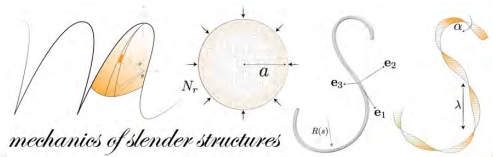
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Bending is constrained by surface tension, as the beam bends with a lower curvature than a free swelling beam.

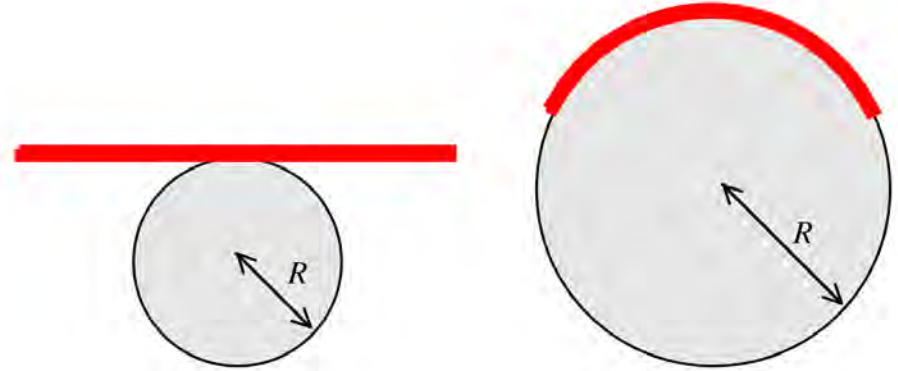
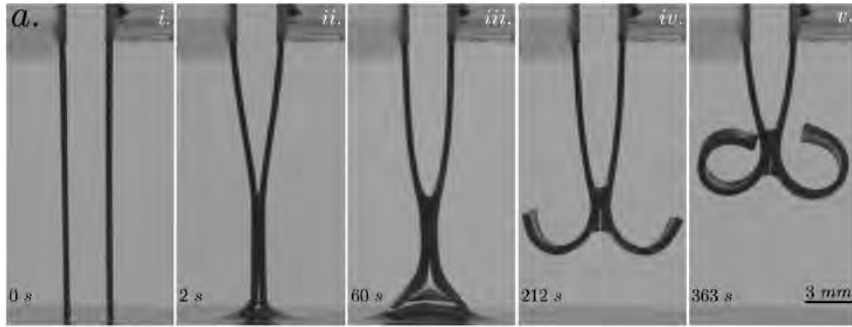
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Capillarity & Swelling



1. Elastocapillary rise between flexible fibers.

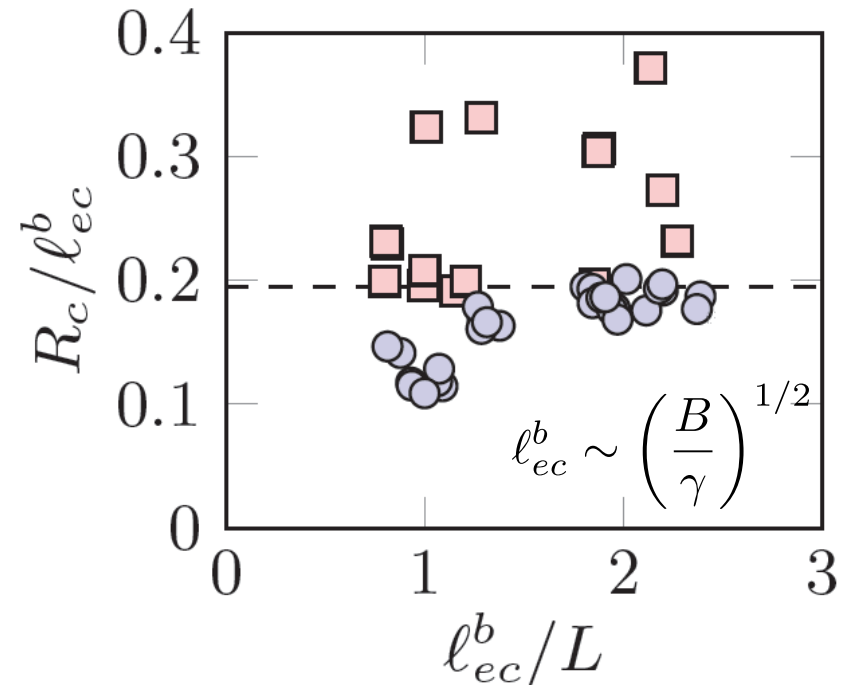
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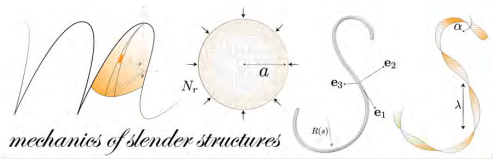
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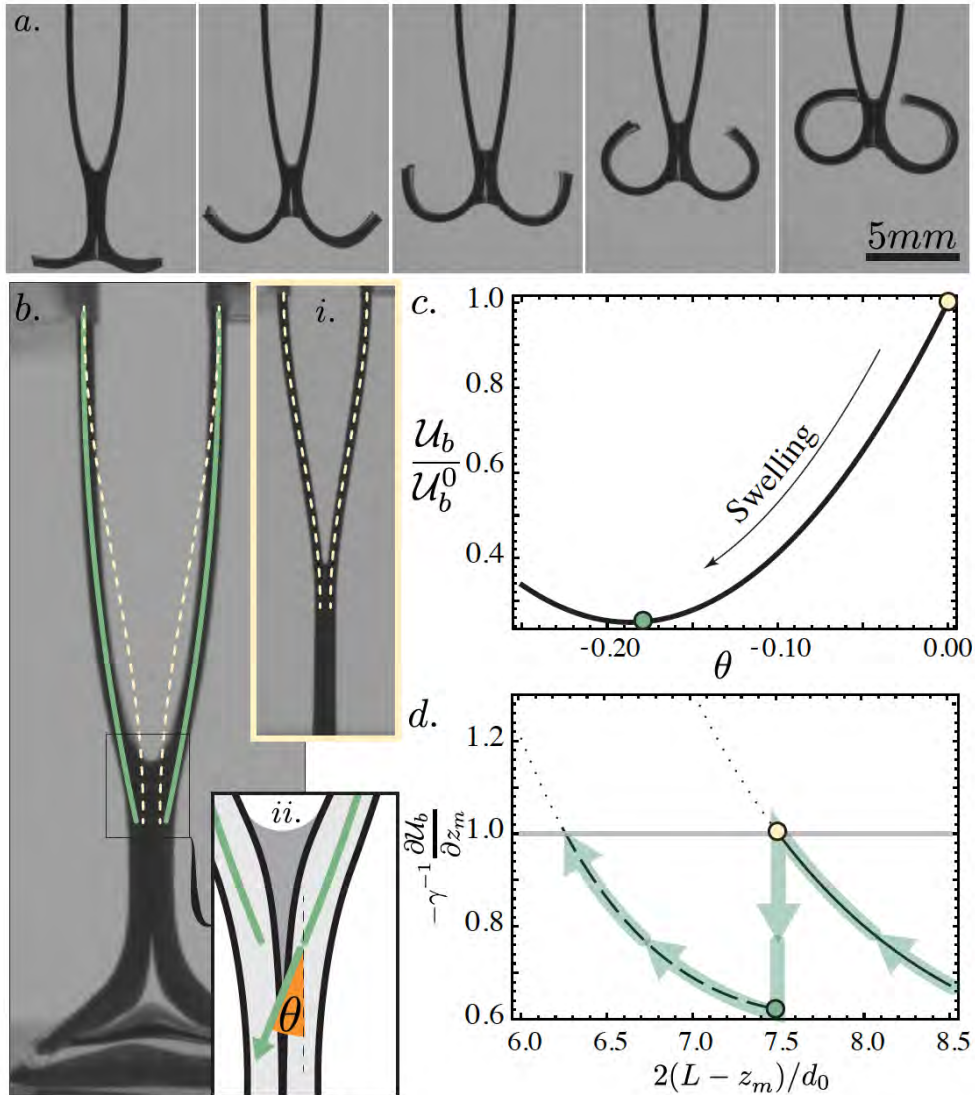
3. Bending dominates surface tension.

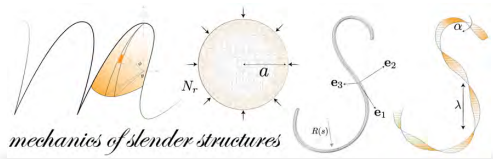
Separation occurs as the "natural" curvature of the beam exceeds the fluids ability to confine it.





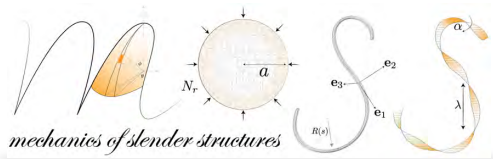
Swelling & Peeling



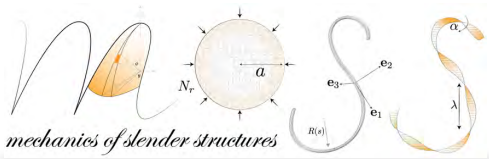


Baobab Flowering



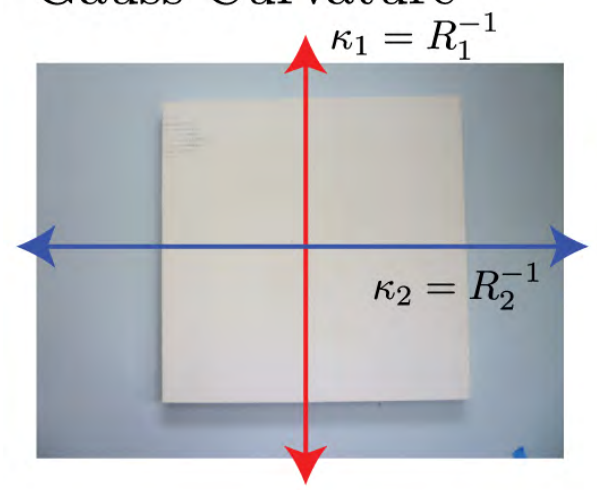


What about geometry?

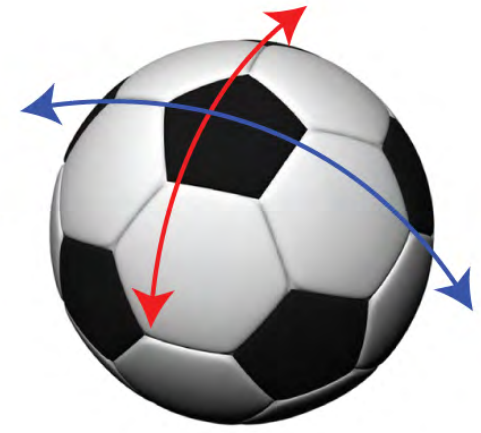


Mechanics of Thin

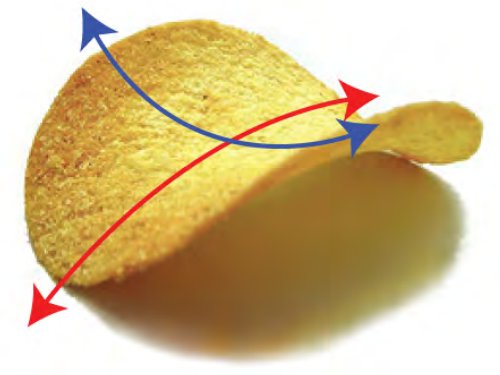
Gauss Curvature



$$K = \kappa_1 \kappa_2 = \diamond^4[w, w,] = 0$$

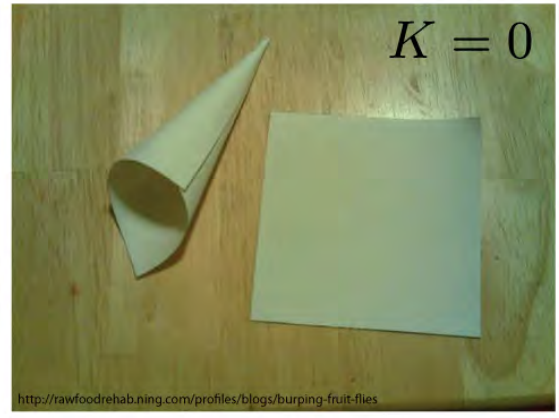
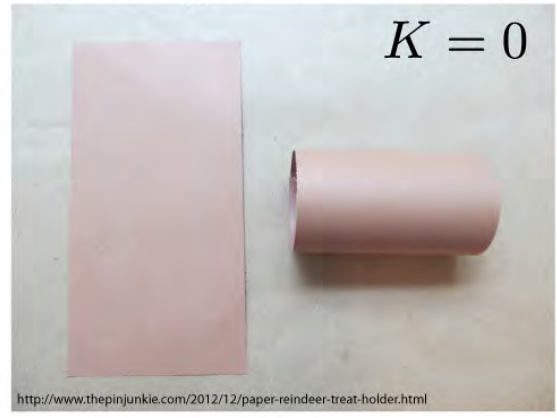


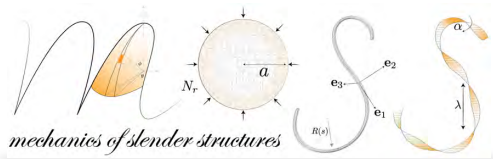
$$K = \kappa_1 \kappa_2 = \diamond^4[w, w,] > 0$$



$$K = \kappa_1 \kappa_2 = \diamond^4[w, w,] < 0$$

Developable

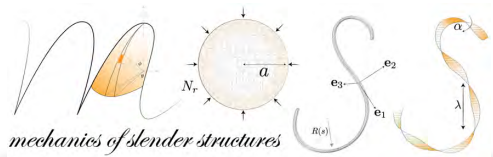




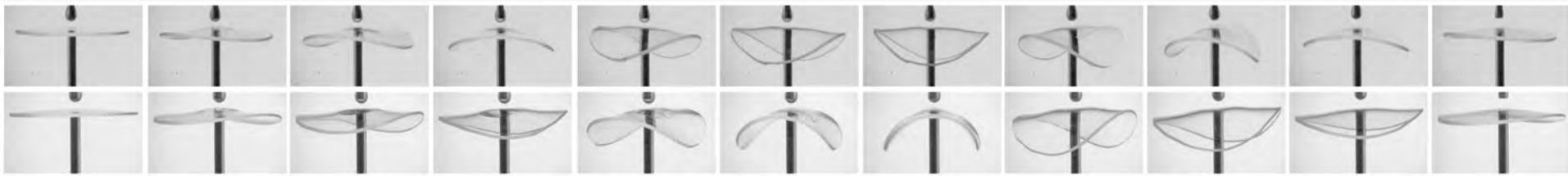
Dancing Disks



Axisymmetric Disks



Dynamic Plate Shape



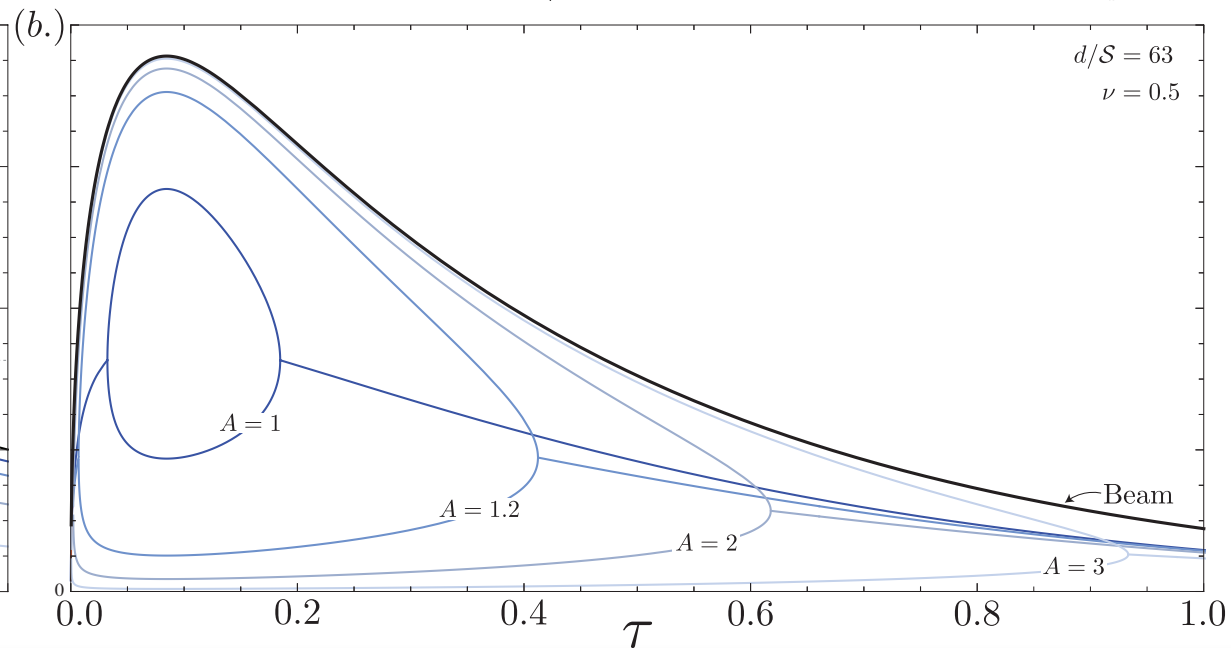
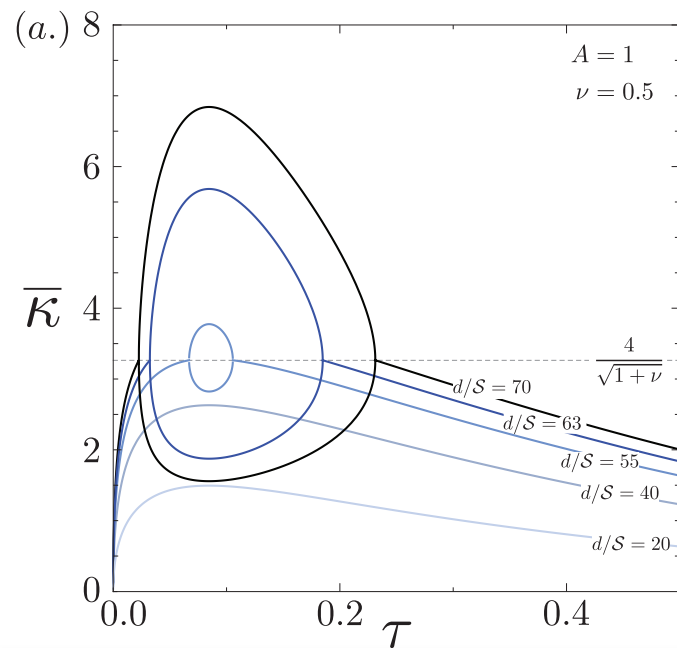
Diffusive dynamics:

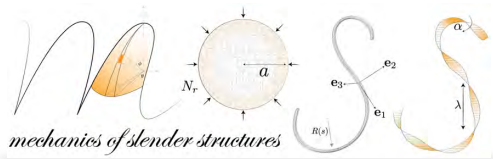
$$\psi(\bar{x}_3, \tau) = 1 + \sum_{n=0}^{\infty} \mathcal{A}(n) \sin[\lambda(n)\bar{x}_3] e^{-\lambda(n)^2 \tau}$$

Bending dynamics:

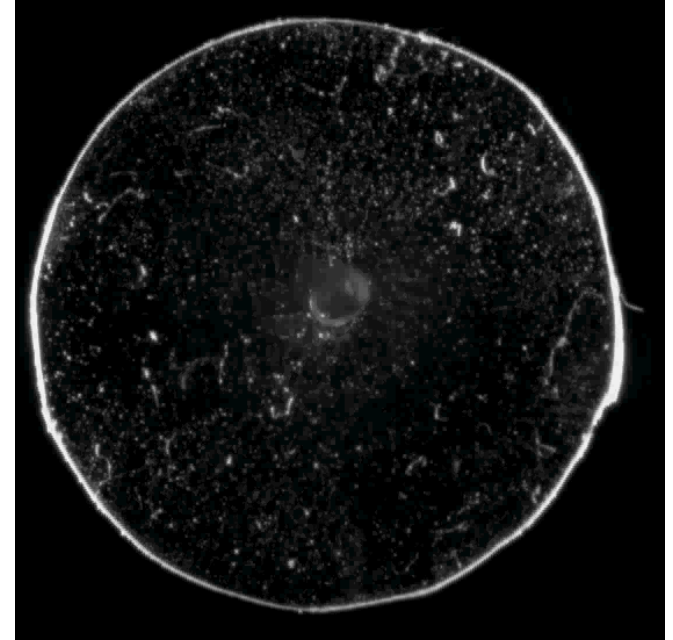
$$\bar{\kappa}_\varepsilon(\tau) = \frac{d}{S} \int_{-1/2}^{1/2} \psi(\bar{x}_3, \tau) \bar{x}_3 \, d\bar{x}_3$$

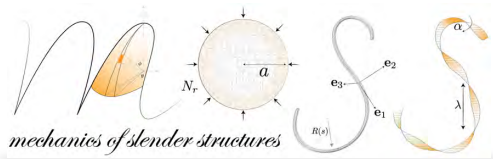
$$d/S = 12\varepsilon_{eq} L^2 / h^2$$



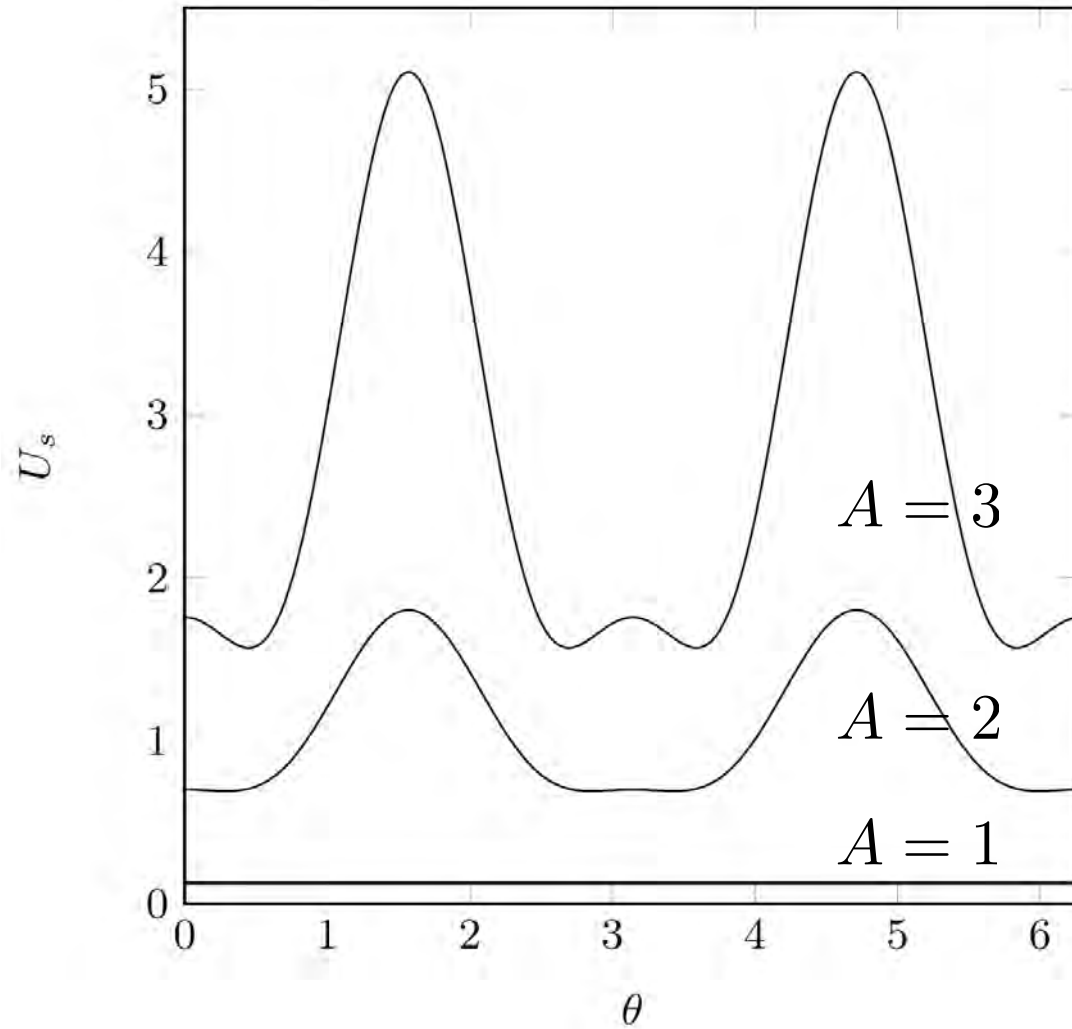


Dynamics: Twisting

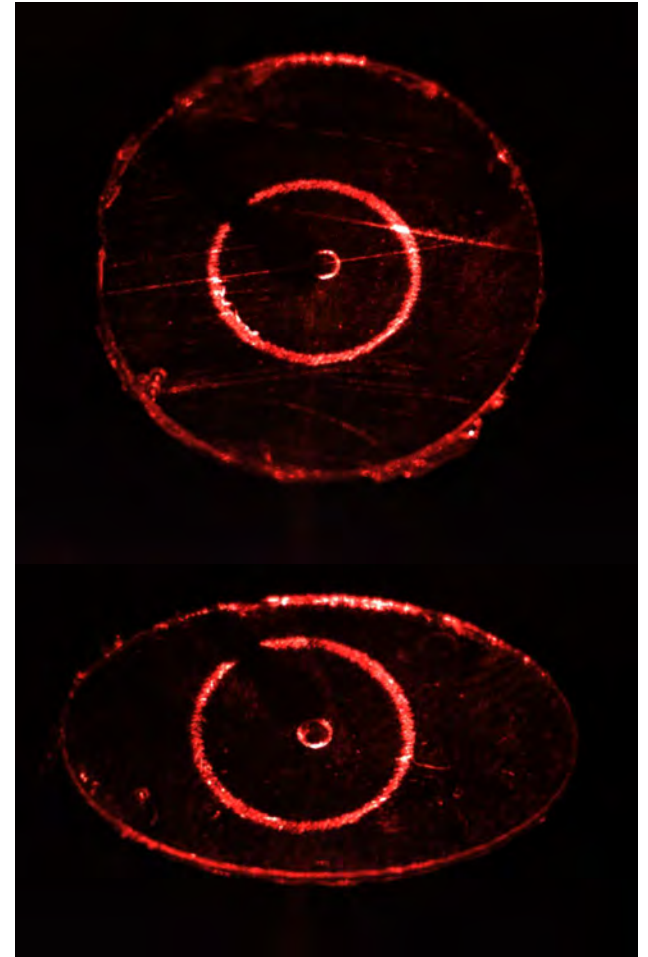




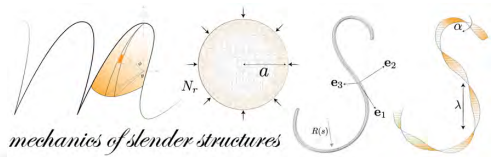
Dynamics: Twisting



$$a/b = 1$$



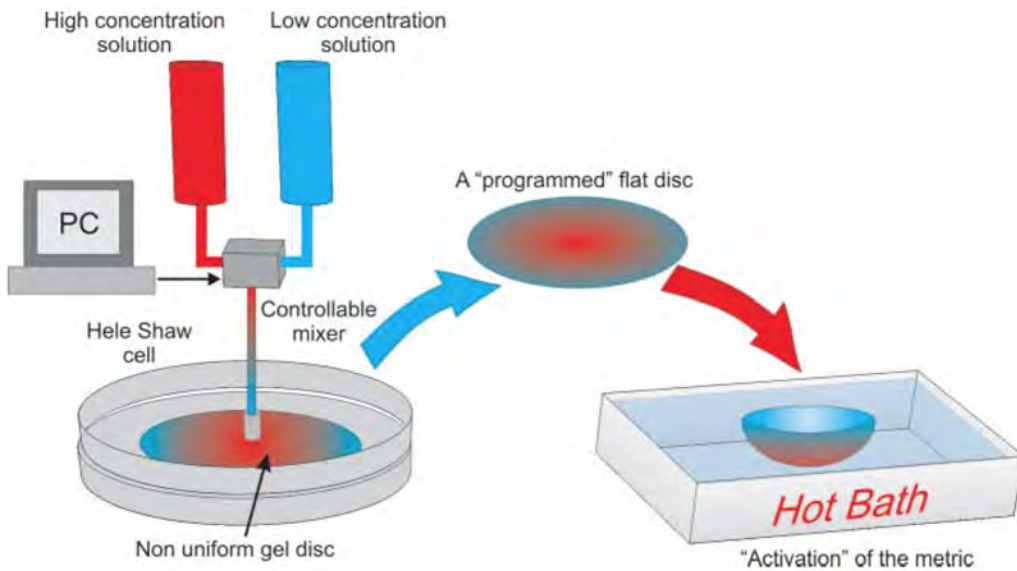
$$a/b = 1.4$$



How do thin structures grow?

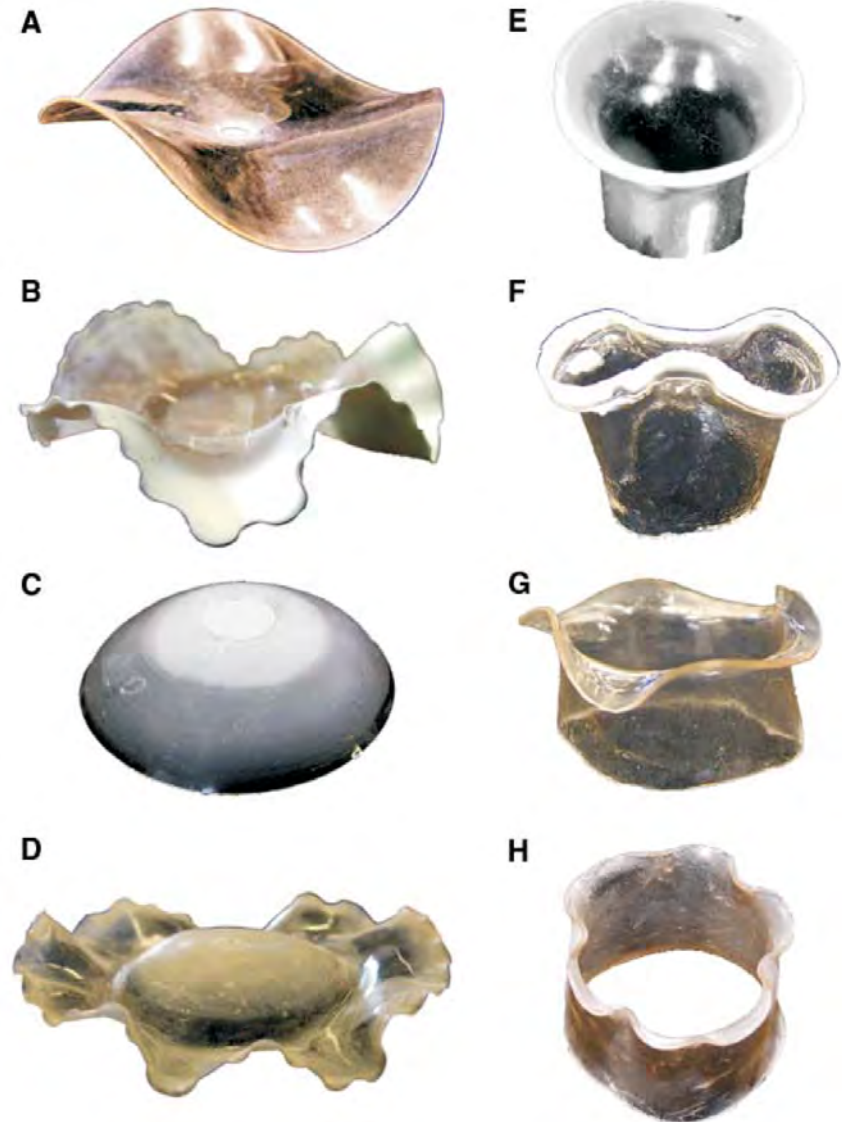
Permanent shape change and mass increase.

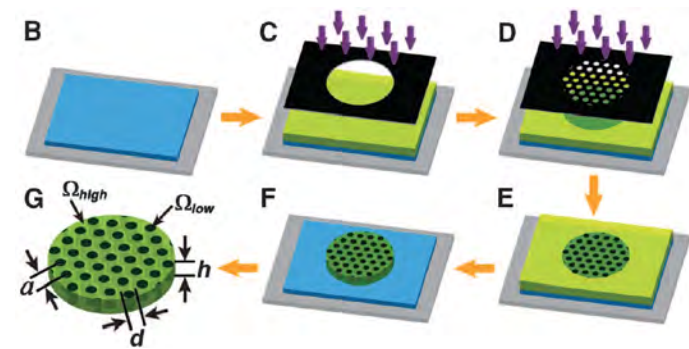
- Radial growth
- Through-thickness growth



Shaping elastic sheets by prescribing non-Euclidean metrics

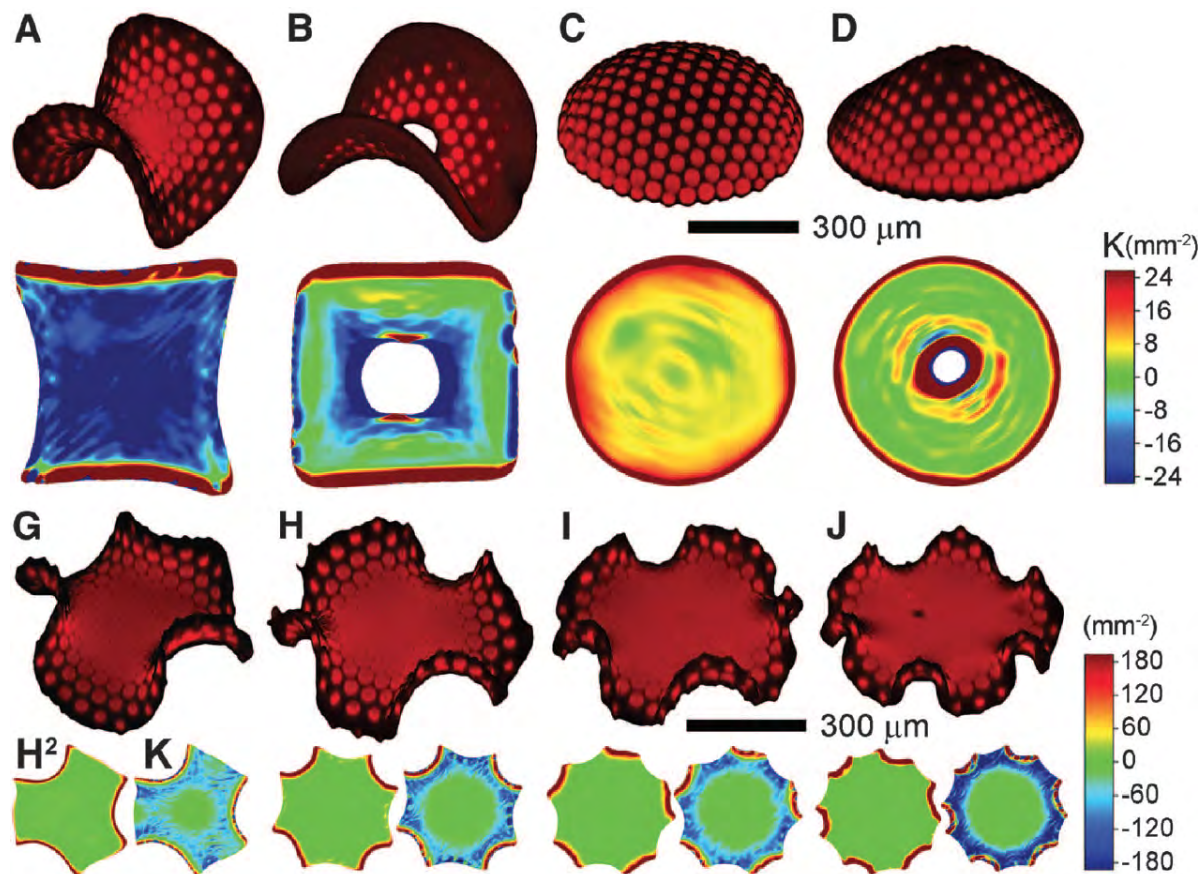
- Prepare gels that undergo nonuniform shrinkage.
- Buckling thin films based on chosen metrics.





Shaping elastic sheets by halftone gel lithography

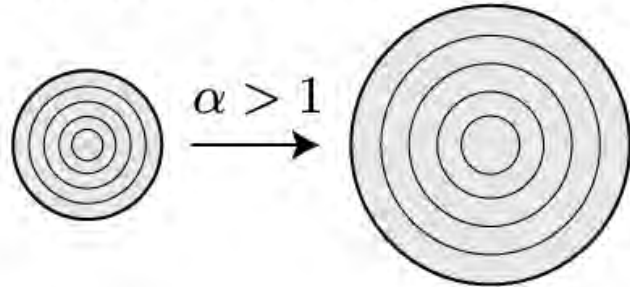
- Photopattern thin films.
- Thermal-actuated shape change.
- Swell to embedding based on prescribed metric.



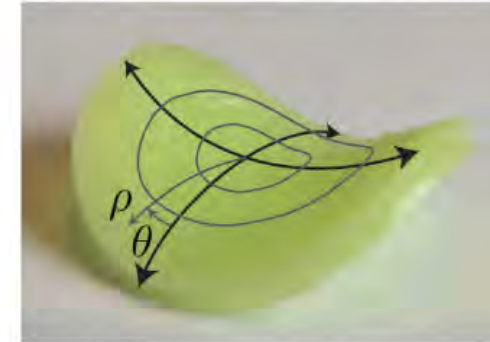
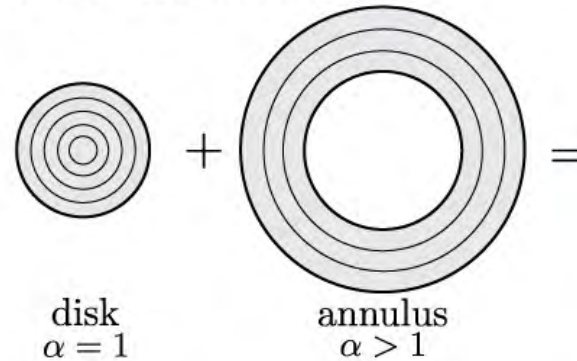
Geometric Composite

Independent Homothety

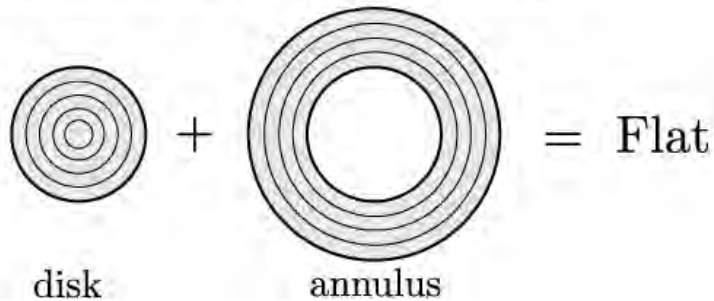
Homothetic Transformation



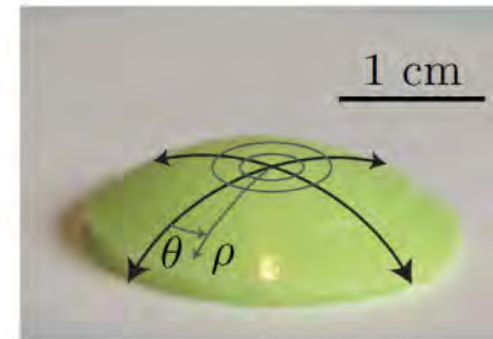
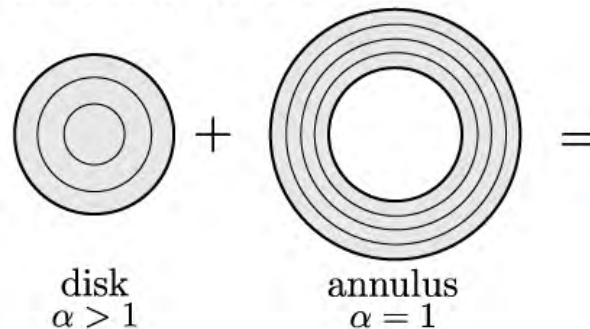
Stretched Annulus



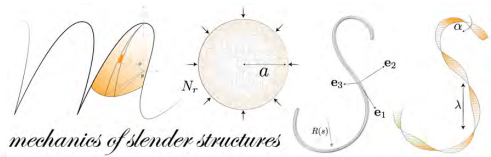
Reference Configuration



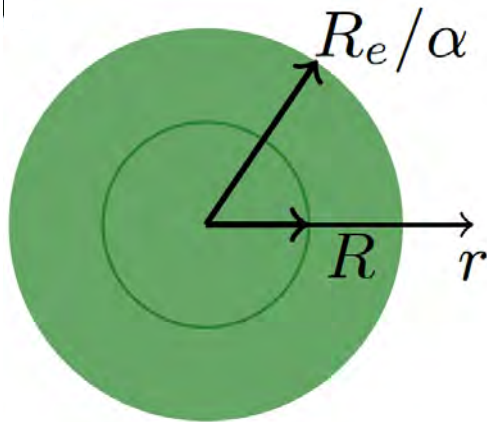
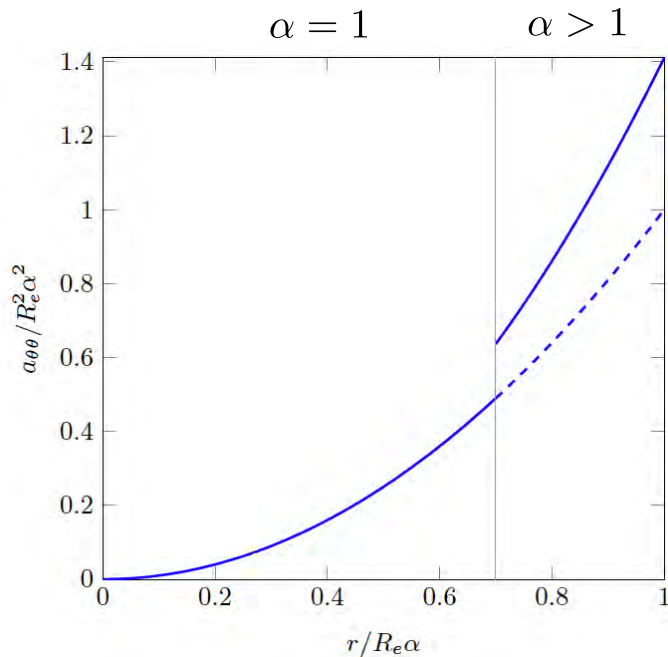
Stretched Disk



Goal: Use swelling to predictably & permanently morph plates into shells



Geometric Composite



Stretching Dominated

Will bend as much as possible while minimizing stretching.

Stretching Energy of the Plate (incompressible)

$$\mathcal{U}_s \simeq h \int_A E [\text{tr}^2(\underset{\substack{\uparrow \\ \text{Realized} \\ \text{metric}}}{a} - \underset{\substack{\uparrow \\ \text{Target} \\ \text{metric}}}{\bar{a}}) + \text{tr}(a - \bar{a})^2] \sqrt{|\bar{a}|} dA$$

Stretching Energy (Assume all strains zero, except: $a_{\theta\theta} - \bar{a}_{\theta\theta}$)

$$\mathcal{U}_s \simeq Eh \int_0^R \frac{(a_{\theta\theta} - r^2)^2}{r^3} dr + Eh \int_R^{R_e/\alpha} \frac{(a_{\theta\theta} - \alpha^2 r^2)^2}{\alpha^2 r^3} dr$$

First Fundamental Form

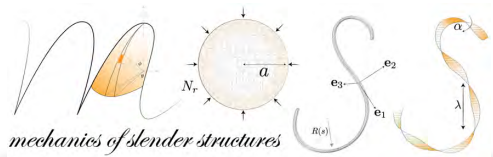
$$ds^2 = d\rho^2 + a_{\theta\theta}(\rho) d\theta^2$$

Gaussian curvature

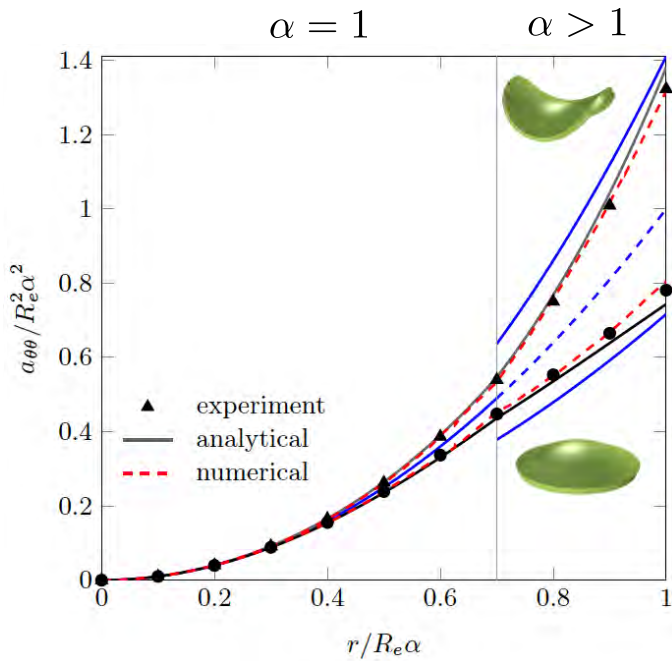
$$-\partial_{\rho\rho} \sqrt{a_{\theta\theta}} / \sqrt{a_{\theta\theta}}$$

Minimize Stretching Energy (Constant K metric)

$$a_{\theta\theta}(\rho) = (\sin(\sqrt{K}\rho) / \sqrt{K})^2$$



Geometric Composite



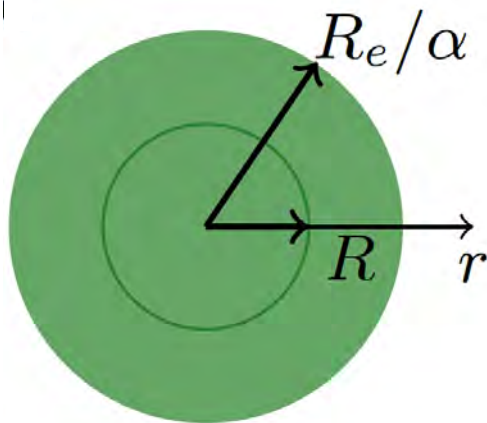
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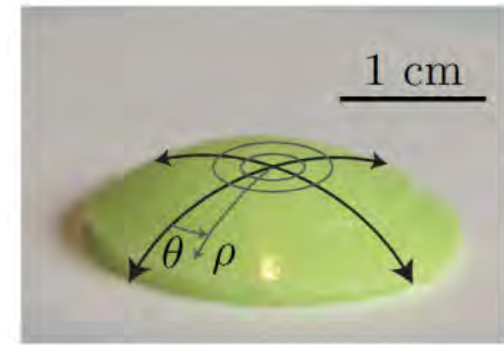
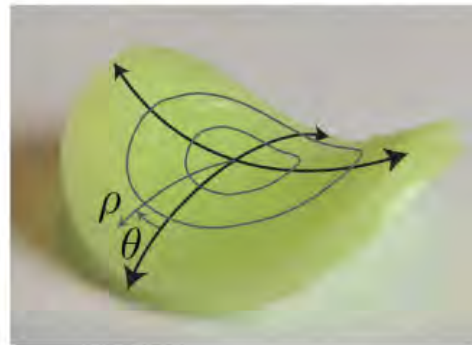
Taylor Expand $a_{\theta\theta}(\rho)$ (Assume: $|K| < \alpha^2 / R_e^2$)

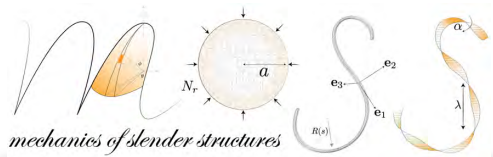
$$a_{\theta\theta}(\rho) = \rho^2 - \frac{K}{3}\rho^4 + \mathcal{O}(\rho^5)$$

\uparrow Flat metric \uparrow Kind of non-Euclidean metric



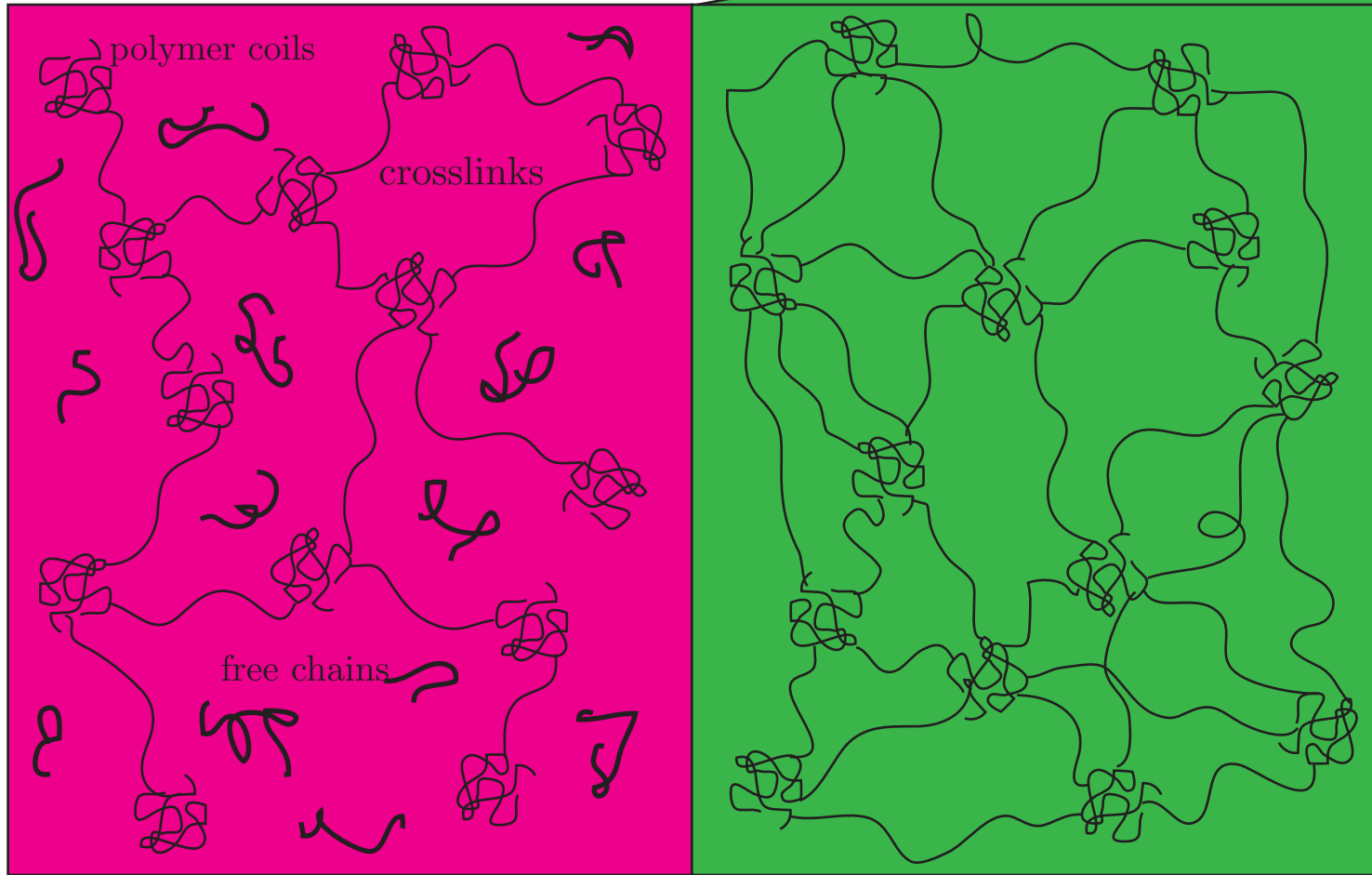
Experiments: Mechanical Strain





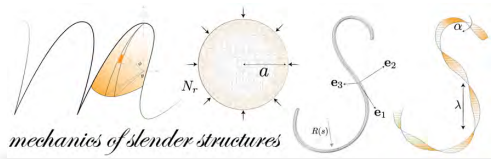
Residual Swelling

The elastomer contains free, uncrosslinked polymer chains.

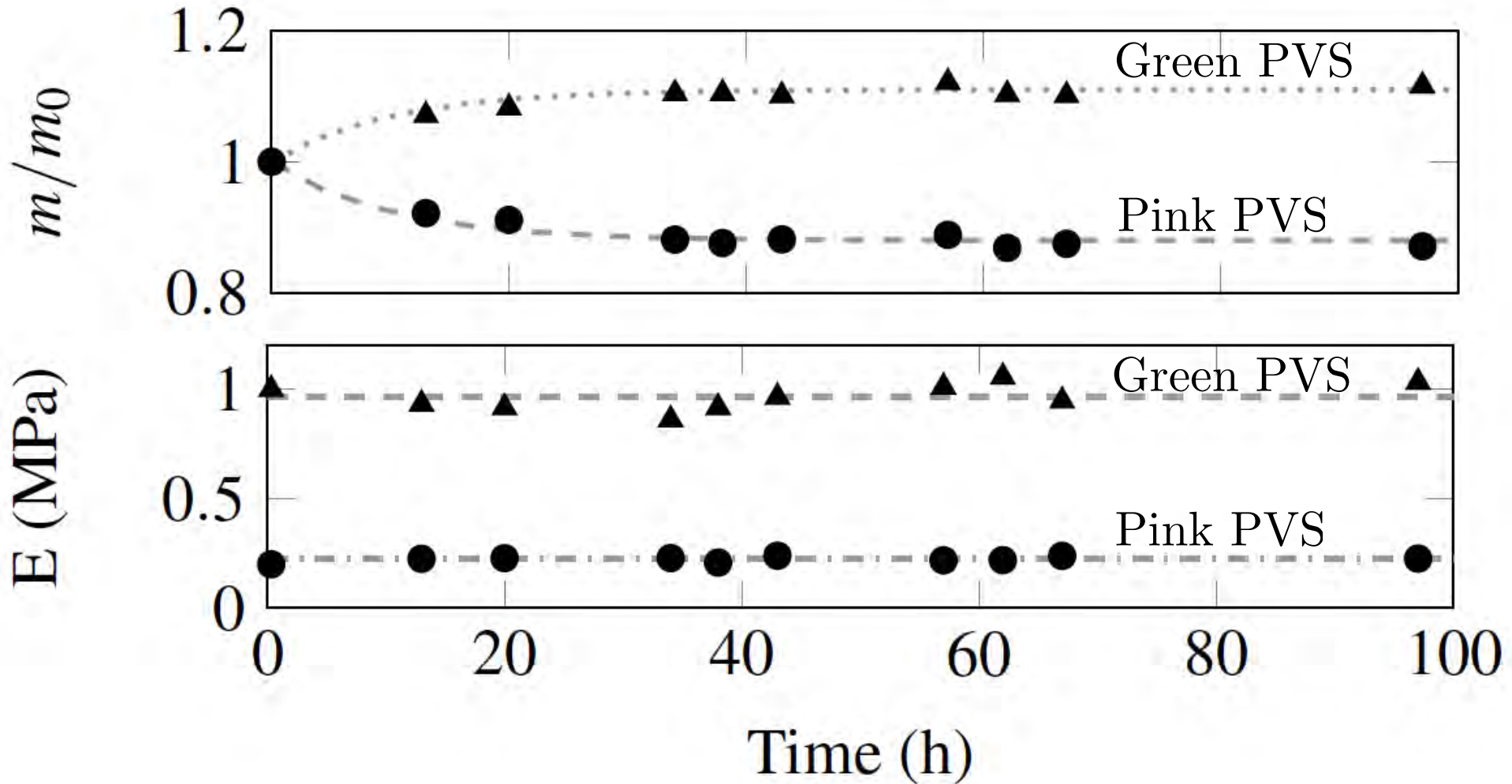


Elite Double 8 $E = 0.226 \text{ MPa}$
 $\nu = 0.5$

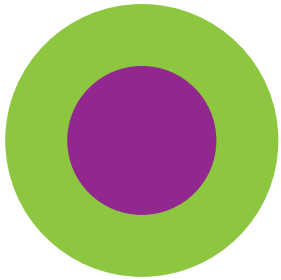
Elite Double 32 $E = 0.963 \text{ MPa}$
 $\nu = 0.5$



Residual Swelling

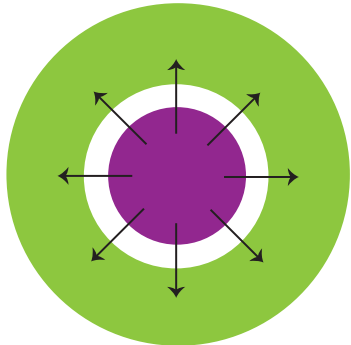


Residual Swelling

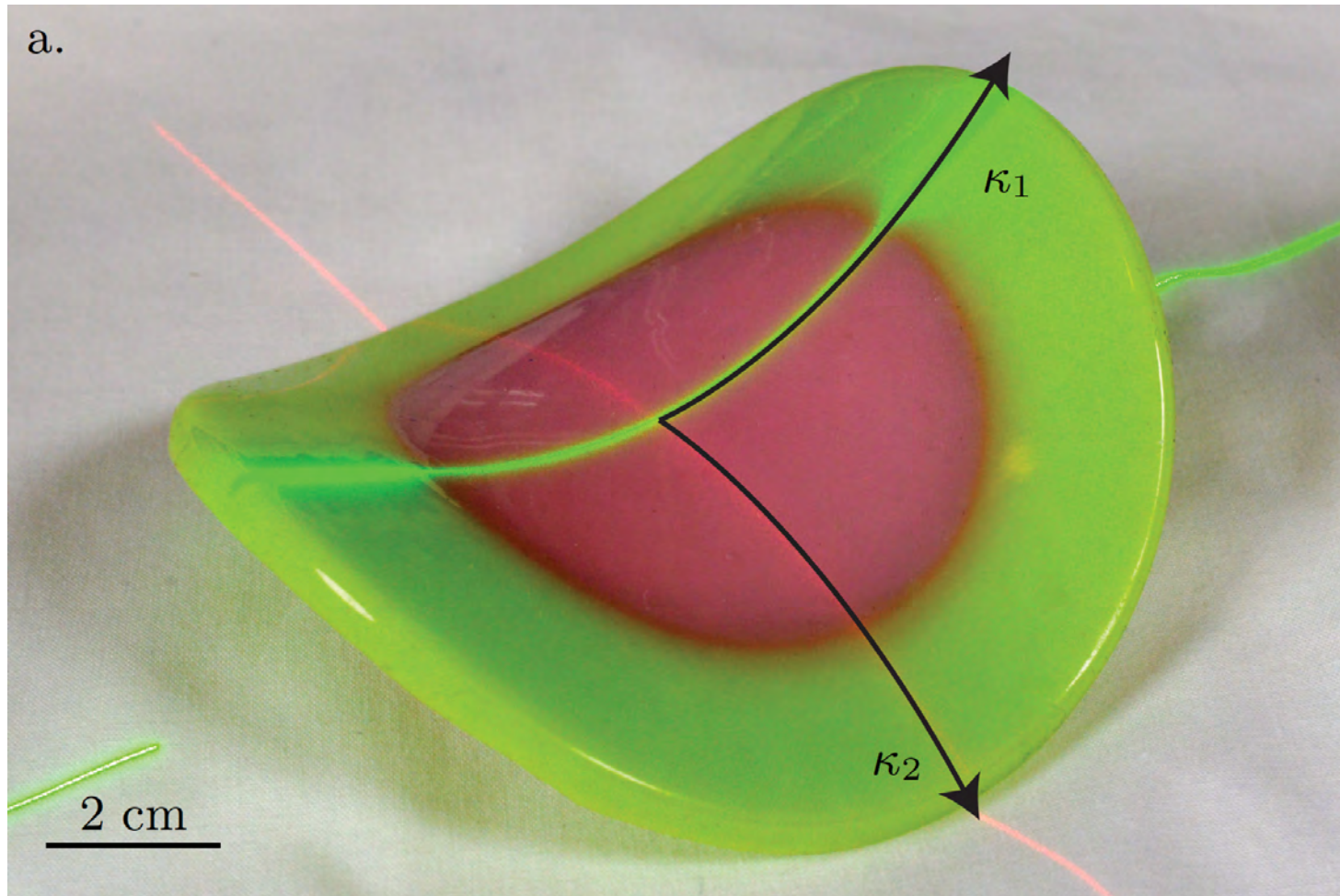


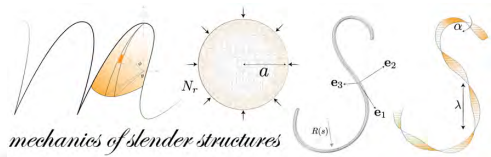
Reference Configuration

Swelling



Virtual Configuration





Residual Swelling

Stretching Energy

$$\mathcal{U}_s \simeq \int_0^R \frac{(a_{\theta\theta} - \alpha^{-2}r^2)^2}{\alpha^{-2}r^3} dr + \frac{E_a}{E_d} \int_R^{R_e} \frac{(a_{\theta\theta} - \alpha^2r^2)^2}{\alpha^2r^3} dr$$

\uparrow
 Modulus difference

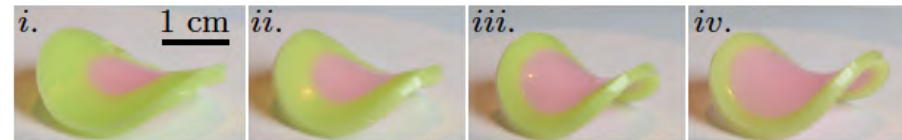
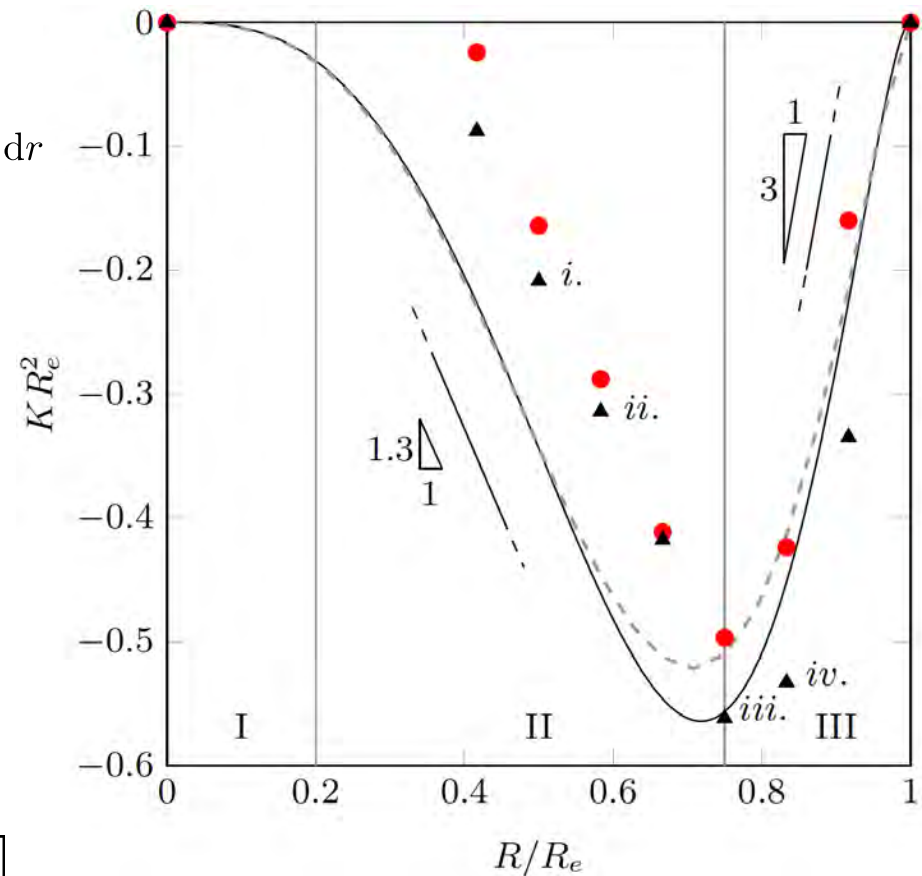
Stretching ratio

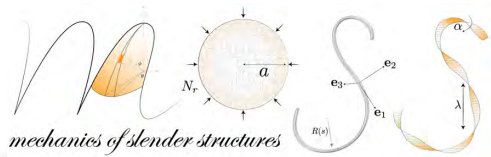
- Depends on chemical and material properties.
- Should vary with R/R_e
- Will depend on concentration gradient of free chains.

ansatz

- from conservation of mass & proportional to mass uptake in annulus.

$$\alpha = 1 + \eta(c_d - c_a) \left(\frac{R}{R_e} \right)^2 \left[1 - \left(\frac{R}{R_e} \right)^2 \right]$$





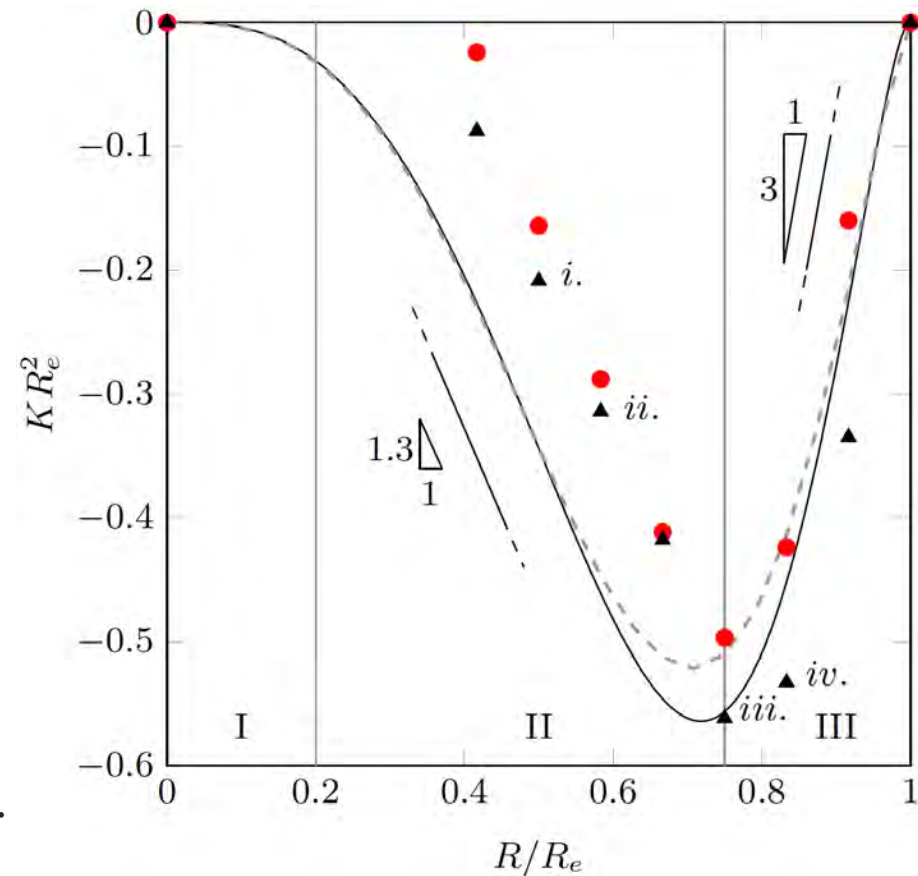
Residual Swelling

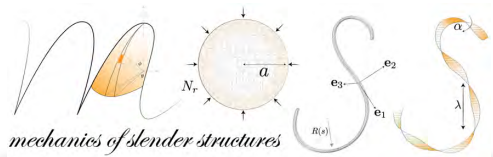
Approximate Analytical Solution

- Taylor expand about $(\alpha - 1)$
- $\bar{E} = E_a/E_d$ and $\bar{R} = R/R_e$

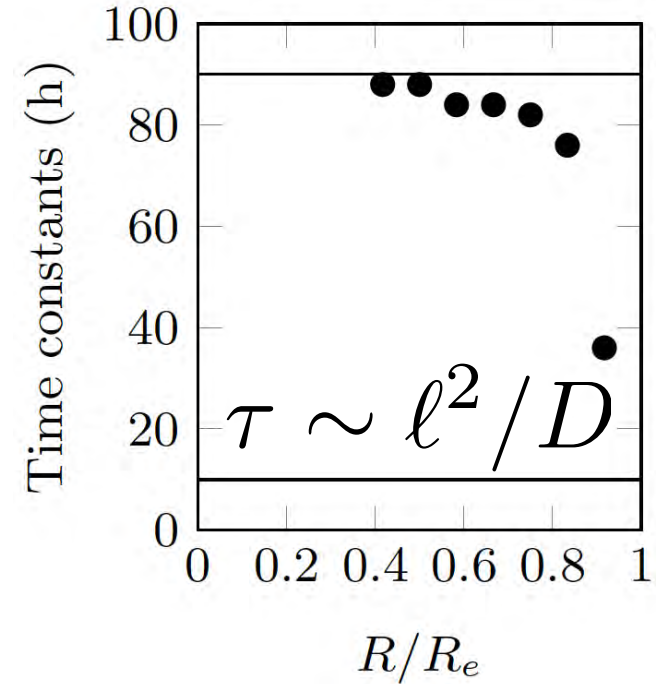
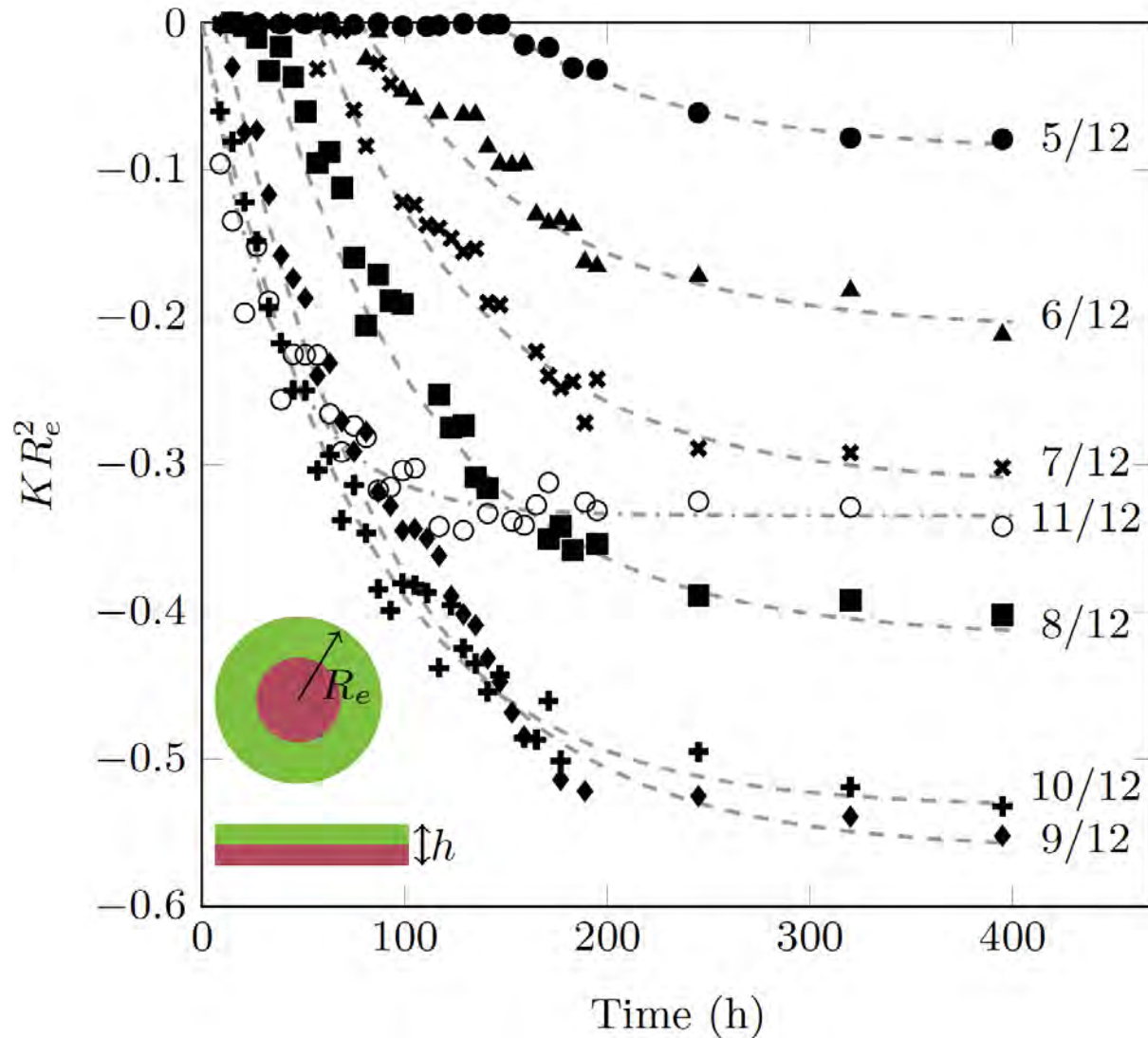
$$KR_e^2 \simeq 96(1 - \alpha_{\max})\bar{E}\bar{R}^3 \frac{(1 - \bar{R}^2)(1 - \bar{R}^3)}{\bar{R}^6(1 - \bar{E}) + \bar{E}}$$

2D analog to Timoshenko's model for thermal beam bending of **bimetallic strips**.





Swelling Dynamics

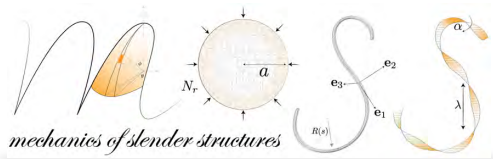


Diffusion of free chains

Characteristic length:

$$R/R_e < 1 \rightarrow \ell \sim R$$

$$R/R_e \approx 1 \rightarrow \ell \sim h$$



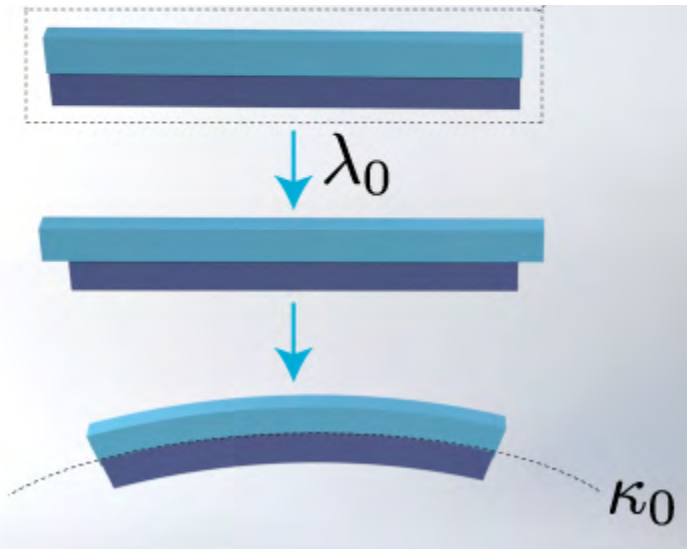
Growing Sheets

Consider a thin structure with a growing top layer.

- Caused by **swelling, growth, heating**, etc.

Elastic energy density depends on material properties & metric tensors

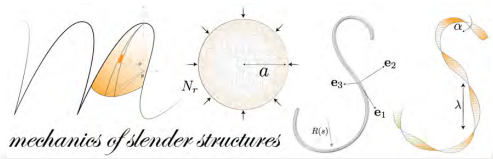
$$\bar{U} = \underbrace{\int [(1 - \nu)|\mathbf{a} - \bar{\mathbf{a}}|^2 + \nu \text{tr}^2(\mathbf{a} - \bar{\mathbf{a}})] \sqrt{|\bar{\mathbf{a}}|} \, dA}_{\text{Stretching Energy}} + \underbrace{\frac{h^2}{3} \int [(1 - \nu)|\mathbf{b} - \bar{\mathbf{b}}|^2 + \nu \text{tr}^2(\mathbf{b} - \bar{\mathbf{b}})] \sqrt{|\bar{\mathbf{a}}|} \, dA}_{\text{Bending Energy}}$$



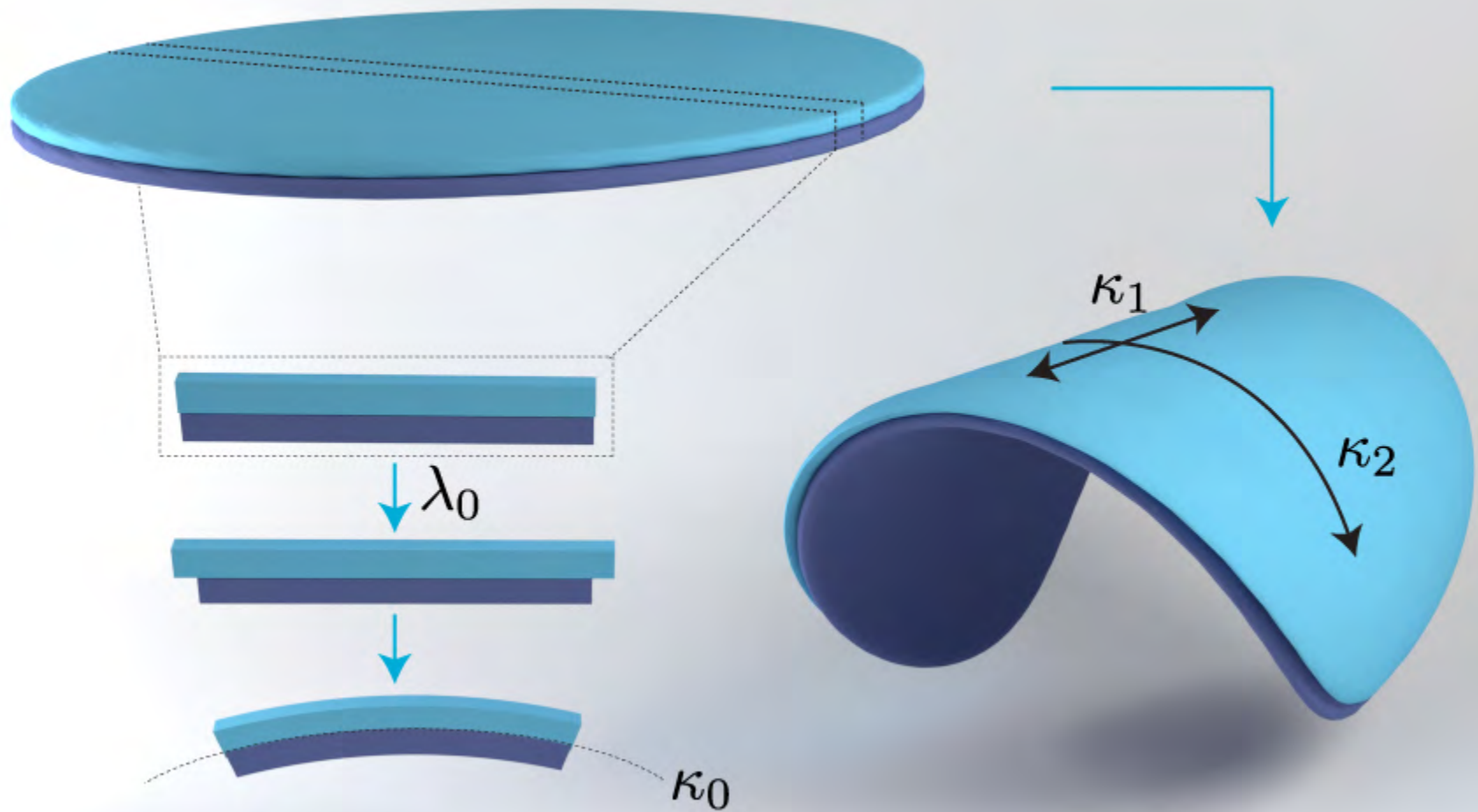
Lateral distances (Λ_0^2) and curvatures (κ_0) that make the sheet stress free.

Metric tensor Curvature tensor

$$\bar{\mathbf{a}} = \Lambda_0^2 \begin{pmatrix} 1 & 0 \\ 0 & 1 \end{pmatrix}, \quad \bar{\mathbf{b}} = \kappa_0 \begin{pmatrix} 1 & 0 \\ 0 & 1 \end{pmatrix}$$



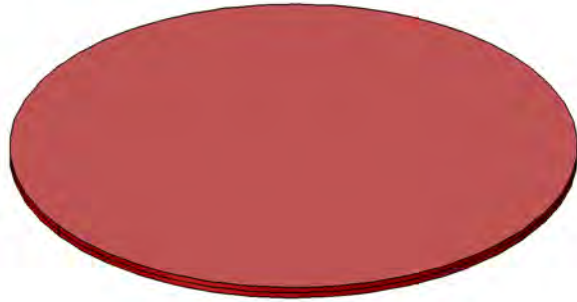
Isometric Limit



Numerical Simulations using COMSOL Multiphysics

- Finite, incompatible tridimensional elasticity with a Neo-Hookean incompressible material.
- Distortions used to simulate prestretch.
- Top layer subjected to distortion field:

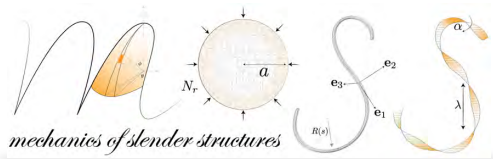
$$\mathbf{F}_o = \lambda(\mathbf{e}_1 \otimes \mathbf{e}_1 + \mathbf{e}_2 \otimes \mathbf{e}_2) + \mathbf{e}_3 \otimes \mathbf{e}_3$$



Residual Swelling Experiments

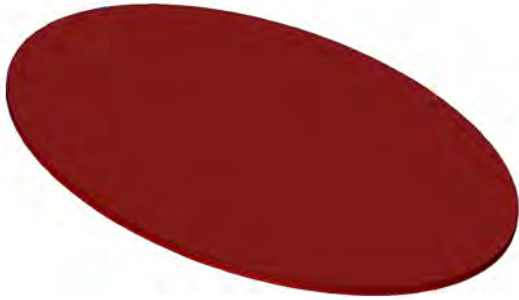
- PVS Bilayer (Total thickness ~ 600um)
- Axisymmetric, circular plate.
- Pink (top) source of swelling for green (bot).
- Real time = 75 minutes (Video: 640x RT)



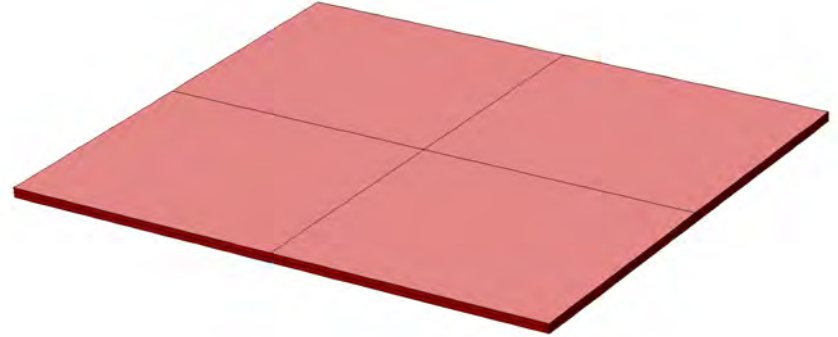


Isometric Limit

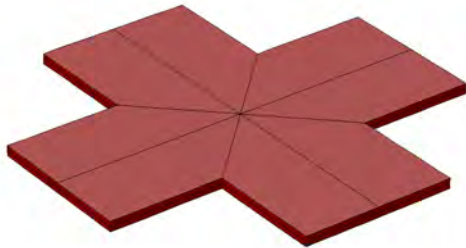
Ellipse



Rectangle

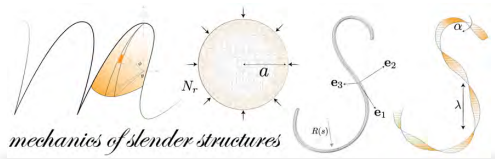


Cross

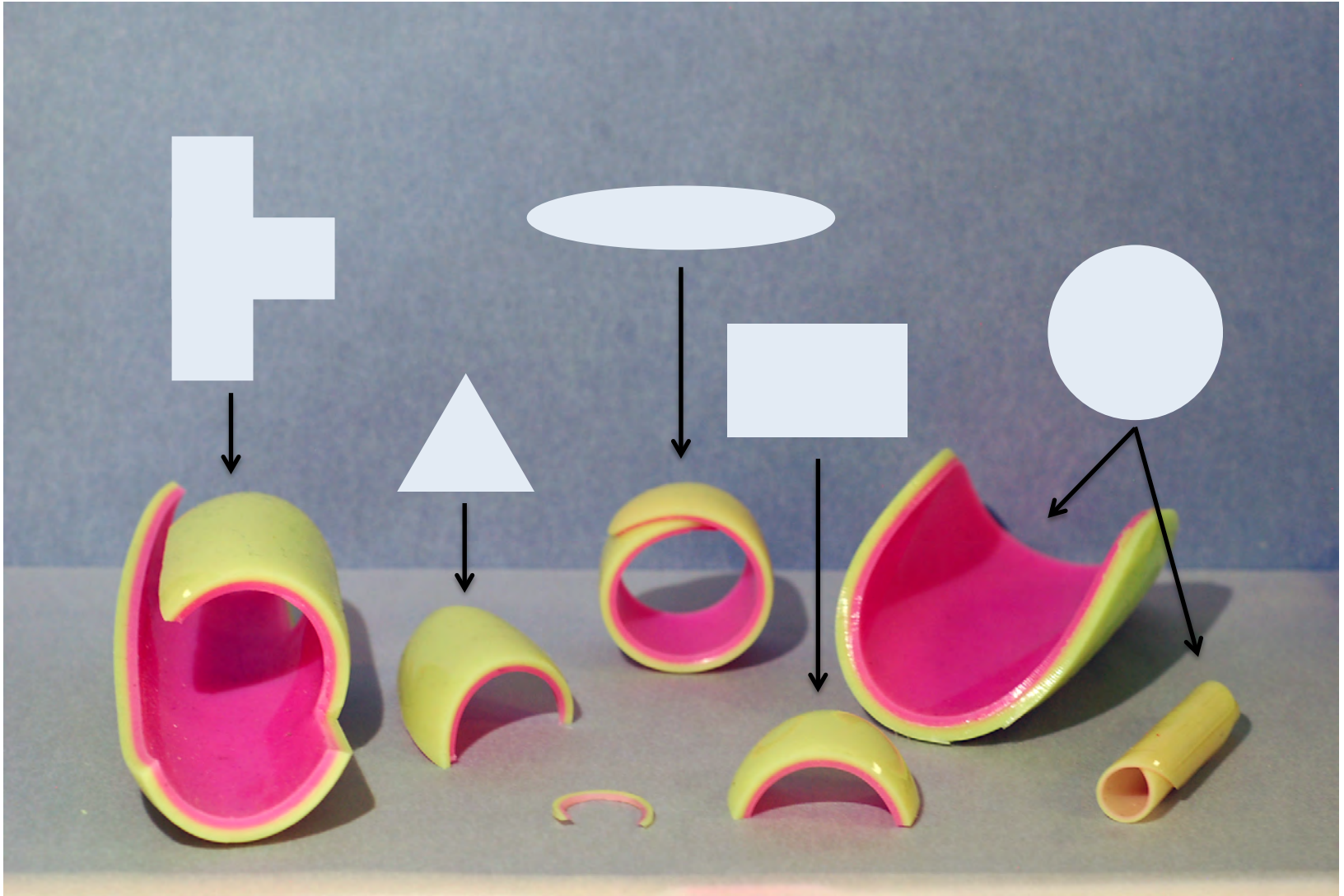


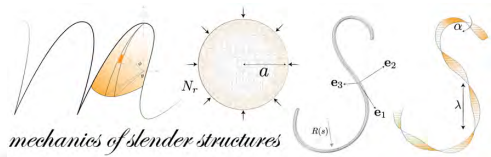
Leaf





Isometric Limit





Isometric Limit

In the **isometric limit**, the stretching energy is zero.

- i.e. $\mathbf{a} = \bar{\mathbf{a}}$

Curvature tensor (Cartesian)

- Second fundamental form:
 $Ldu^2 + 2Mdudv + Ndv^2$

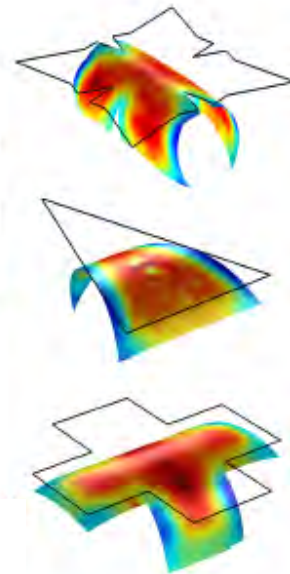
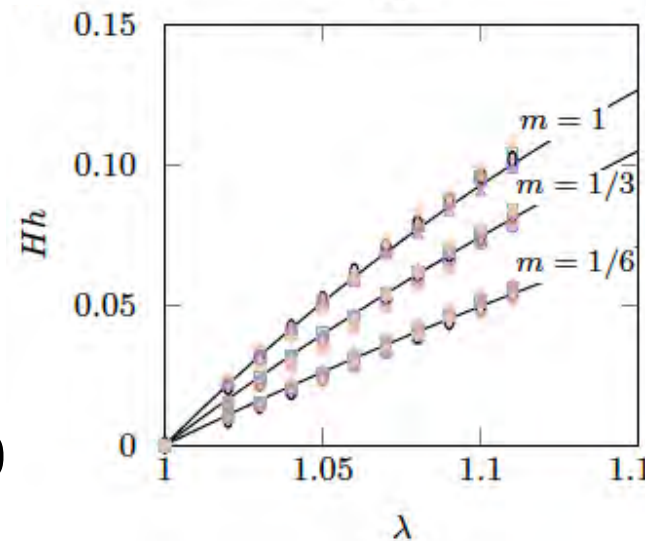
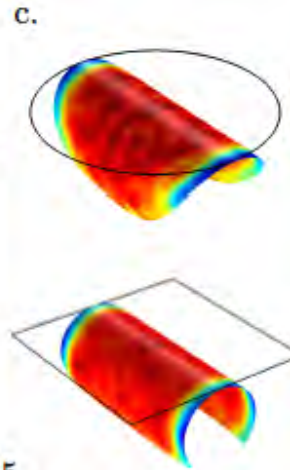
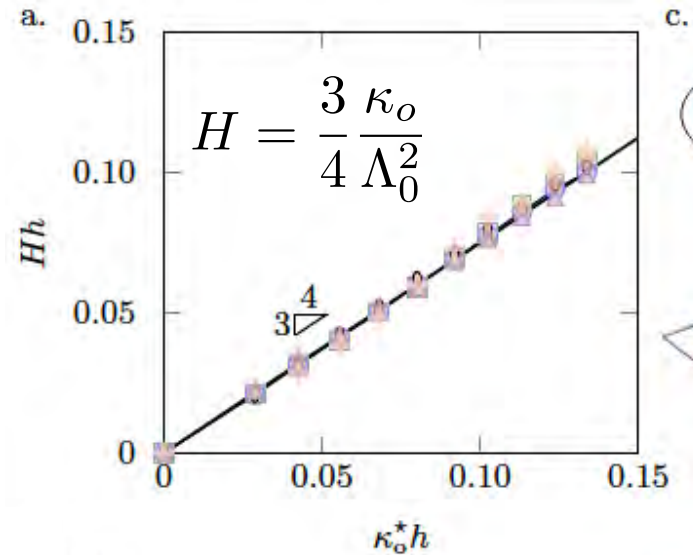
Minimize bending energy

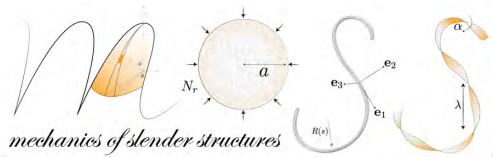
- Constrain the mid-surface to be flat
- Impose Lagrange multiplier enforcing

$$\underbrace{\Lambda_0^{-4} (LN - M^2)}_{\text{Gaussian Curvature}} = 0$$

Minimization yields:

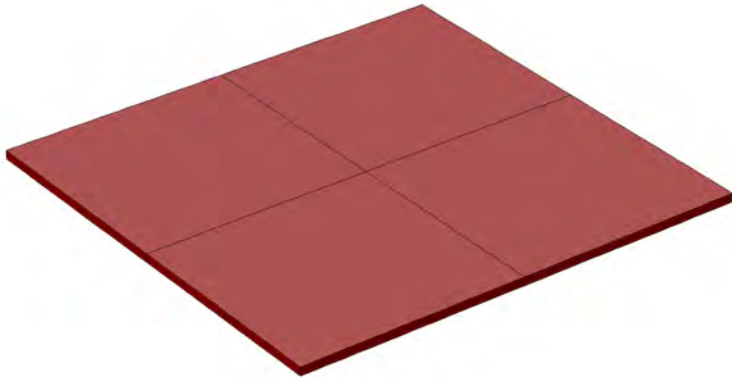
$$L + N = \kappa_0(1 + \nu) \quad \text{and} \quad K = 0$$





Bifurcation

Rectangular Sheet



Stretching Energy Density



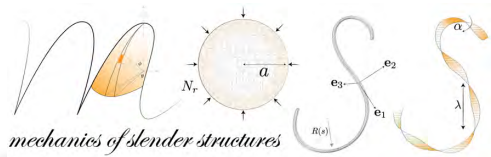
In the limit of **large stretching**, the sheet adopts an **isometry**.

For **small stretching**, the sheet is initially spherical curved.

- Bifurcation from spherical to cylindrical.

Classical problem (limited to circular and elliptical disks)

- Stoney formula relating stress to curvature.
- Strain mismatch work (Hyer, Freund, Seffen, etc.)



Bifurcation

Stretching Energy

Assuming a metric with constant K
(Gauss normal coords)

$$\bar{U}_s = \frac{1}{9} \lambda_0^6 K^2 \int_A r^4 dA$$

Bending Energy

In the spherical shape: $L = N$

$$\bar{U}_b = h^2 \lambda_0^{-2} A (L - \kappa_0)^2$$



Shape factor:

- Assume metric is axisymmetric.
- Assume K is homogenous.

$$a_{rr} = \bar{a}_{rr} \quad a_{r\theta} = 0$$

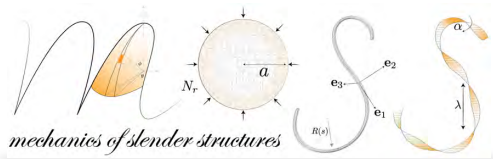
$$a_{\theta\theta}(\rho) = \rho^2 - \frac{K}{3} \rho^4 + \mathcal{O}(\rho^5)$$

Shape factor:

$$\mathcal{S} \equiv \left(\frac{2}{9} \frac{1}{A} \int_A r^4 dA \right)^{1/4}$$

Structural Slenderness:

$$\gamma = h / \mathcal{S}$$



Bifurcation

Energetic cost to continue bending into a **spherical cap**:

$$\bar{U}_{bb} = (1/9)L^4 \Lambda_o^{-2} \int r^4 dA + h^2 A \Lambda_o^{-2} (L - \kappa_o)^2$$

Energy to bend as a **cylinder**:

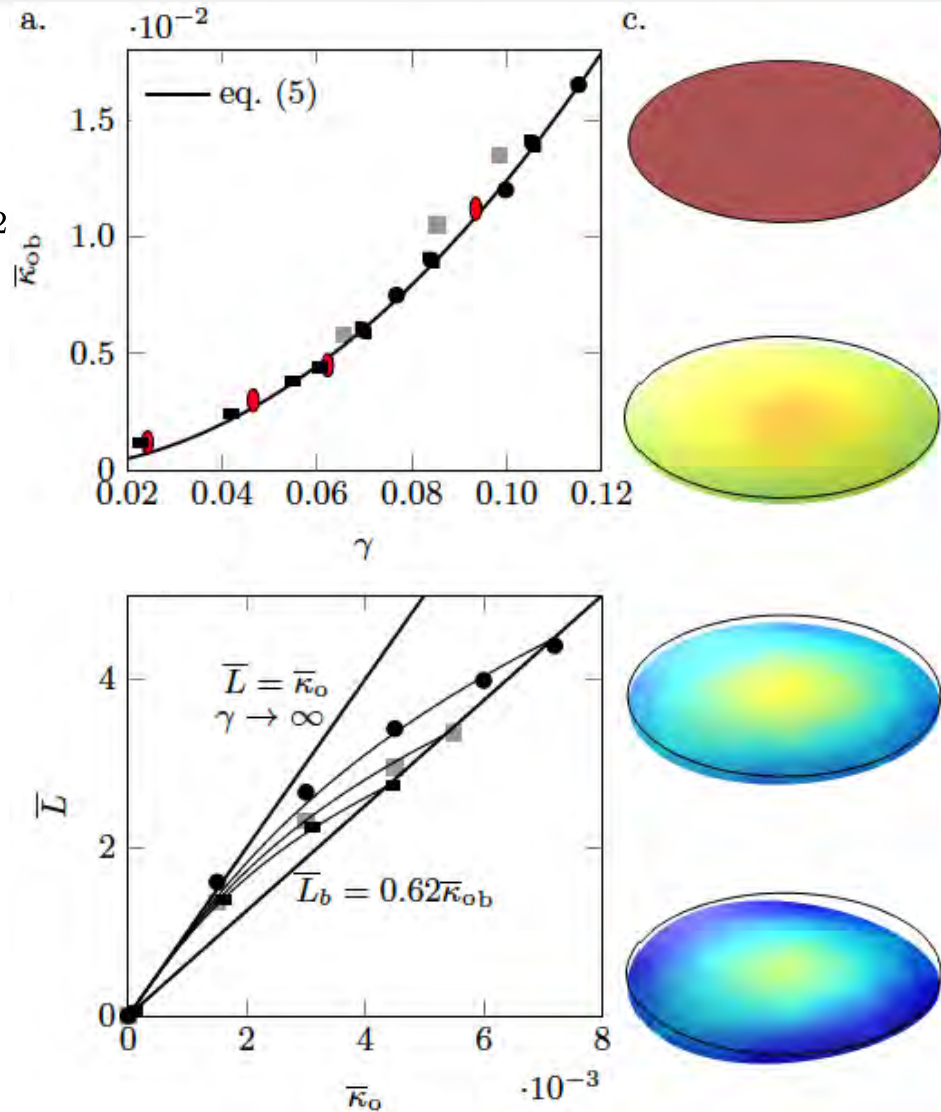
$$\bar{U}_{ab} = (1/4)h^2 A \Lambda_o^{-2} \kappa_o^2$$

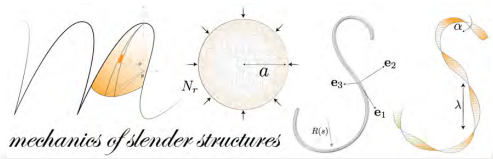
Energy balance:

$$\frac{1}{2} \bar{L}_b^4 + \gamma^4 \left(\bar{L}_b^2 - 2\bar{L}_b \bar{\kappa}_{ob} + \frac{3}{4} \bar{\kappa}_{ob}^2 \right) = 0$$

Bifurcation curvature:

$$\bar{\kappa}_{ob} = \frac{2\sqrt{2}}{3^{3/4}} \gamma^2$$



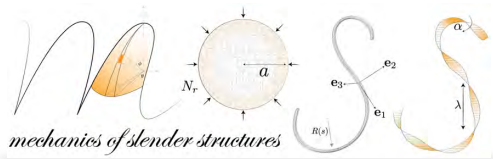


mechanics of slender structures

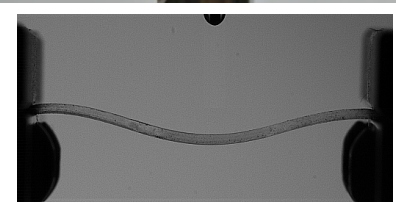
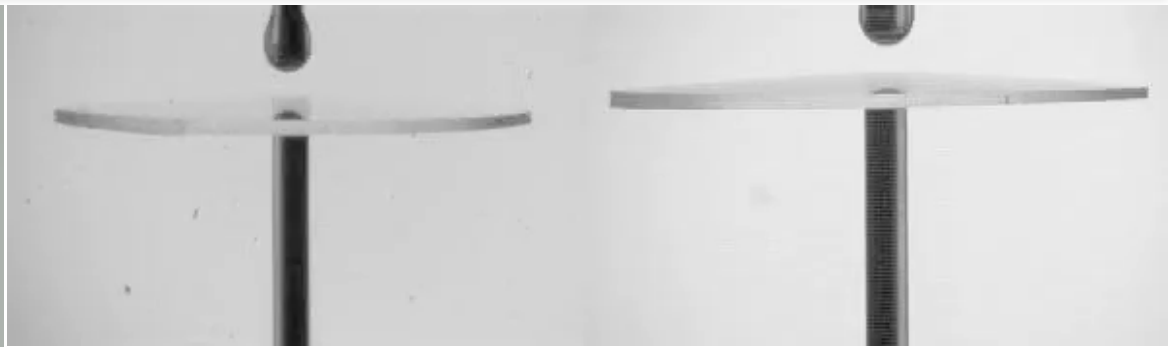
Shell Growth



...with M. Trejo, J. Bico, and B. Roman.



Swelling Structures



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Publications

1. A. Pandey and D.P. Holmes. "Swelling Induced Deformations: A Materials-Defined Transition from Structural Instability to Surface Instability." *Soft Matter*, **9**, 7049, (2013).
2. D.P. Holmes, M. Roché, T. Sinha, and H.A. Stone. "Bending and Twisting of Soft Materials by Non-homogenous Swelling" *Soft Matter*, **7**, 5188, (2011).
3. M. Pezulla, S. Shillig, P. Nardinocci, and D.P. Holmes. "Morphing of Geometric Composites via Residual Swelling," *Soft Matter*, **11**, 5812-5820, (2015).